

# Modeling Challenges for Ocean Acidification

Scott Doney (WHOI)

Ivan Lima (WHOI), Nathalie Goodkin (BIOS)

## Processes, Magnitudes & Uncertainties

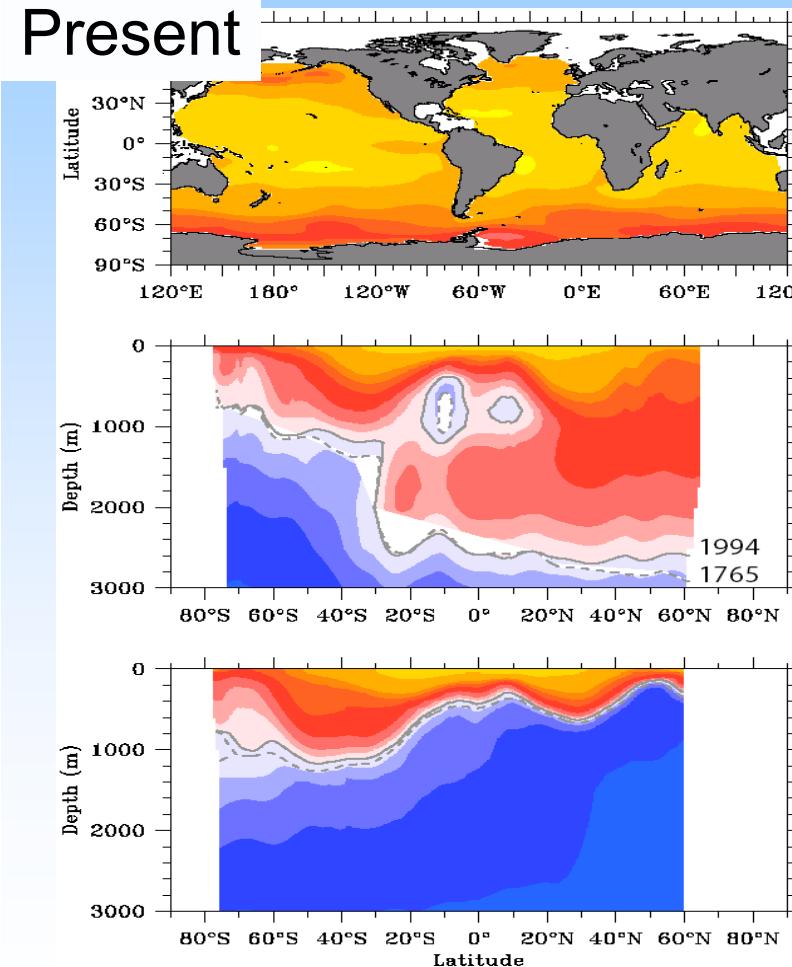
*(with a bottom-up, pelagic bias)*

- anthropogenic  $\text{CO}_2$ , pH &  $\text{CaCO}_3$  saturation  $\Omega$
- other acidity/alkalinity inputs
- biogeochemical and ecological impacts
- policy, economic & social dimensions



# CO<sub>2</sub> Effects on CaCO<sub>3</sub> Saturation Ω (aragonite)

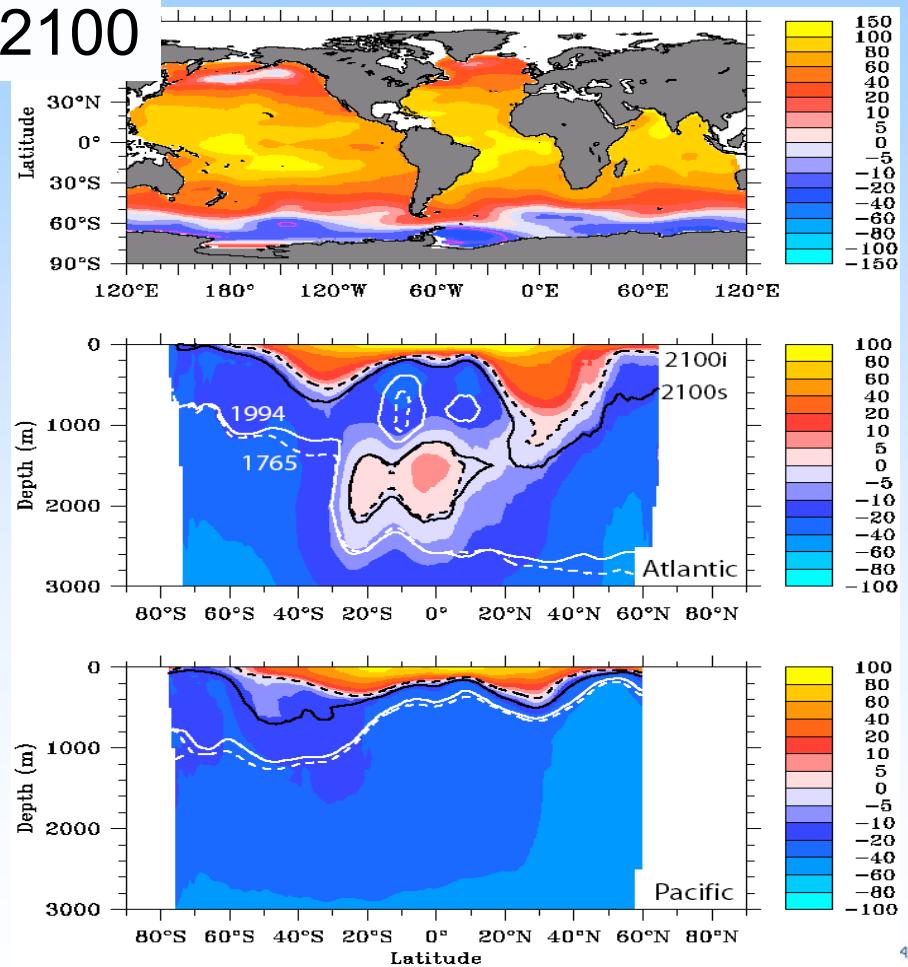
Present



$$\Omega = [\text{CO}_3^{2-}][\text{CO}_3^{2-}] / K_{\text{sp}}$$

$$\Delta[\text{CO}_3^{2-}] = [\text{CO}_3^{2-}]_{\text{obs}} - [\text{CO}_3^{2-}]_{\text{sat}}$$

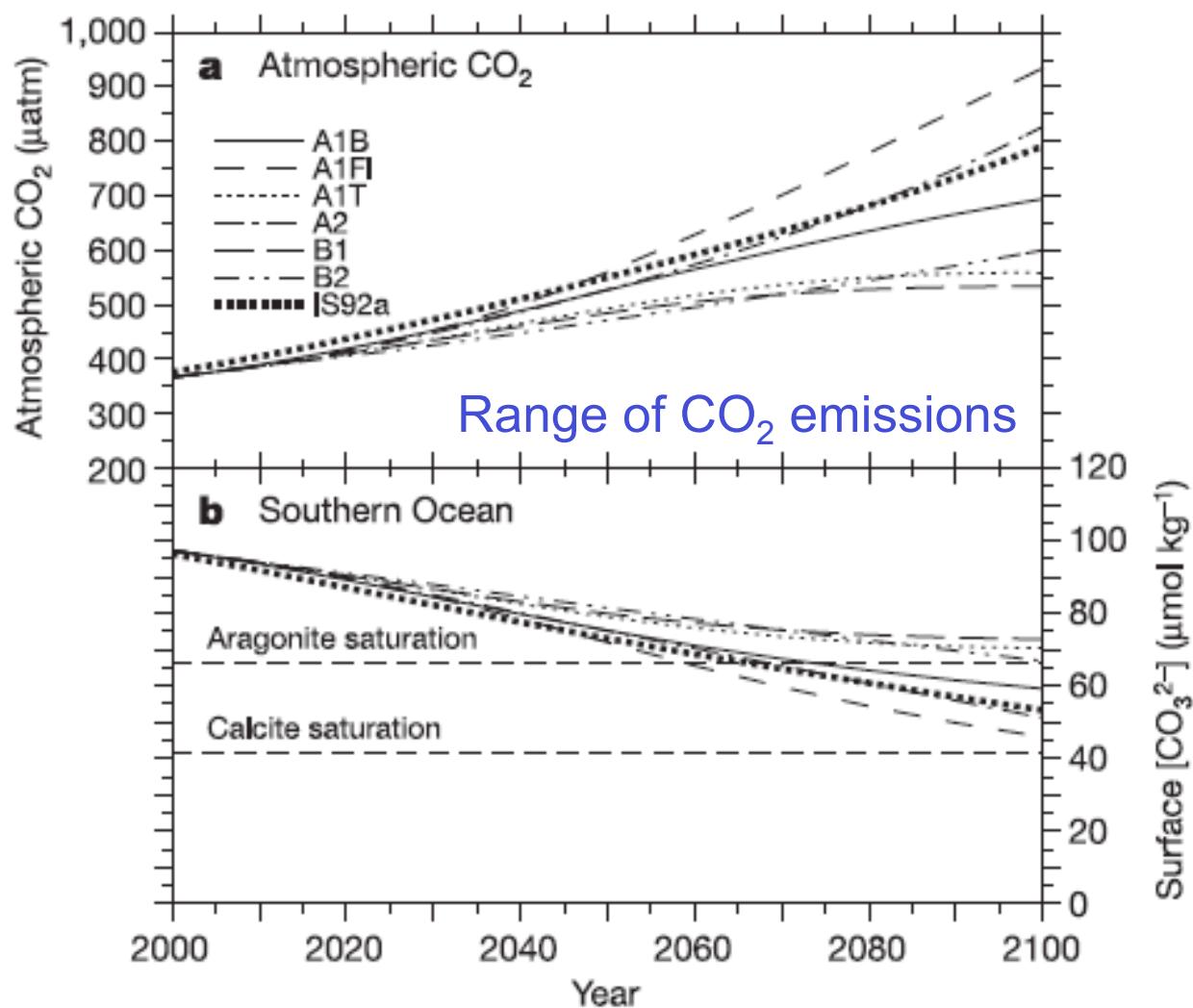
2100



Orr et al. Nature (2005)

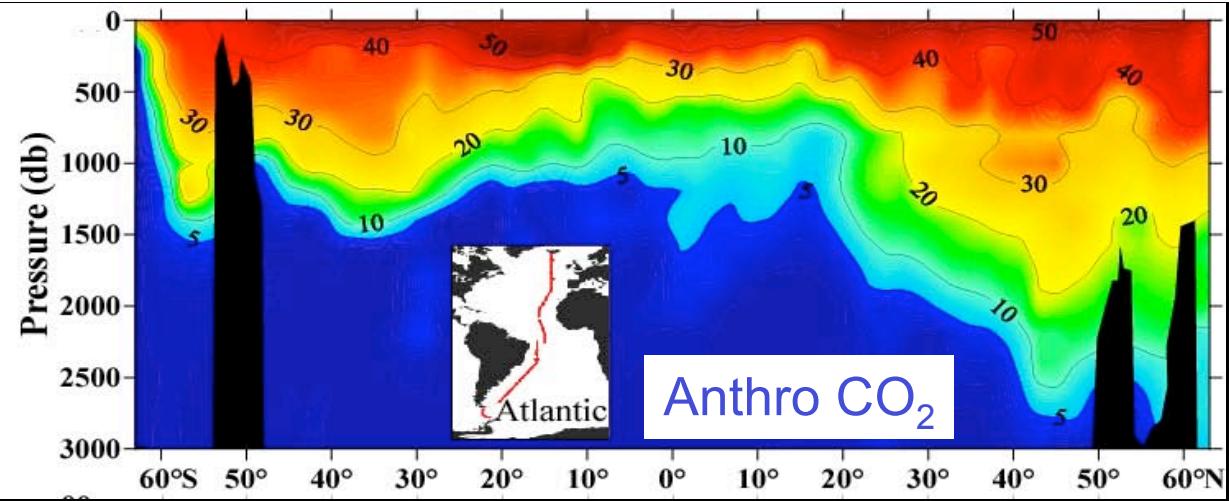
# Anthropogenic $CO_2$ & Ocean pH

- $CO_2$  emissions  
(social, political,  
economic)
- atmospheric  $CO_2$ 
  - land & ocean  
uptake;
  - climate-carbon  
feedbacks
- ocean circulation



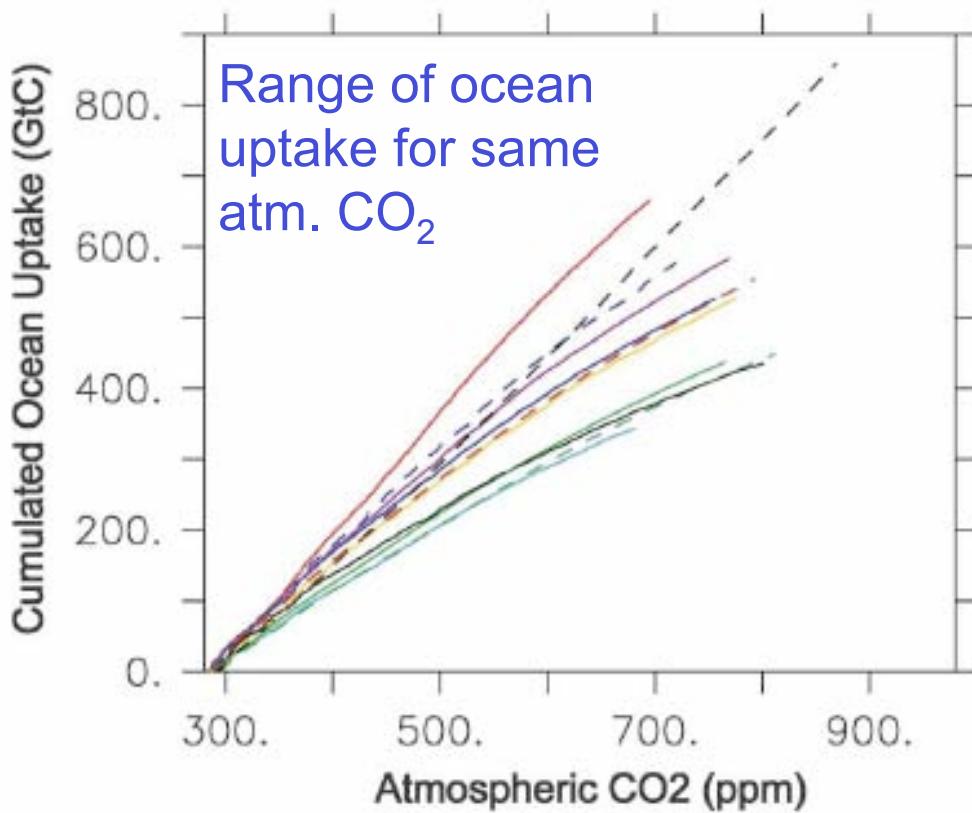
Orr et al., Nature (2005)



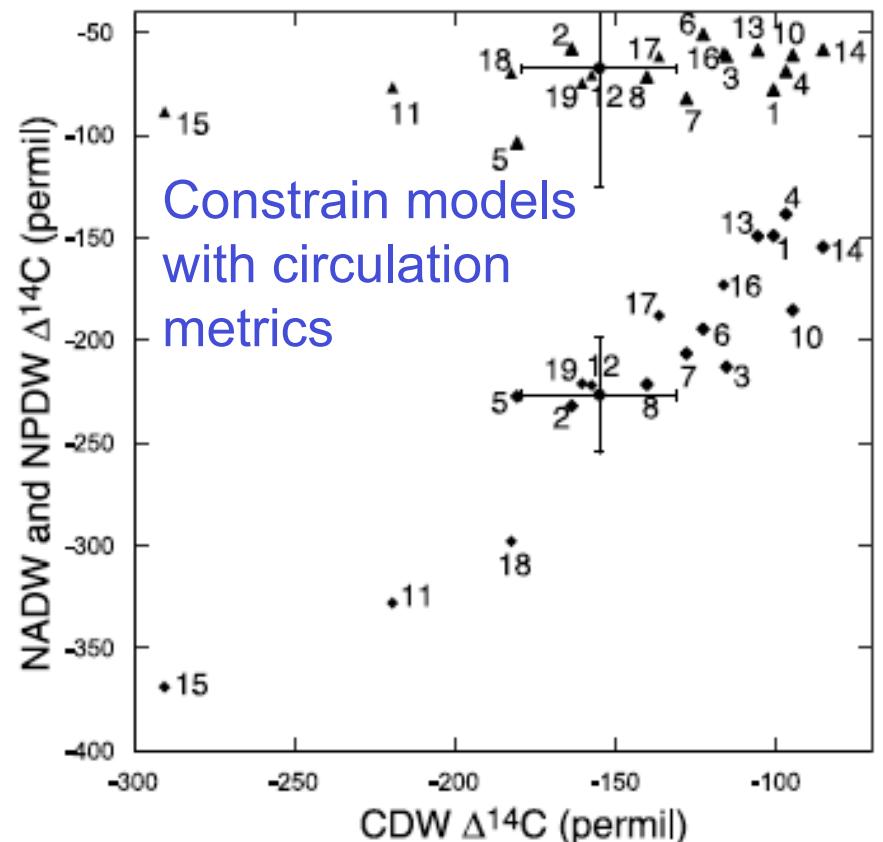


## CO<sub>2</sub> Uptake & Ocean Circulation

Sabine et al.,  
Science (2004)



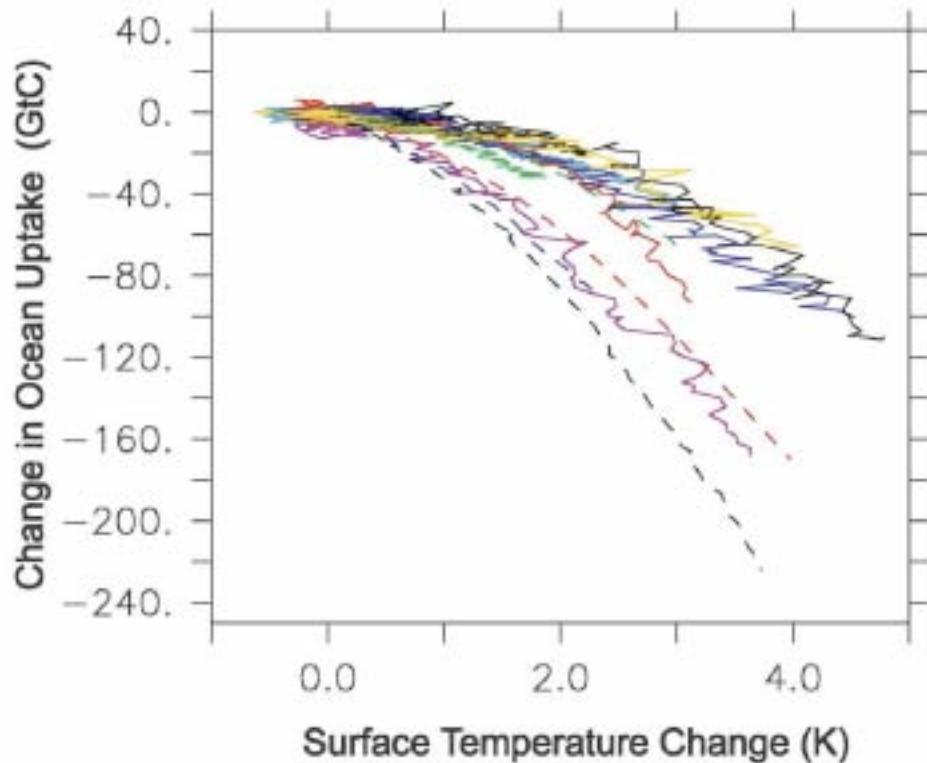
Friedlingstein et al., J. Climate (2006)



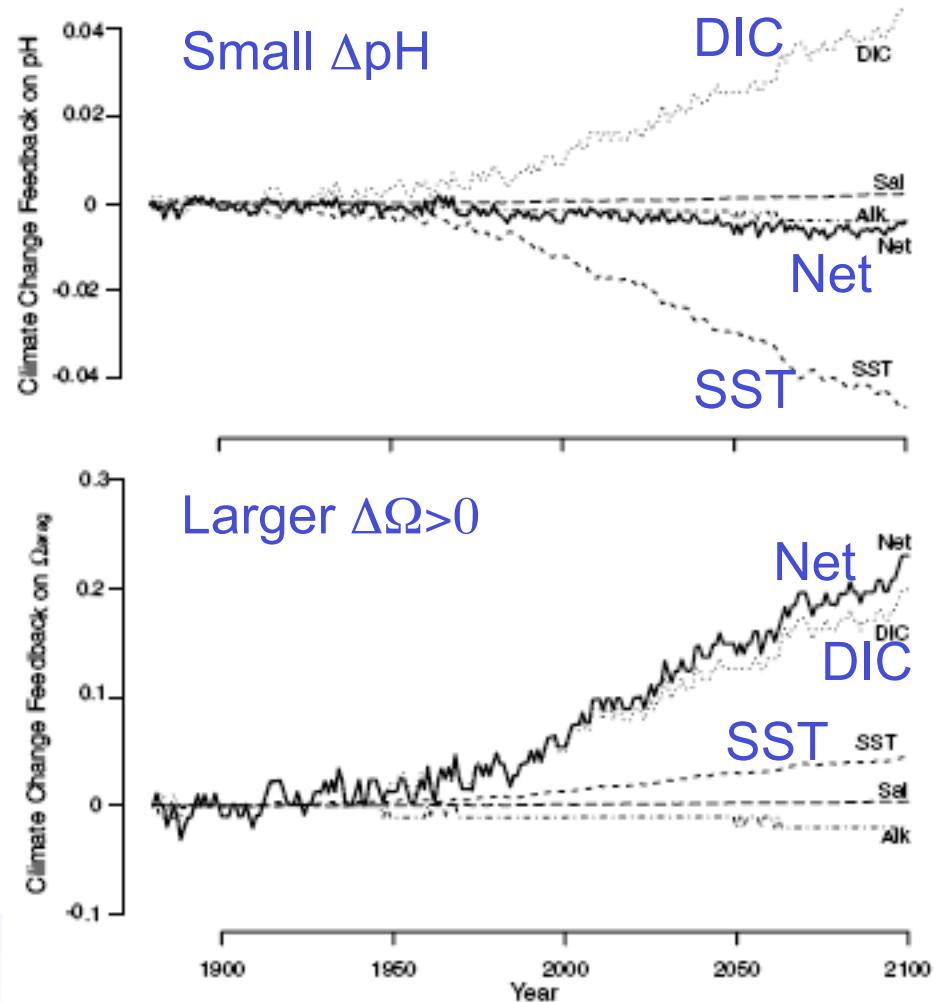
Matsumoto et al., GRL (2004)

# Effect of Climate Change Feedbacks

Reduced CO<sub>2</sub> uptake due to warming  
& altered circulation



Friedlingstein et al. J. Climate  
(2006); Fung et al. PNAS (2005)

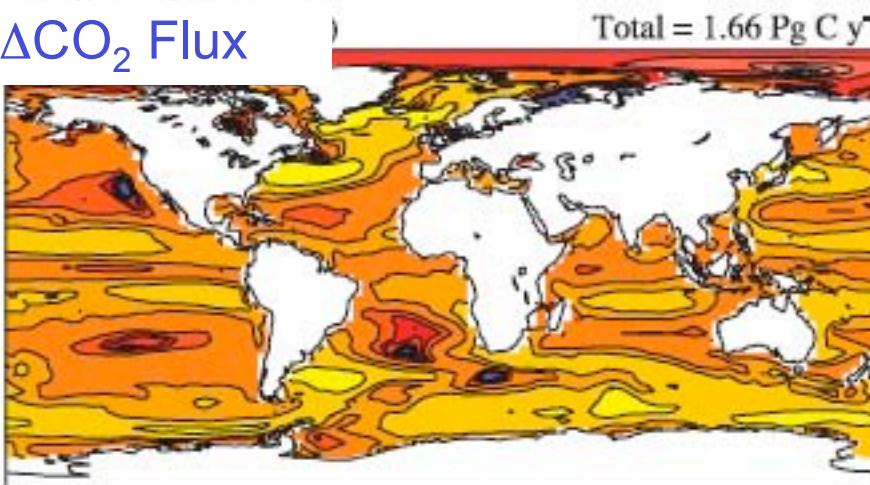


McNeil and Matear, Tellus (2007)  
Cao et al., GRL (2007)

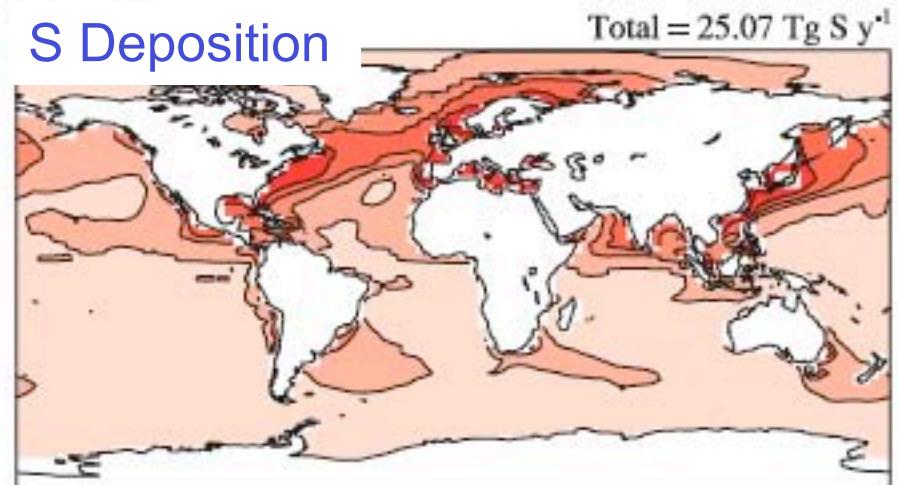


# Anthropogenic Fluxes of $\text{CO}_2$ , N and S

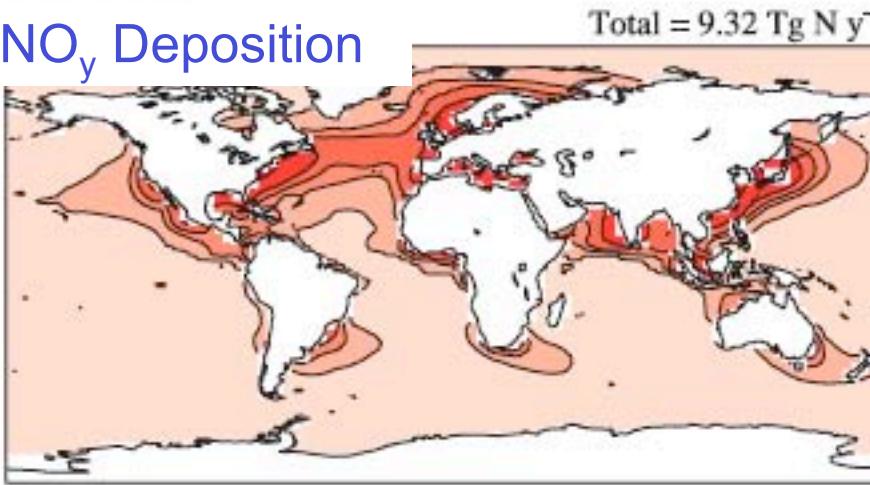
$\Delta\text{CO}_2$  Flux



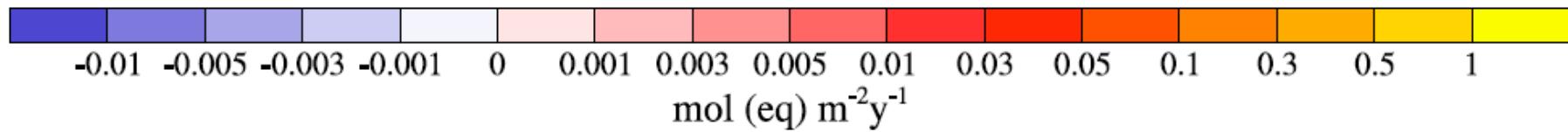
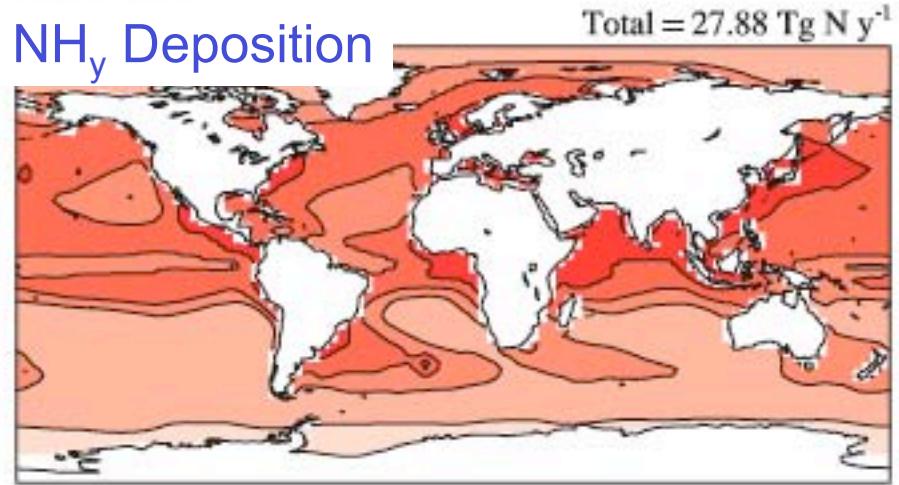
S Deposition



$\text{NO}_y$  Deposition



$\text{NH}_y$  Deposition



Doney et al., PNAS, 2007

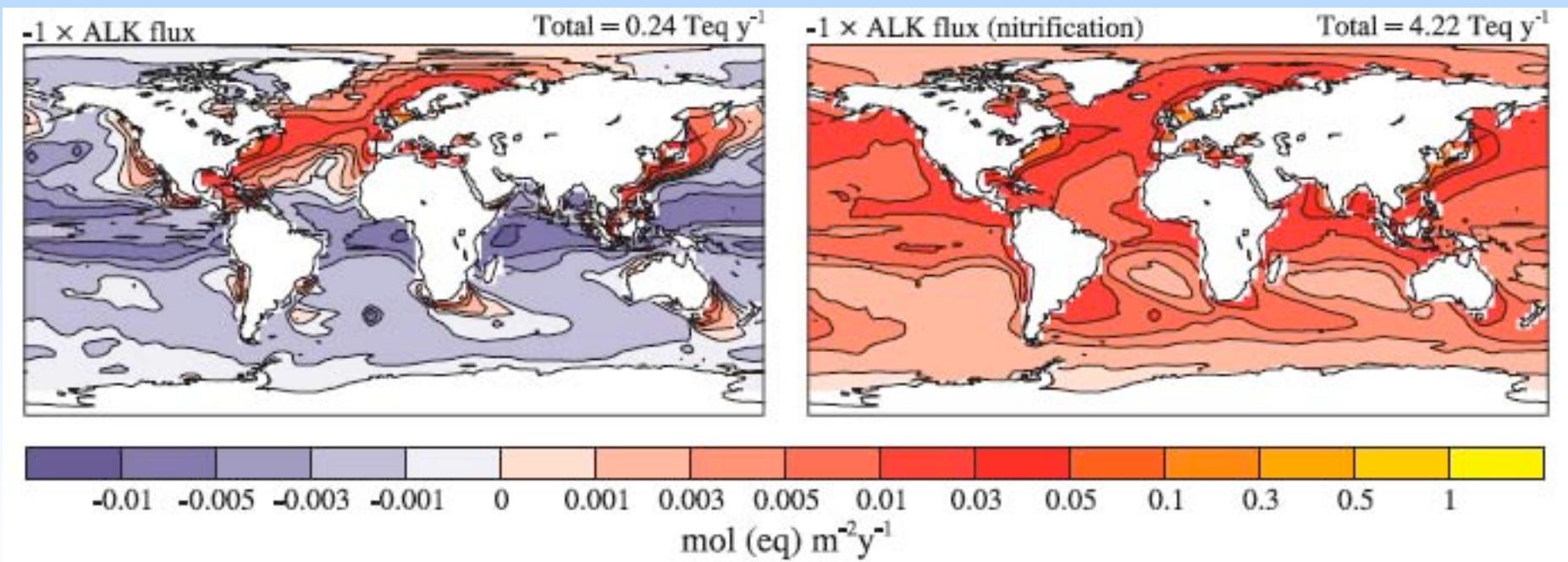


# Atmospheric Alkalinity Deposition

$$fAlk = 2fCa^{2+} - 2fSO_2 - 2fSO_4^{2-} - fNO_3^- + fNH_3 + fNH_4^+$$

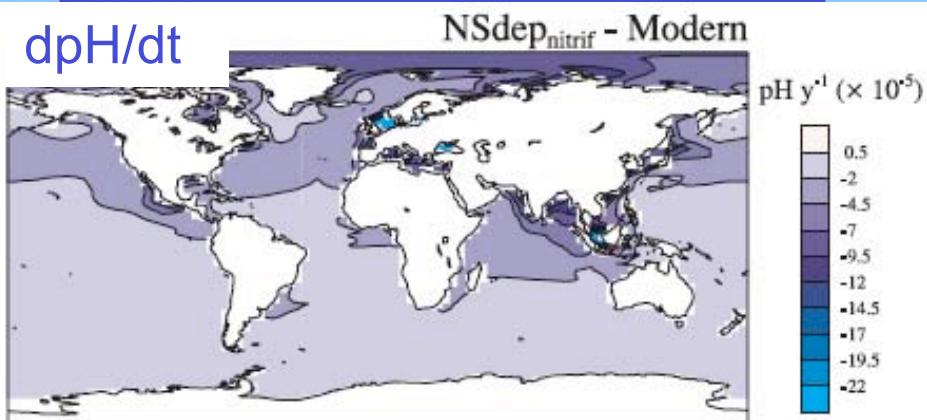
With full nitrification of  $NH_3/NH_4$

$$fAlk_{nitrif} = 2fCa^{2+} - 2fSO_2 - 2fSO_4^{2-} - fNO_3^- - fNH_3 - fNH_4^+$$

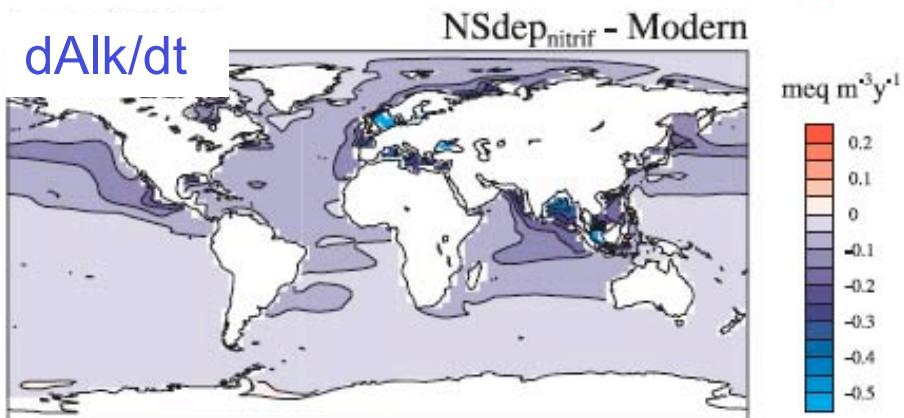


# Geochemical Responses

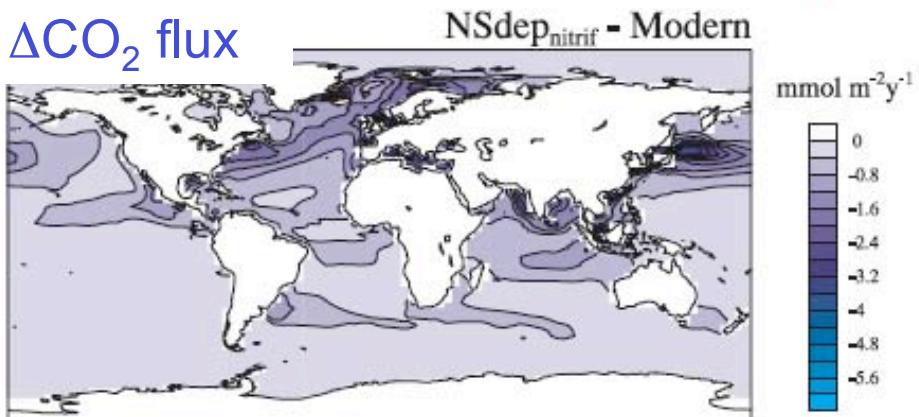
$d\text{pH}/dt$



$d\text{Alk}/dt$



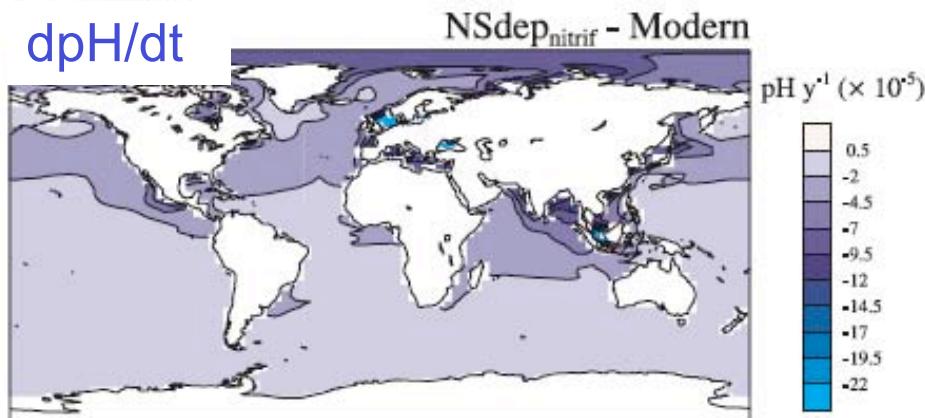
$\Delta\text{CO}_2$  flux



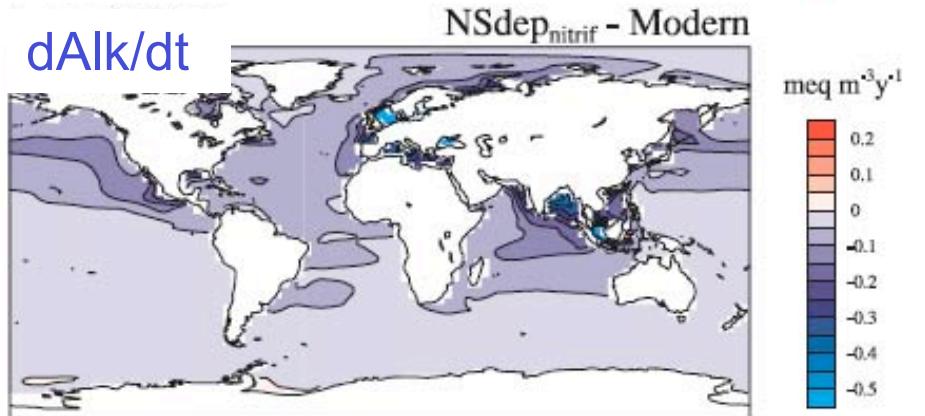
# Geochemical Responses

# w/ Nutrient Eutrophication

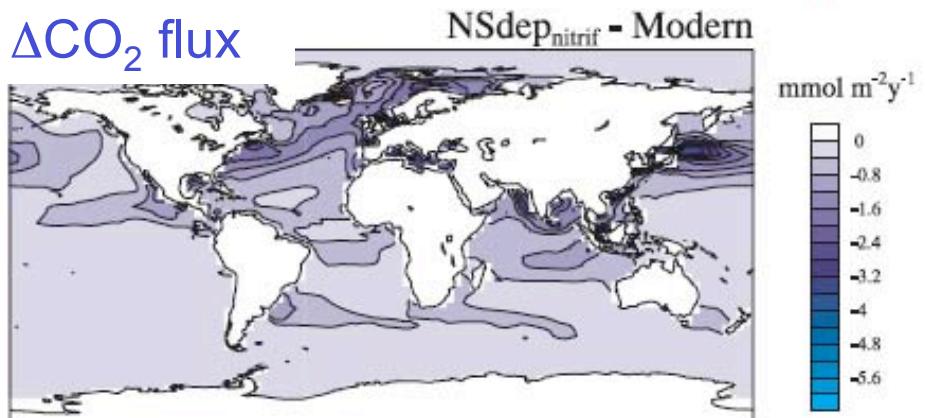
$d\text{pH}/dt$



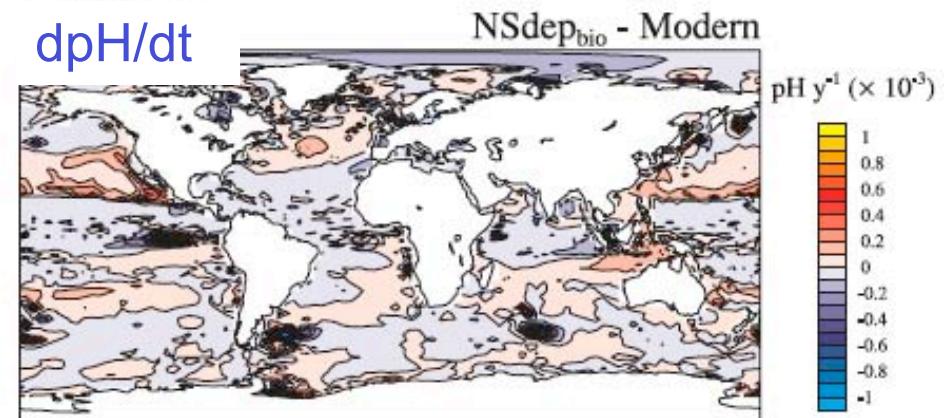
$d\text{Alk}/dt$



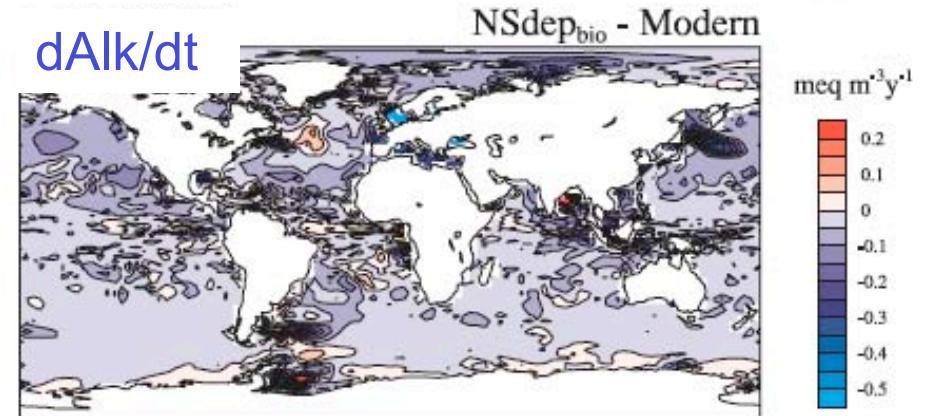
$\Delta\text{CO}_2 \text{ flux}$



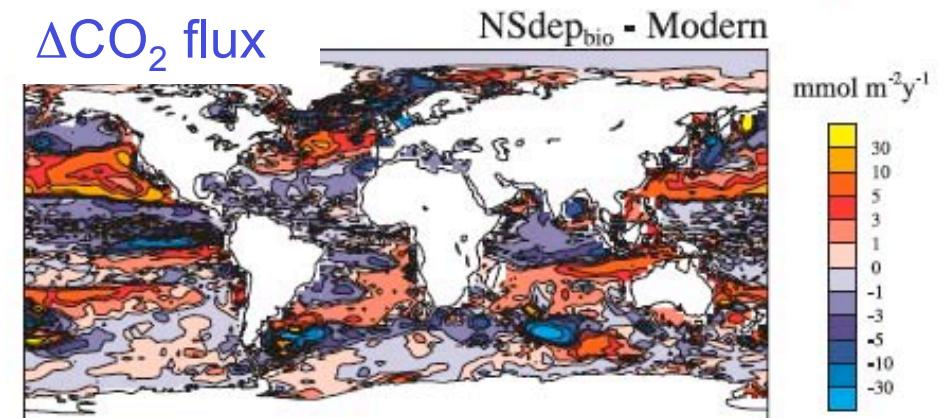
$d\text{pH}/dt$



$d\text{Alk}/dt$

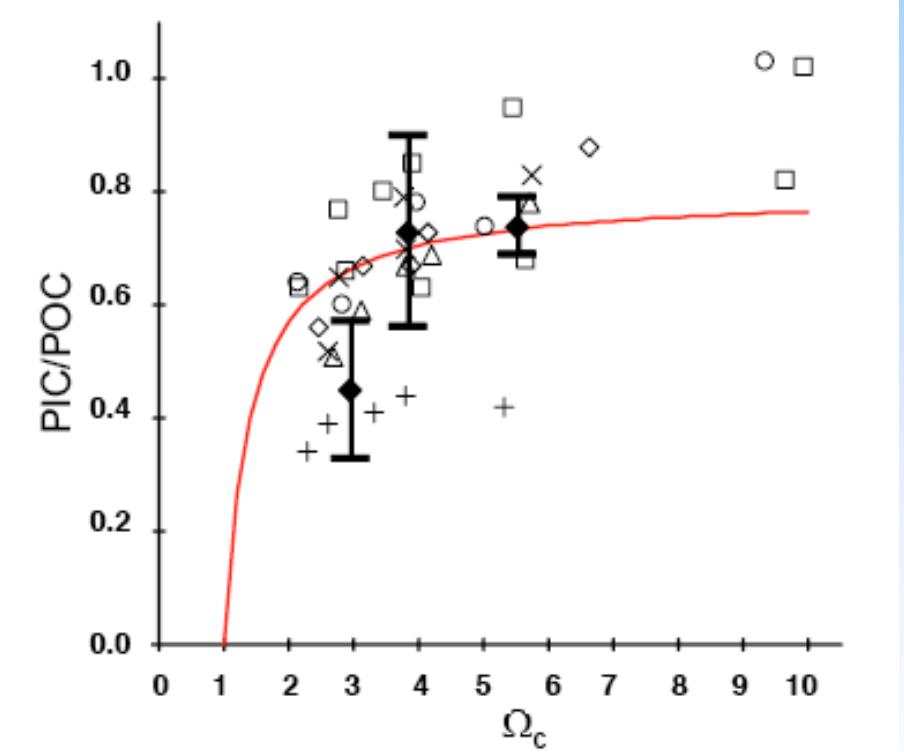
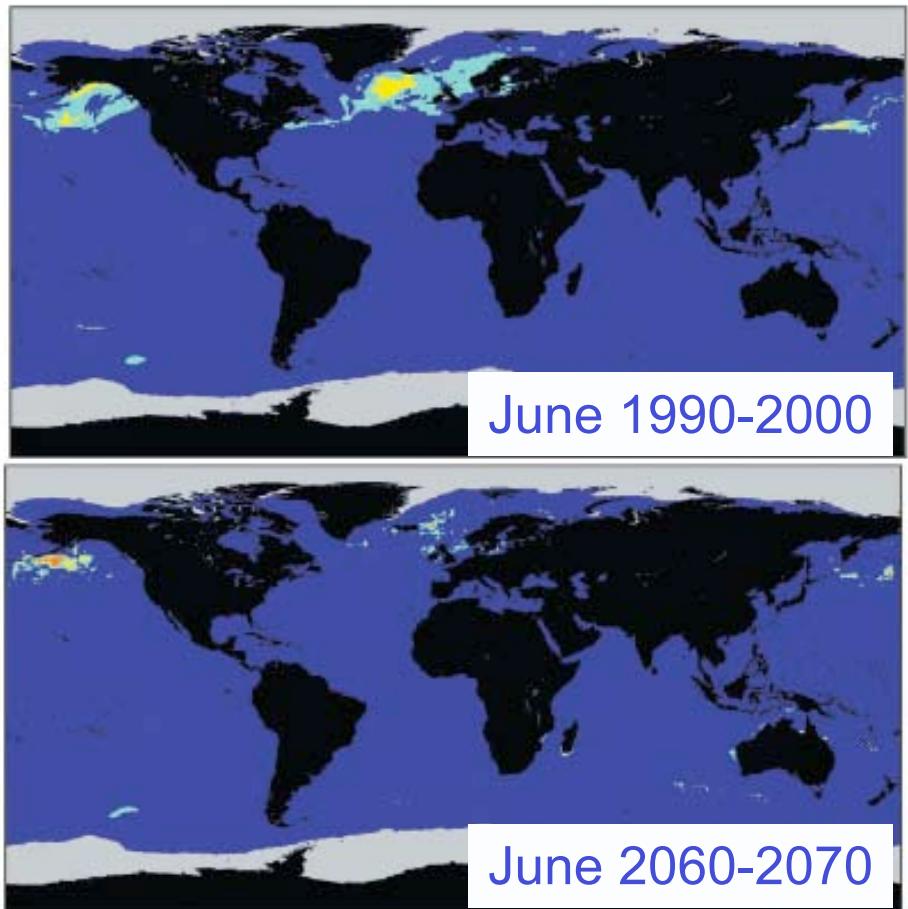


$\Delta\text{CO}_2 \text{ flux}$



Climate Forcing: Synergistic effects of changes in temperature, nutrients, trace metals, sea-ice, mixed-layer depth,  $\text{CO}_2(\text{aq})$ , pH and  $\Omega$

Biotic Interactions: Competition, predatory-prey, viruses, ...



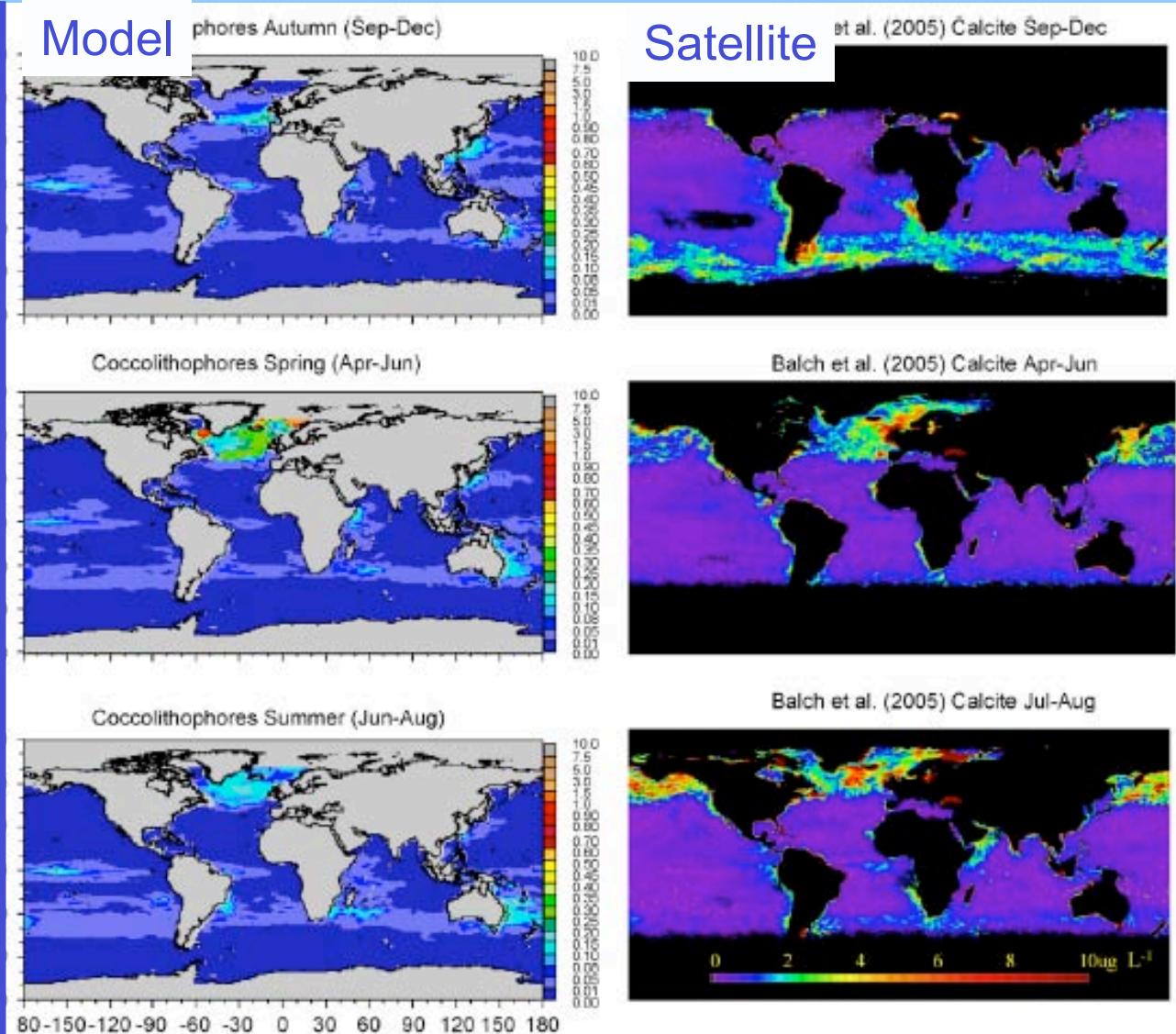
Gehlen et al. Biogeosciences (2007)

Iglesias-Rodriguez et al. Global Biogeochem. Cycles (2002)



# Calcification in Ecosystem Models

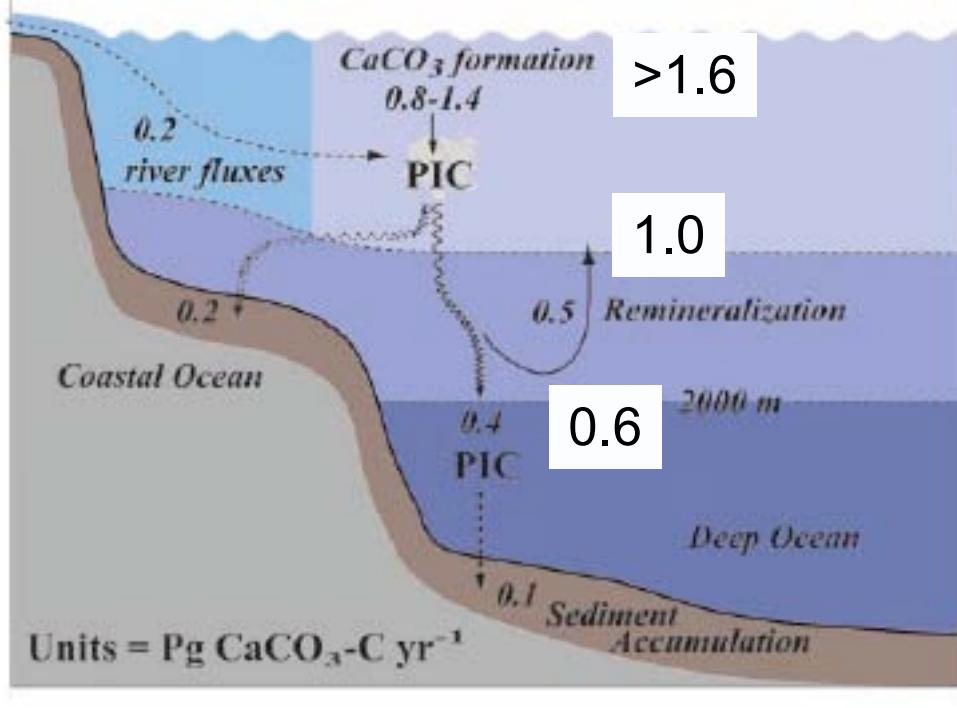
- Model approach fixed PIC/POC, statistical, dynamic
- Prognostic models (light, MLD, SST, nutrients, grazing)
- Functional groups coccilithophorids, forams, pteropods
- Data limitations for verification and parameterization (satellites, field, laboratory)



Gregg & Casey Deep-Sea Res. II (2007)  
Moore et al. (2004); Le Quere et al. (2005)



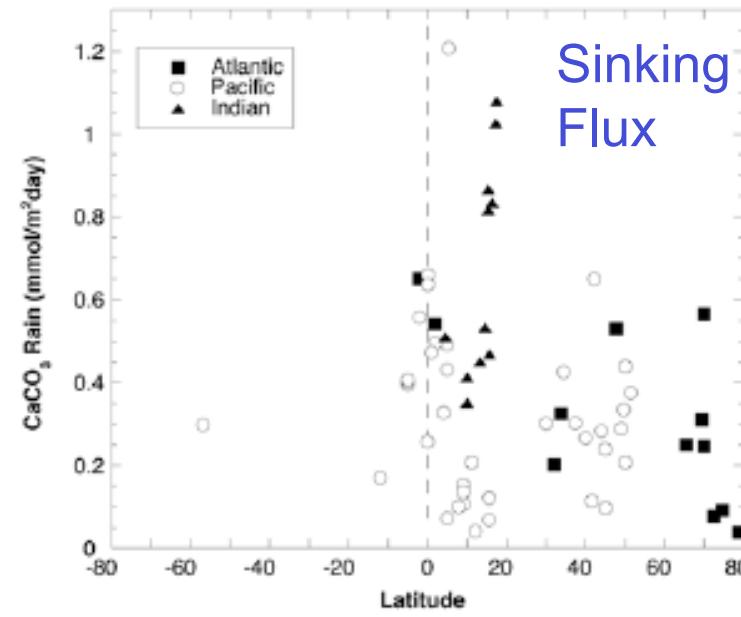
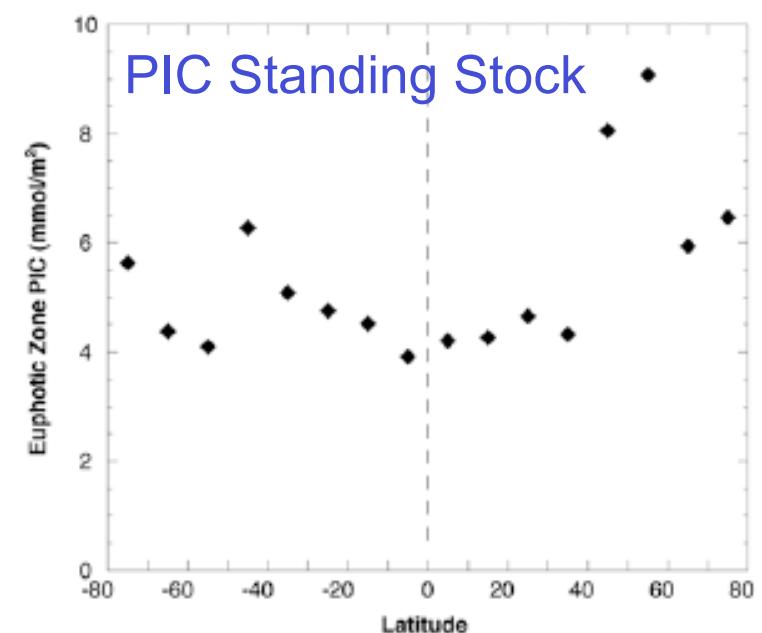
## Global Budget Pg CaCO<sub>3</sub>-C/yr



Feely et al. Science (2004);  
Berelson et al. GBC (2007);  
Sarmiento et al. GBC (2002)

Global rates & regional patterns  
of water column dissolution, flux  
from deep traps

## Geochemical constraints



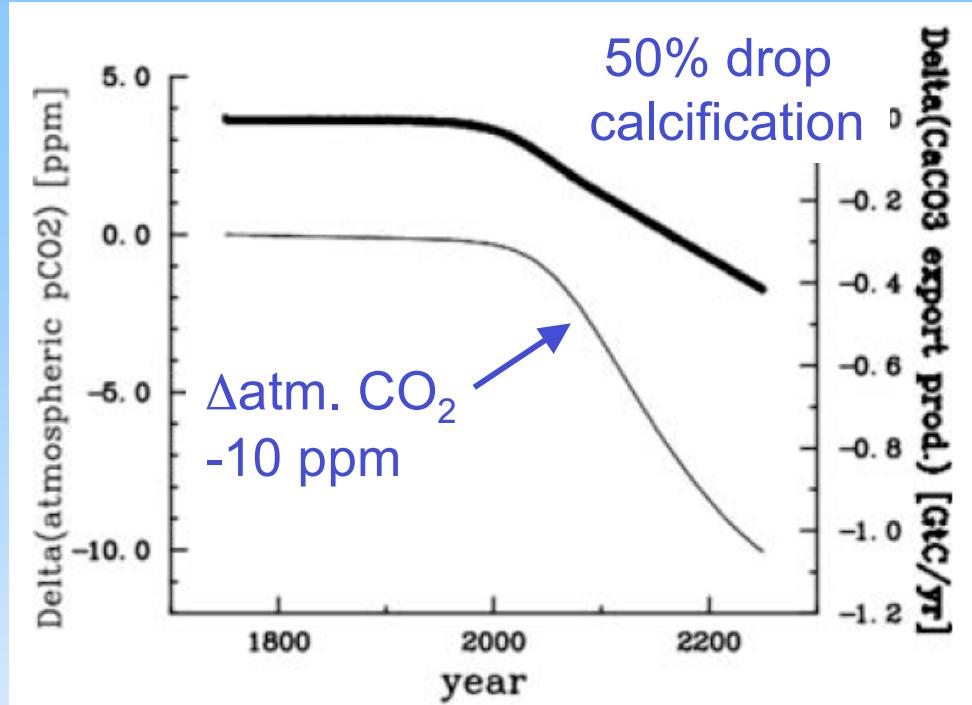
# Acidification, Calcification & Climate Feedbacks

Negative (damping) climate feedbacks

$$\Delta \text{atm. CO}_2 < 0$$

$$\Delta \text{DIC}_{\text{surf}} < 0$$

$$\Delta \text{Alk}_{\text{surf}} > 0$$



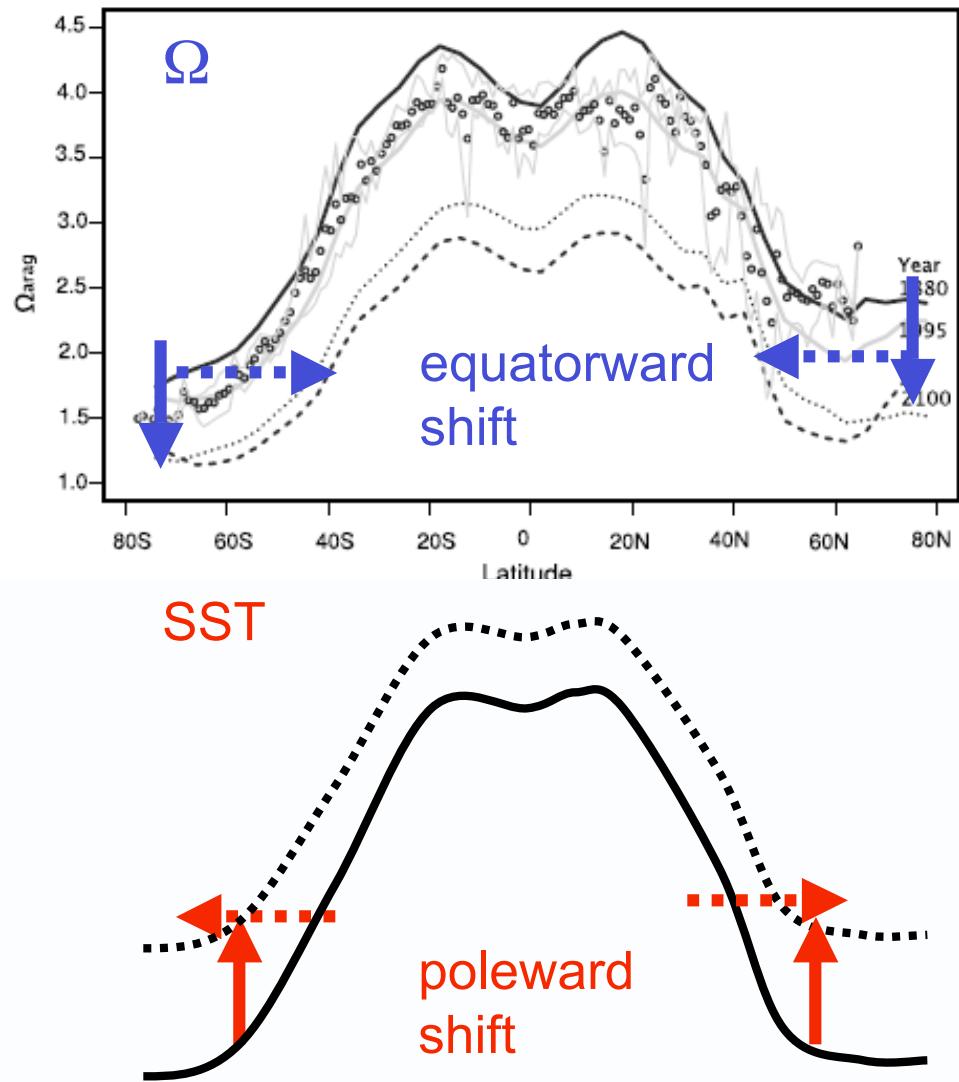
Heinze, Geophys. Res. Lett. (2004)  
Zondervan et al. Global Biogeochem. Cycles (2001); Gehlen et al. Biogeoscience (2007); Ridgwell et al. Biogeoscience (2007)

- Reduced formation of biogenic  $\text{CaCO}_3$
- Decrease organic matter remineralization lengthscale (ballasting)
- Increase subsurface  $\text{CaCO}_3$  dissolution

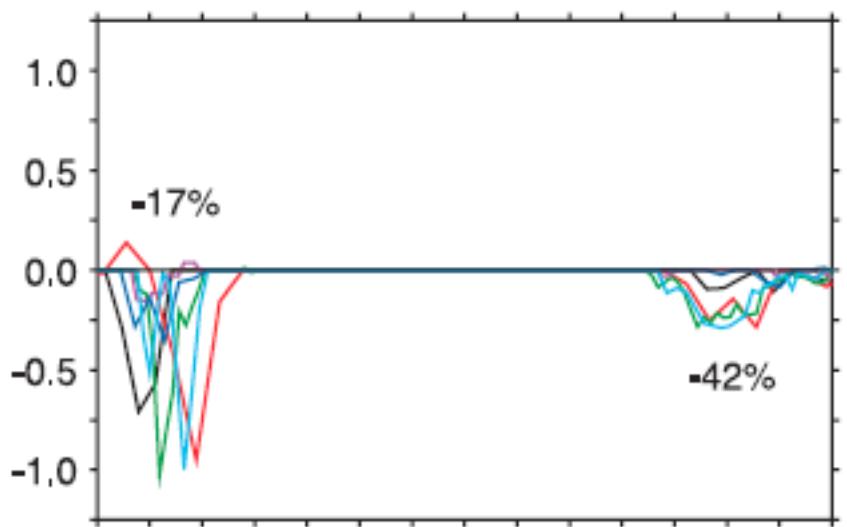


# Niches & Plasticity

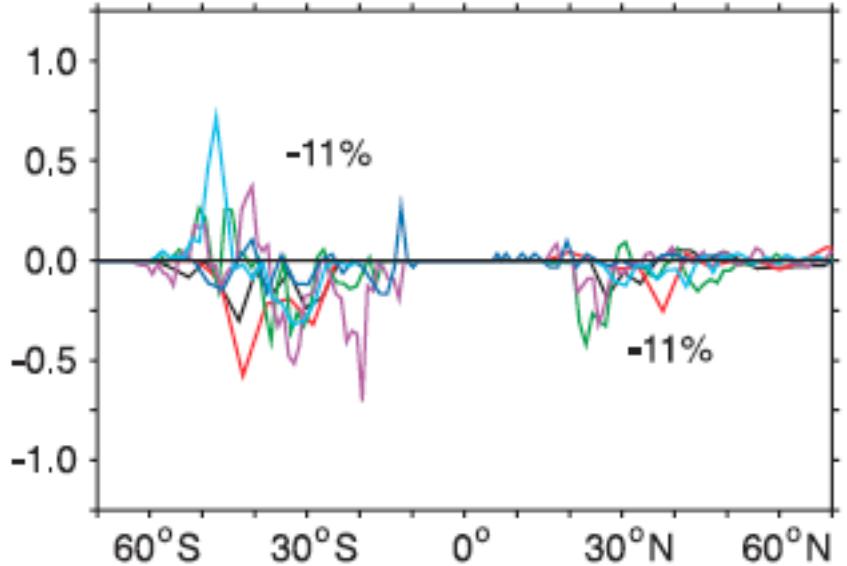
No current analogues to some future climate conditions??



## Marginal Sea Ice

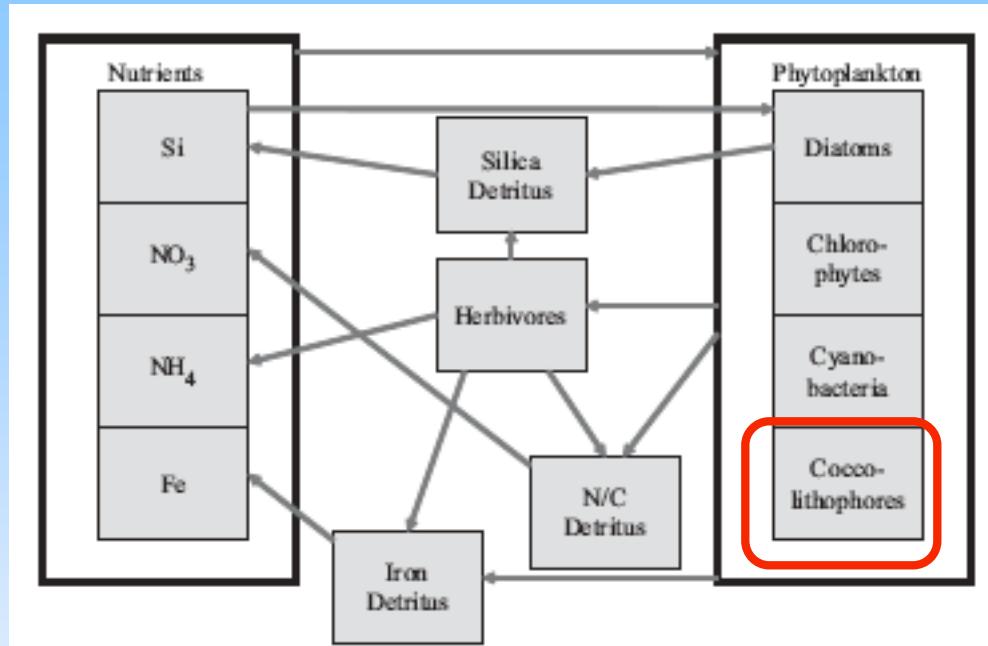


## Seasonally Stratified Subtropical Gyre



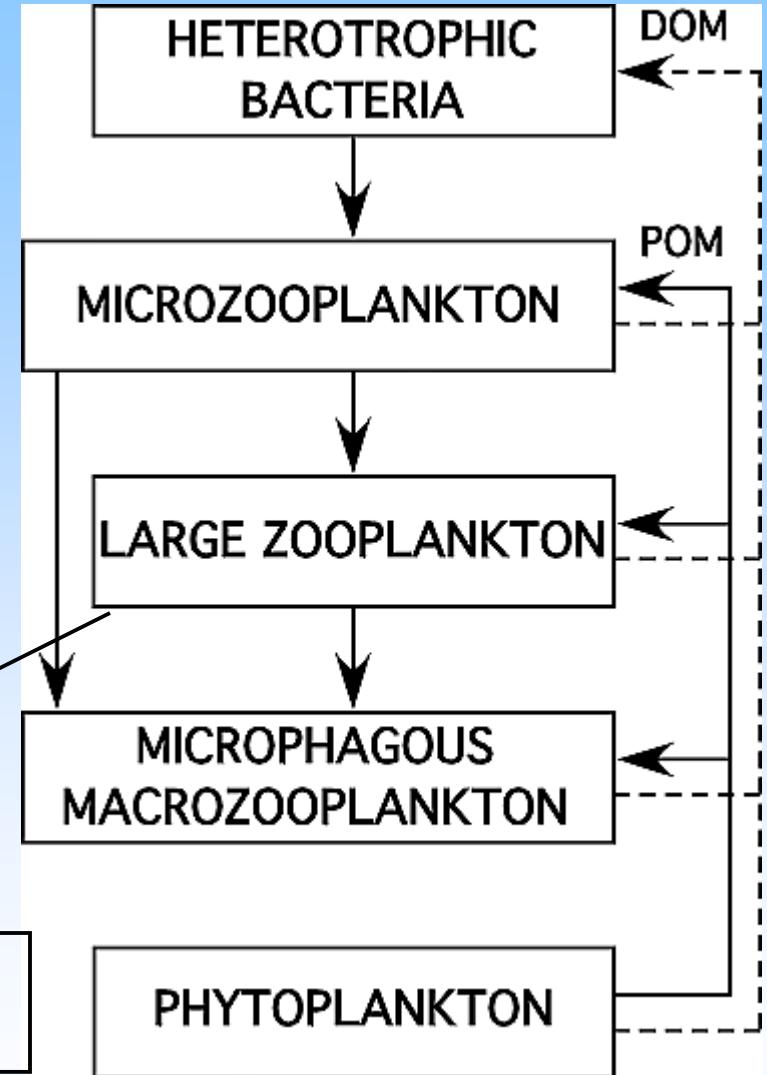
Sarmiento et al. GBC (2004)

# Food-web Interactions



Gregg & Casey Deep-Sea  
Res. II (2007)

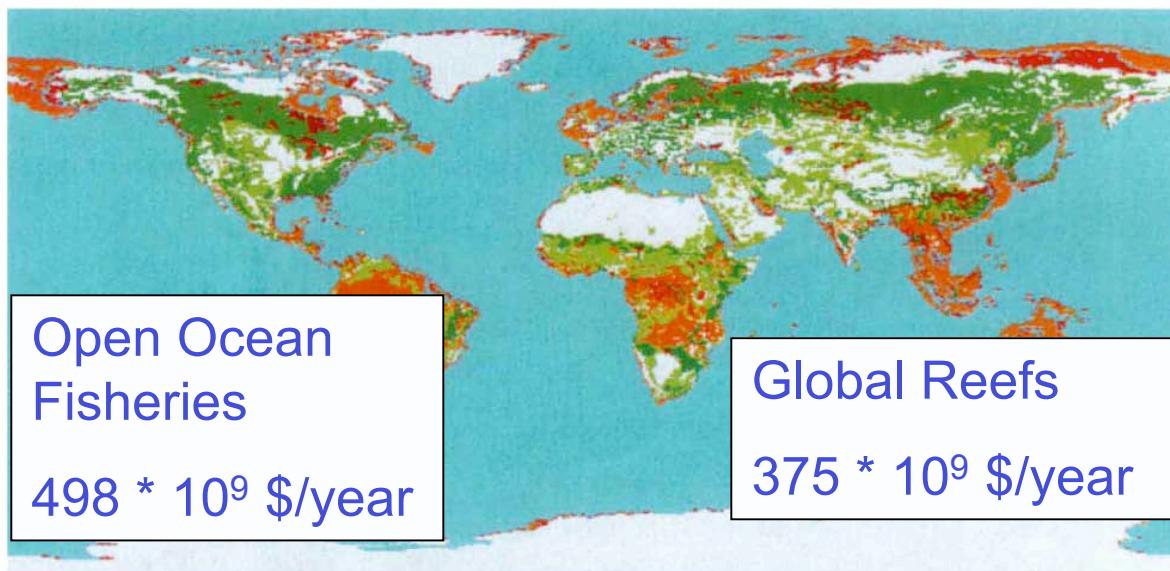
- Complicated (complex?) food-web interactions
- How do we test model forecasts? (spatial & interannual variability)



From Legendre and Rivkin (2005)



Resource	Impacted by Acidification
Fisheries	Plankton - food chain dynamics Mollusks and Crustaceans Reefs - critical habitat
Coastal Protection Ocean CO <sub>2</sub> Uptake	Reefs - Protect shore lines Erosion of CaCO <sub>3</sub> deposits Surf. alkalinity; biological pump



Costanza et al. (1997)

## Ecosystem Services

- Provide direct benefits to society
- Valuation estimates:
  - difficult to replace by technology
  - cost of total loss dramatic
  - moral choices involved in valuation
  - both market and non-market value





Special Report

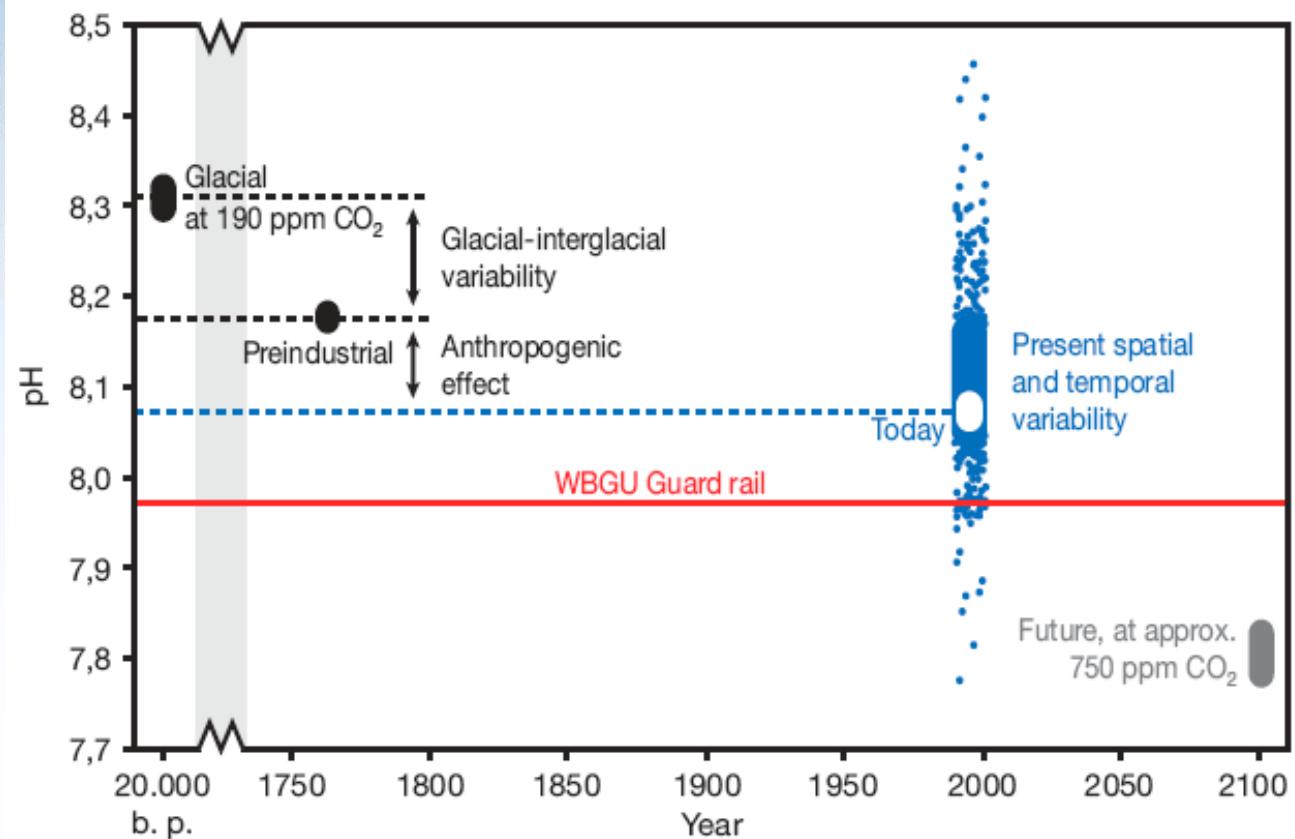
## The Future Oceans – Warming Up, Rising High, Turning Sour



R. Schubert  
H.-J. Schellnhuber  
N. Buchmann  
A. Epiney  
R. Grießhammer  
M. Kulessa  
D. Messner  
S. Rahmstorf  
J. Schmid

### 4.4.1 Proposed guard rail

To prevent undesirable or high-risk changes to the marine food web due to aragonite undersaturation (Section 4.3), the pH value of near surface waters should not drop more than 0.2 units below the pre-industrial average value of 8.18 in any larger ocean region (nor in the global mean). A pH drop of 0.2





The Center for Biological Diversity respectfully requests that the North Coast Water Quality Control Board recommend that:

**All ocean waters under Region 1's jurisdiction be included in the state List of Impaired Waters ("303(d) List") under section 303(d) of the Clean Water Act as impaired for pH due to absorption of anthropogenic carbon dioxide pollution.**

[5] The U.S. Environmental Protection Agency [1976] Quality Criteria for Water state: "For open ocean waters where the depth is substantially greater than the euphotic zone, the pH should not be changed more than 0.2 units outside the range of naturally occurring variation ..." Atmospheric CO<sub>2</sub> concentrations would need to be stabilized at <500 ppm for the ocean pH decrease to remain within the 0.2 limit set forth by the *U.S. Environmental Protection Agency [1976]* (Table 1).

Caldeira et al. Geophys. Res. Lett. (2007)

## -anthropogenic $CO_2$ , pH & $CaCO_3$ saturation state $\Omega$

- fossil fuel emissions & atmosphere  $CO_2$  (med.)
- ocean pH &  $\Omega$ ; surface (high) & subsurface (med.)
- small climate/carbon feedbacks (med.)

## -other acidity/alkalinity inputs

- atmosphere N & S => reduce coastal Alk & outgas  $CO_2$  (med.)
- sediments, rivers, & groundwater (low/med.)

## -biogeochemical and ecological impacts

- lower calcification + reduced subsurface dissolution  
=> modest atm.  $CO_2$  drawdown (med.)
- altered particle ballast, elemental stoichiometry, trace gases (low)
- population adaptation and plasticity
- effects on foodwebs & higher trophic levels (low)

## -policy, economic & social dimensions

- atm.  $CO_2$  guard rail (0.2 pH drop?) (low/med.)
- economic value fisheries, coral reefs, biodiversity (low)
- adaptation strategies for resource managers