Living Large in Microscopic Nooks

Newly discovered deep-sea microbes rearrange thinking on the evolution of the Earth—and life on it

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Between a rock and a hard place is the proverbial worst spot for people to find themselves in. But for certain deep-sea microbes, it’s the place to be. In 2000, to our surprise, we found that microscopic nooks and pits within volcanic seafloor rocks harbor abundant colonies of previously unidentified microbes.

These microbes are different from other microorganisms living in the sunless depths. They do not obtain the energy they need to grow and multiply by metabolizing chemicals dissolved in seawater or in hydrothermal fluids venting from the seafloor. Instead, these newly discovered microbes are living directly off minerals in solid seafloor rocks.

The microbes are oxidizing iron in the rocks, chemically altering the rocks, and harnessing the energy produced by this chemical reaction to live. Their discovery has raised a slew of intriguing questions:

• Does our planet sustain abundant and ubiquitous populations of these microbes?
• Do they play a pivotal role in chemically altering Earth’s crust?
• Were they pioneering life forms on an early Earth, which was largely devoid of oxygen but full of iron?
• Do they exist on other iron-rich, oxygen-
poor planetary bodies such as Mars?

These previously inconspicuous microorganisms may turn out to have starring roles in shaping the evolution of life on Earth and other planets, and shaping the evolution of the planet itself.

So why didn’t we notice them before? Beyond the inherent difficulties and expense of searching for microorganisms at the bottom of the ocean, the answer is that we hadn’t really looked for them before. But now these easy-to-overlook microbes have become hard to ignore.

**Pumping iron on the seafloor**

More and more, we are learning how life on the Earth and the Earth itself—biology and geology—are intimately intertwined and evolve together. Microbes are ubiquitous catalytic agents, sparking chemical reactions that alter the physical and chemical properties of their surroundings. Beyond our scope of vision, their cumulative metabolic activities play a fundamental role in shaping and regulating our environment. (Our world would be completely different, for example, if microorganisms did not continuously decompose organic matter and transform it back into inorganic material.)

A new field has arisen called geomicrobiology. Scientists are now taking a closer look at many unexplored regions of our planet, and other planets, searching for populations of unknown microbes that may play major roles in cycling chemicals through planetary systems.

In geomicrobiology, the borders between rocks and living things are not so ironclad. Many rocks are, however, and the microbes we found steal electrons from iron atoms in the rock, changing them from ferrous \((\text{Fe}^{+2})\) to ferric \((\text{Fe}^{+3})\). With the energy produced by this chemical reaction, they convert carbon dioxide (from seawater) into organic matter—much the way plants and plankton use solar energy and photosynthesis to accomplish the same.

**Microscopic, but mighty**

Iron is one of the most abundant and reactive elements in the environment near Earth’s surface, so the discovery of iron-oxidizing microbes raises the potential that massive communities of them may exist on Earth. If so, they could continually extract huge amounts of carbon dioxide from seawater and microscopically exert a huge influence on ocean chemistry over geologic time.

Does this large-scale drawdown of carbon dioxide from seawater help the oceans absorb carbon dioxide, a critical greenhouse gas, from the atmosphere? If so, it would revise our understanding of how carbon cycles through the planetary system—perhaps giving iron-oxidizing microbes an important, previously unknown role in the evolution of Earth’s climate.

In their own way, the rise of microscopic photosynthetic plants caused one of the most devastating, permanent alterations in all of Earth’s history. They changed the chemical composition of the near-surface environment that all life depended on, by simply pumping oxygen into Earth’s atmosphere.

Before then, neither the atmosphere nor the oceans contained much oxygen, but the oceans were filled with iron-rich rocks and tons of dissolved iron. In such an iron-rich, oxygen-poor environment, iron-oxidizing microbes may have been dominant, pioneering life forms—a concept that compels us to reassess our thinking about the evolution of life on early Earth.

The existence of iron-oxidizing microbes also redirects our search for life elsewhere in the universe. Similar microbes could have thrived, or still thrive, in other iron-rich, oxygen-poor locales—such as Mars, with its red, iron-rich soil, or on the volcanic seafloor below the ice-covered ocean of Jupiter’s moon, Europa.

**A search for unknown life**

These unexpected new lines of inquiry began in 2000 when former WHOI Postdoctoral Scholar Tom McCollom and I, with funding from the Mellon Foundation and the National Science Foundation (NSF), joined a research cruise aboard R/V *Atlantis* off the Oregon coast.

Since the late 1970s, when hydrothermal vents were discovered, scientists have focused on deep-sea chemosynthetic microbes that derive energy from dissolved hydrogen, hydrogen sulfide, and methane emitted from these sites. Though it is easier for microbes to draw
energy from chemicals dissolved in seawater, WHOI biologist Carl Wirsen and others had found evidence of sulfur-oxidizing bacteria that used solid minerals as their only source of energy. (See “Is Life Thriving Beneath the Seafloor?” page 72.)

Enormous amounts of sulfur and sulfides are found in vent chimney rocks, in broken chimney rubble on the seafloor, and in fine-grained mineral particles that precipitate and “rain” out of plumes of hydrothermal fluids spewing out of chimneys. We speculated that this little-recognized but potentially large source of chemical energy may sustain important microbial communities, which, in turn, could play pivotal roles in altering the chemistry of seafloor rocks and the ocean itself.

Our goal in 2000 was to try to identify unknown microbes that live off solid minerals and that might be mediating large-scale geochemical changes on Earth.

The perfect niche for microbes

To explore what might be down there, we used the submersible Alvin to place a variety of microbe-free samples of natural seafloor rock back on the seafloor. Our aim was to see what might “grow” on these “blank slates.”

WHOI geochemist Meg Tivey retrieved our experimental samples for us during an Alvin dive two months later (See “The Remarkable Diversity of Sea-floor Vents,” page 60). To our surprise, we found that many of the samples had thick burnt-orange coatings of oxidized iron (or “rust”).

Using a scanning electron microscope, we saw that the surfaces of the samples were scarred with abundant pits and pores less than 20 microns (0.0004 inches) deep and wide. In these tiny pits were large accumulations of corkscrew-shaped stalks made of iron oxide, which created the thick rusty coating.

Here’s what we believe is happening: Iron-oxidizing microbes exploit a niche where the chemistry is just right. At first, oxygen-loving microbes move into the pits. They consume the available oxygen, which is not replenished because seawater does not readily flow into the restricted pit areas.

That creates an ideal situation for the iron microbes, which need low-oxygen conditions. The tiny sheltered coves within seafloor rocks contain just enough oxygen from seawater for the iron microbes to respire, but not an overwhelming amount that would oxidize all the iron—without microbial intervention—before the microbes could use it.

As a byproduct of their iron-oxidizing process, the microbes produce bundles of iron-oxide stalks that resemble a little girl’s braids. These stalk accumulations effectively cap the pits, maintaining the iron microbes’ preferred low-oxygen environment and securing their turf.

FeMO—a microbial observatory

The rapid proliferation and sheer abundance of these iron microbes and...
Katrina Edwards grew up in central Ohio, where she pursued an initial early career in the family business of running a small municipal airport just north of Columbus. She spent several years assisting her father and siblings in general airport operations (graduating to the role of chief flight instructor), which she continued as she pursued a bachelor’s degree in geology at Ohio State University. Edwards then “retired” to attend the University of Wisconsin, Madison, where she earned a Ph.D. in geomicrobiology—the first degree in this field ever awarded by the university. Edwards and her family moved to Massachusetts in 1999 to join WHOI, where she established a geomicrobiology lab. It focuses on “the tooth decay of the solid Earth,” she says, or more specifically, the transformation and degradation of Earth materials (rocks, minerals, organic matter) by microbes. Edwards now enjoys deep-sea exploration, as long as someone else “flies” the submarine and she can focus on geomicrobiological research.