

What can we learn from natural iron sources ?

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Thanks to the KEOPS and CROZEX teams

What can we learn from natural iron fertilization ?

NEWS

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Only mother nature knows how to fertilize the ocean

Web site Nature, Q. Schiermeier



Natural iron fertilization : what is it?

Two criteria :

enhanced supply of iron (field data) and
enhanced biological activity

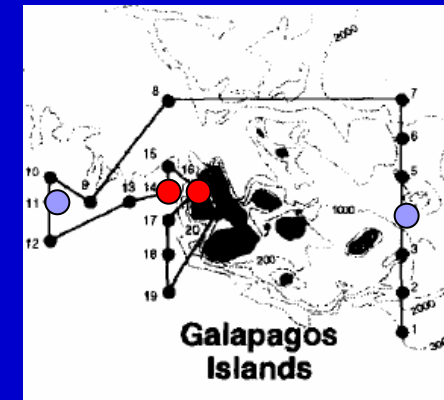
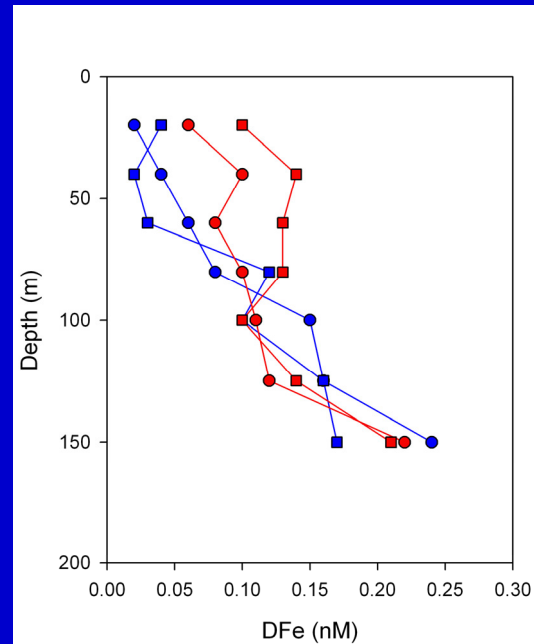
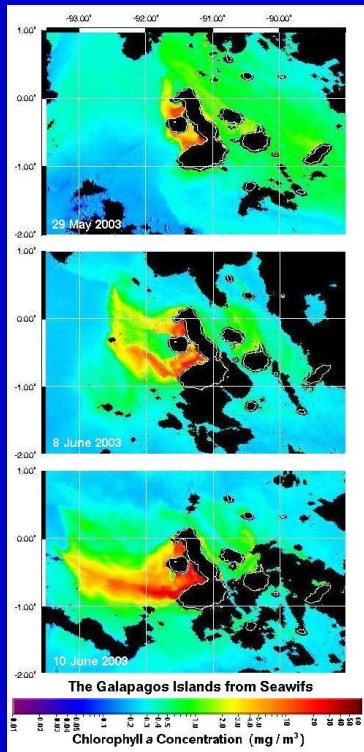
compared to the background of the region

Is natural iron fertilization a common
process in the ocean ?

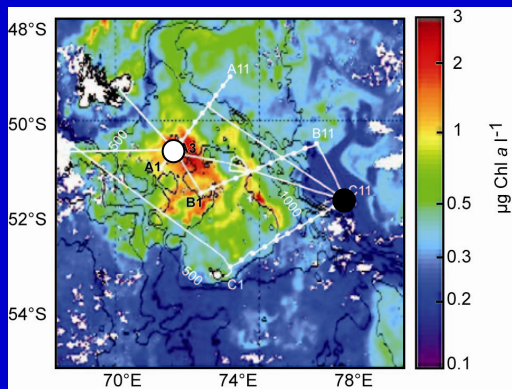


“...In conjunction with this research a study will be made of the HNLC water west of the Galapagos Islands. This area is of interest since it appears to represent a natural enrichment experiment.”

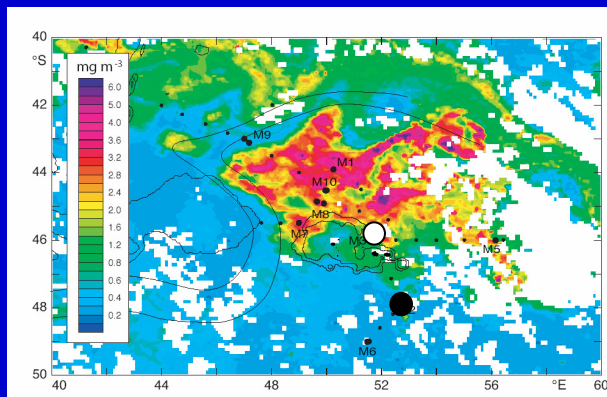
Summary of the NSF proposal, J. Martin 1992.



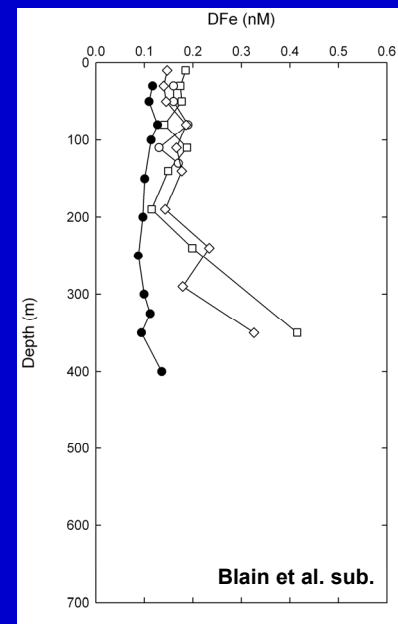
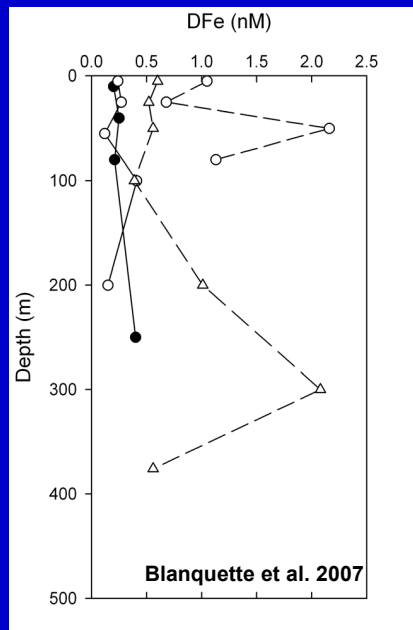
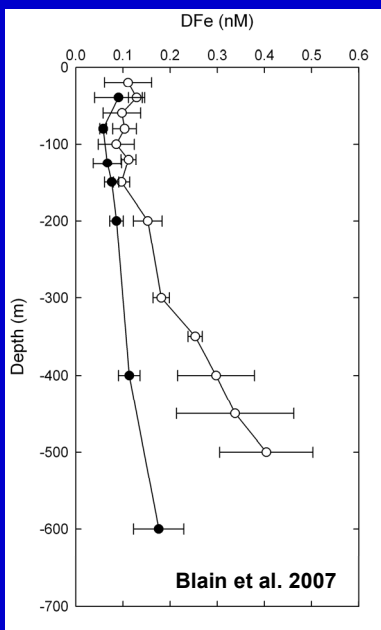
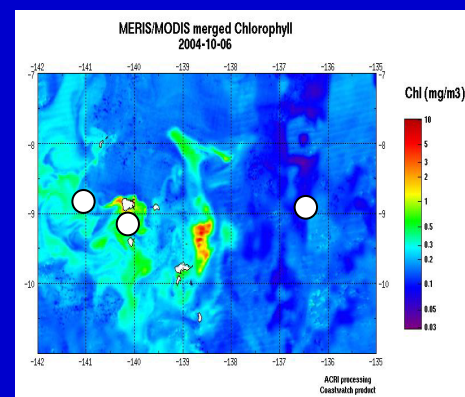
Kerguelen I.



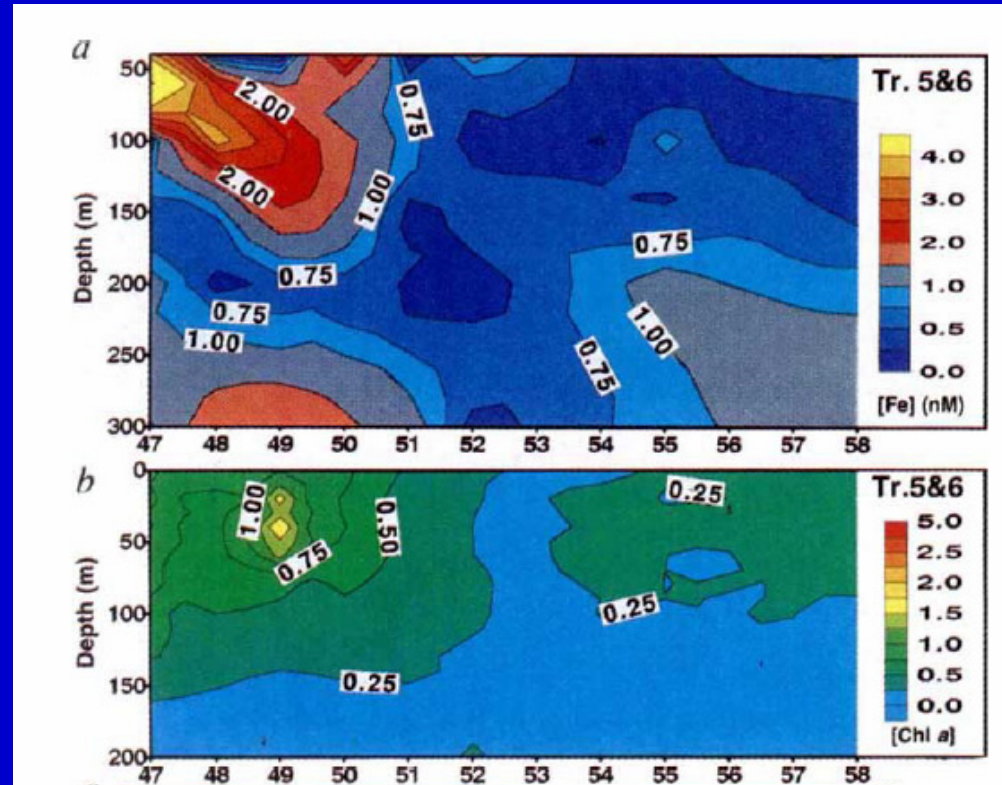
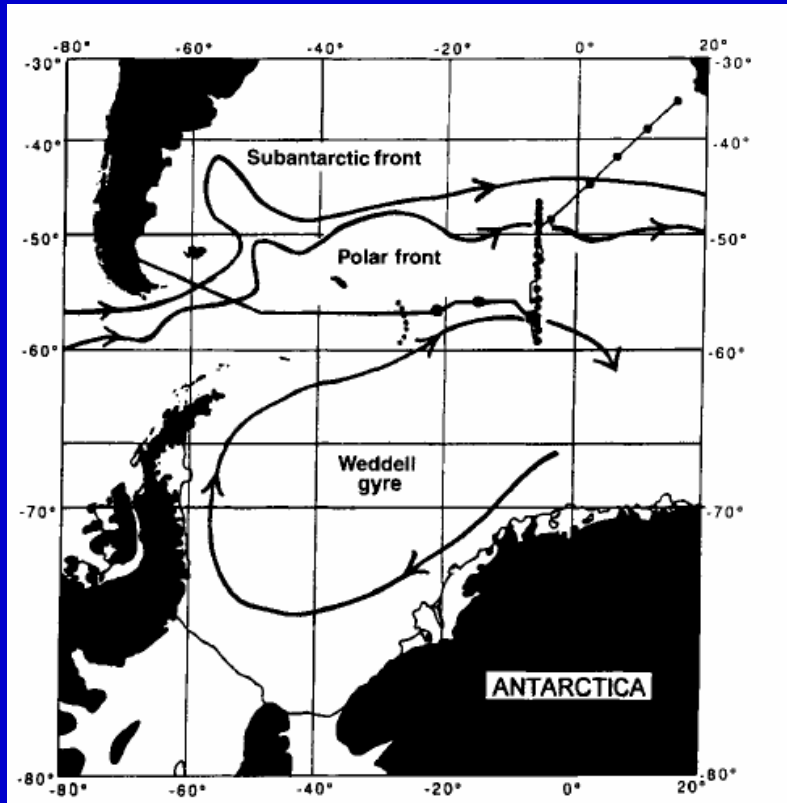
Crozet I.



Marquesas I.

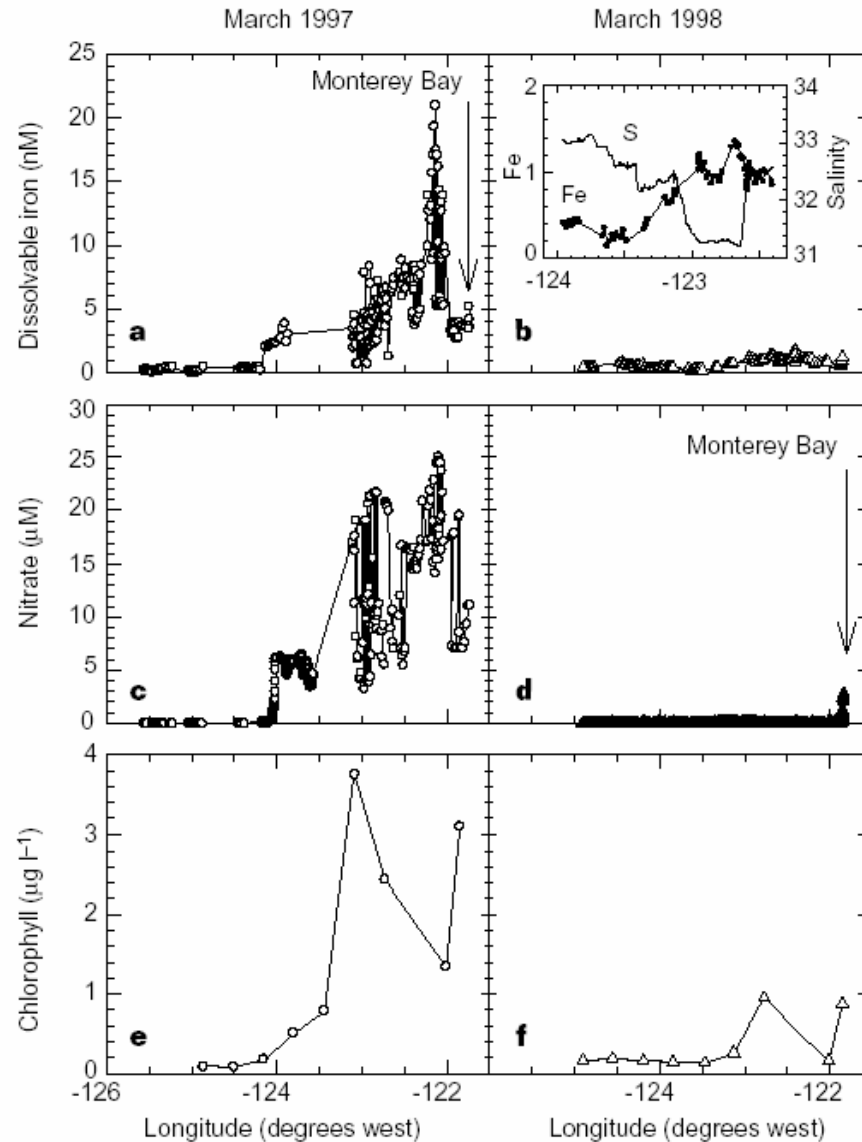
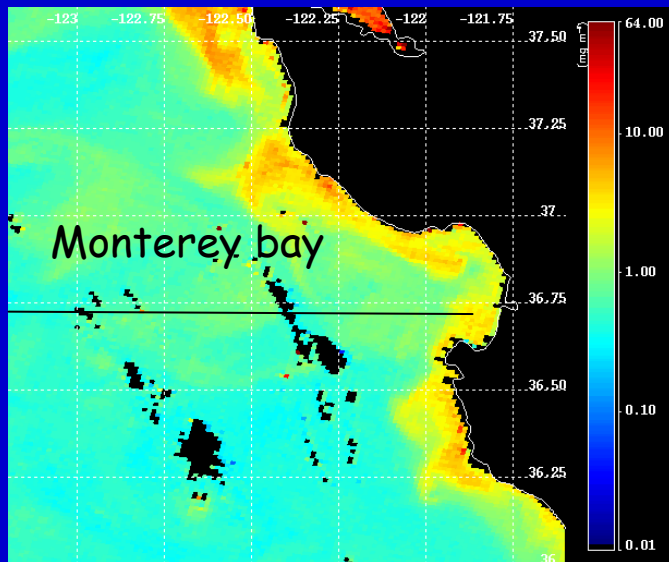


Polar Front in the Atlantic Southern Ocean

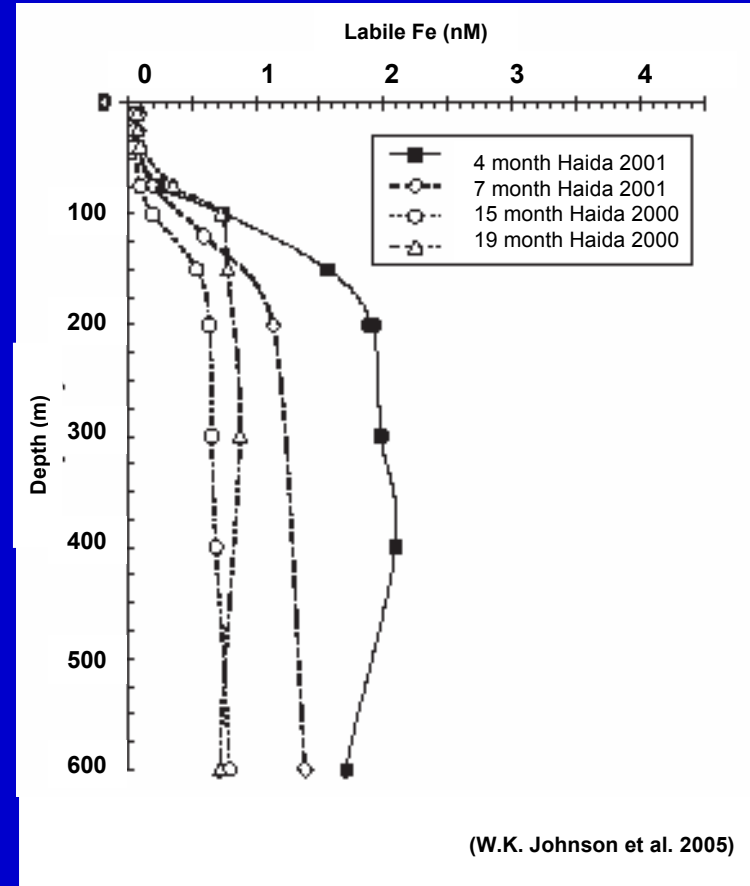
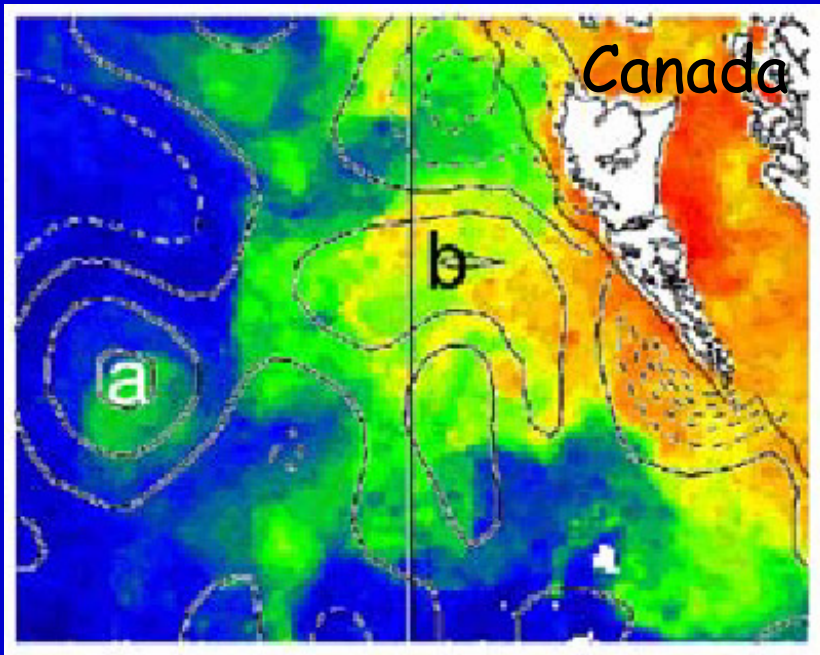


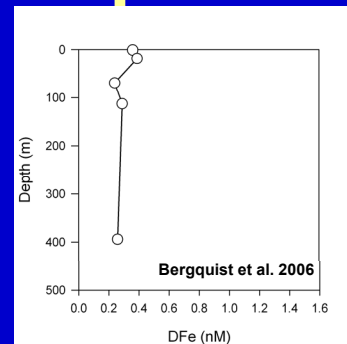
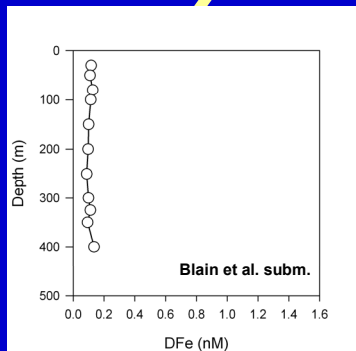
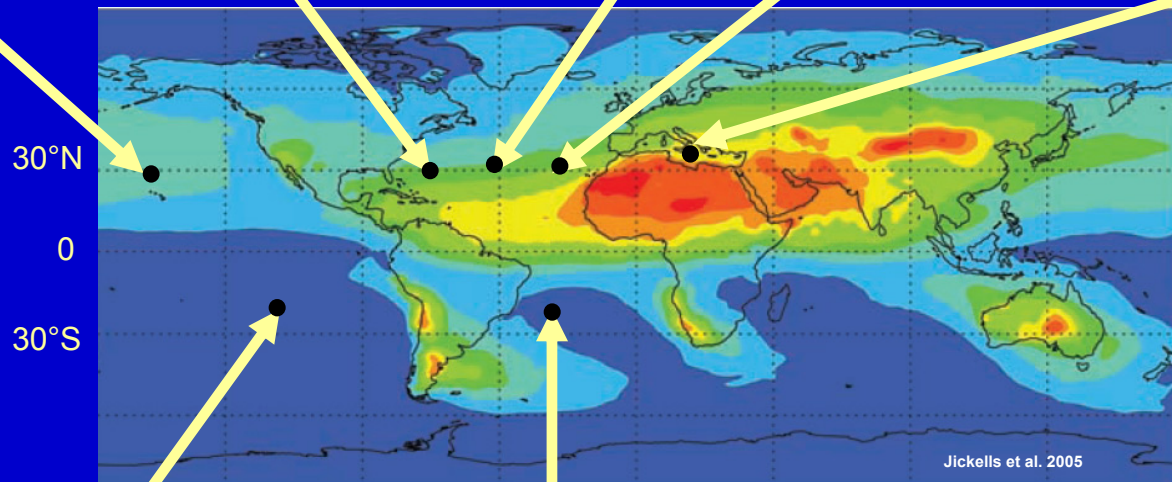
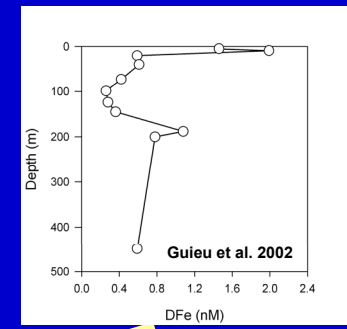
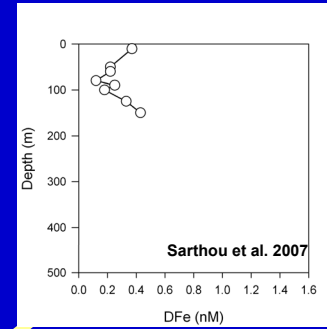
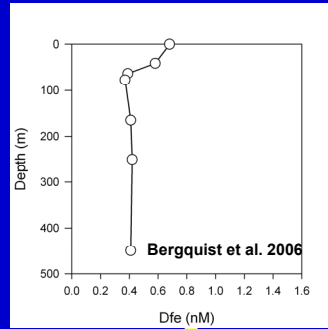
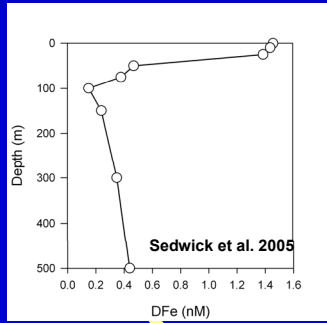
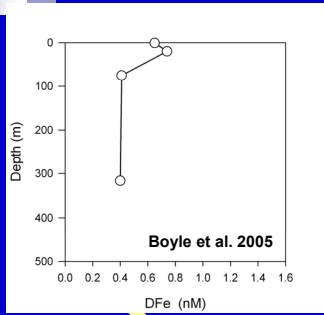
(de Baar et al. 1995)

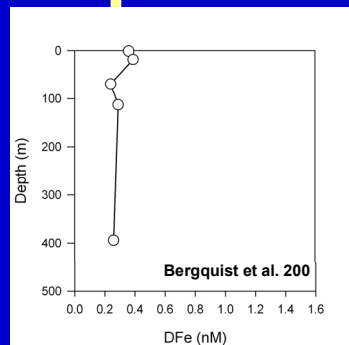
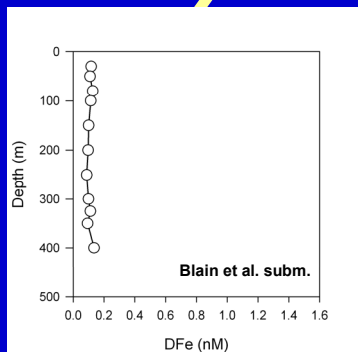
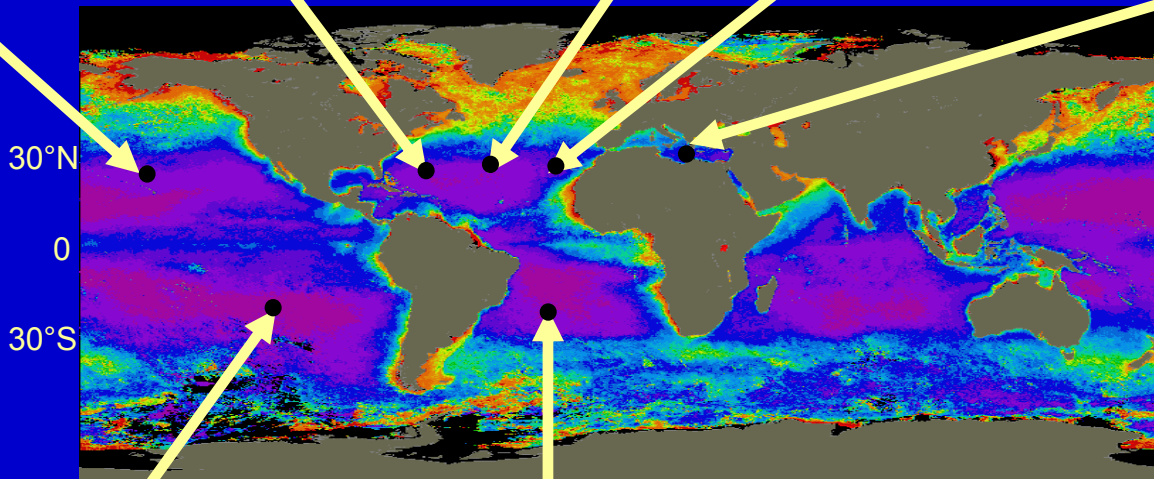
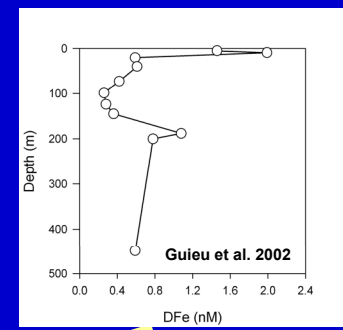
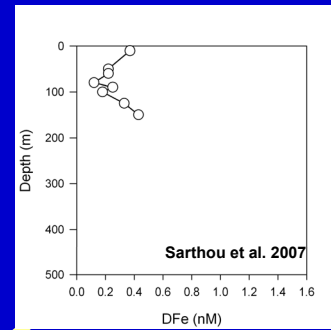
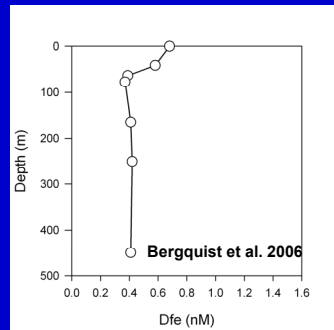
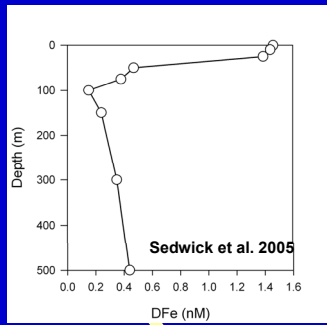
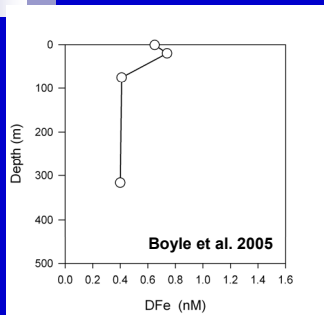
Californian upwelling

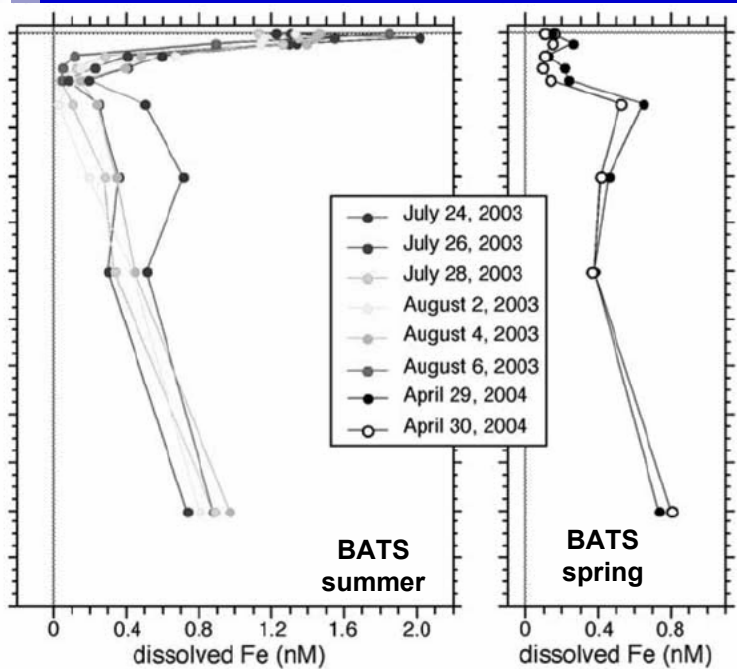


Anticyclonic mesoscale Haida eddies in the eastern North Pacific Ocean

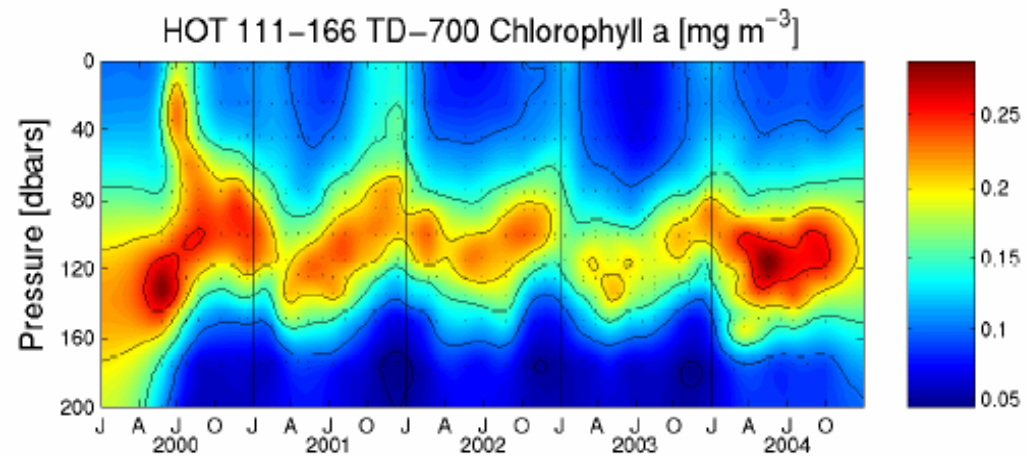




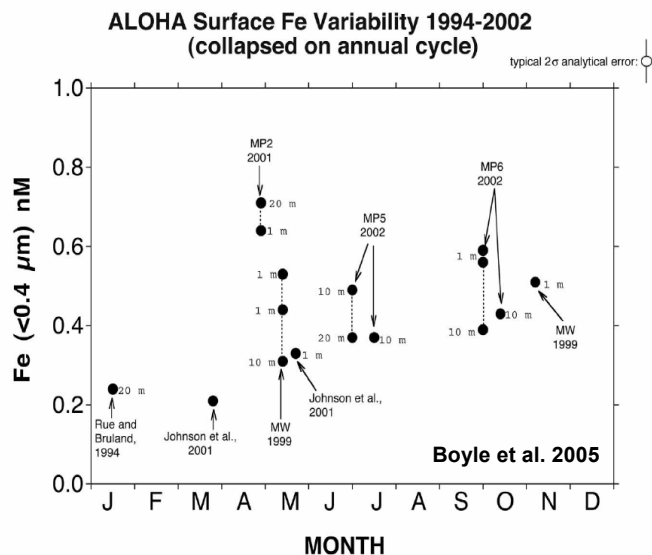




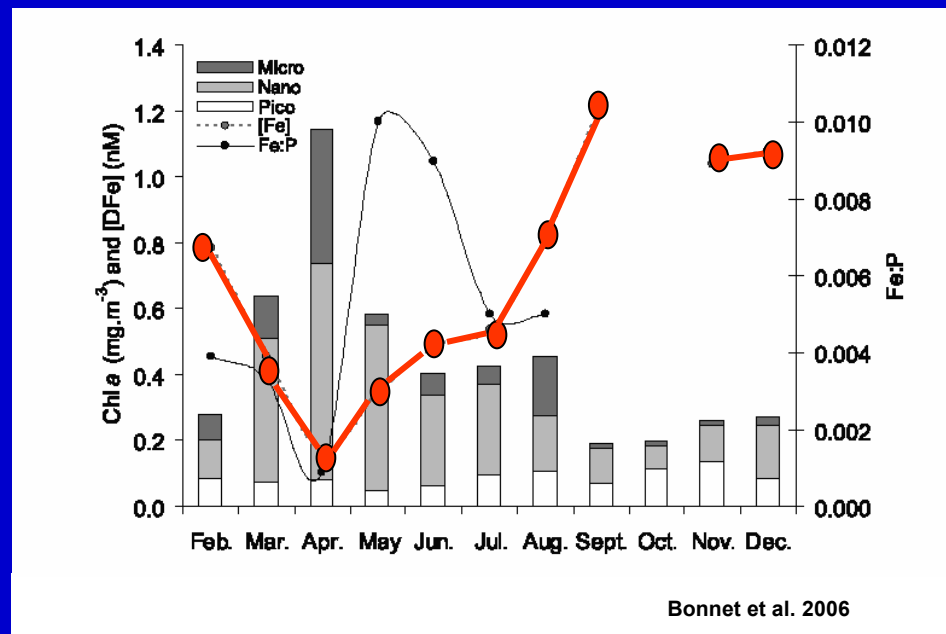
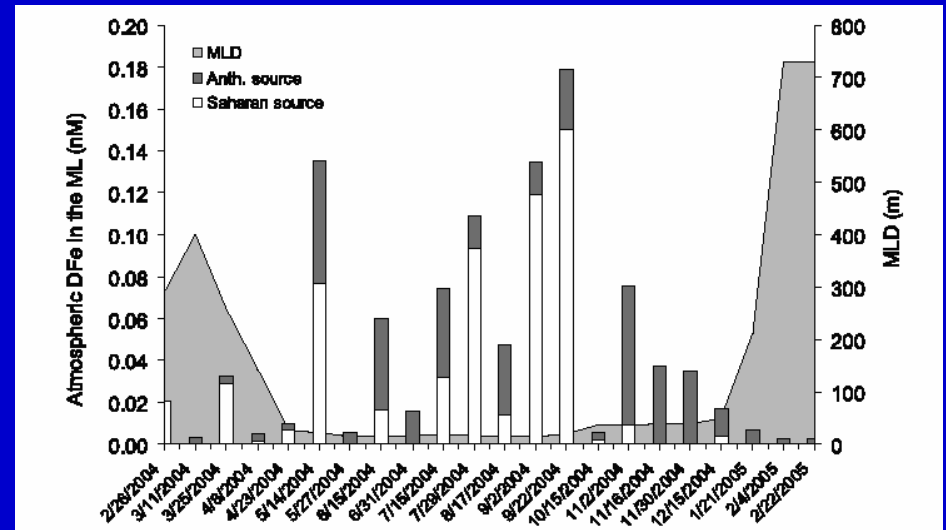
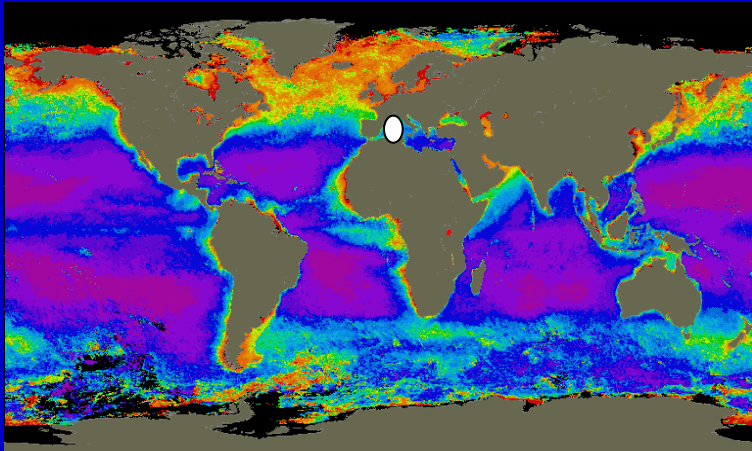
Sedwick et al. 2003



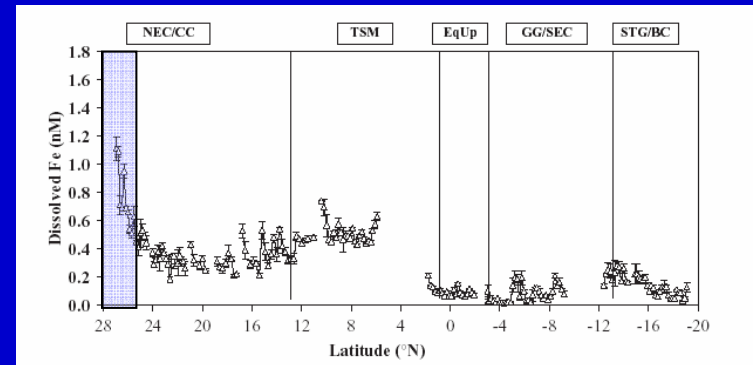
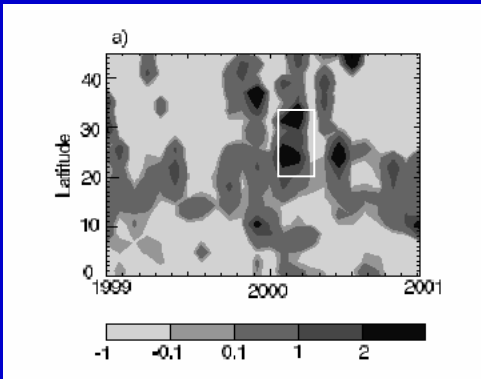
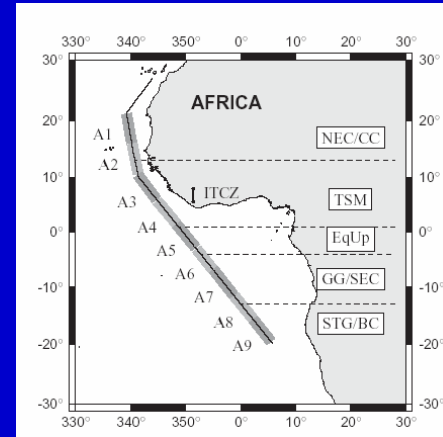
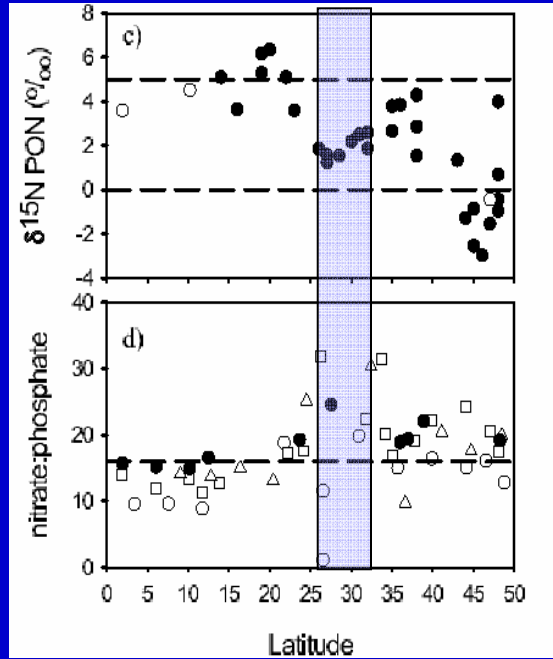
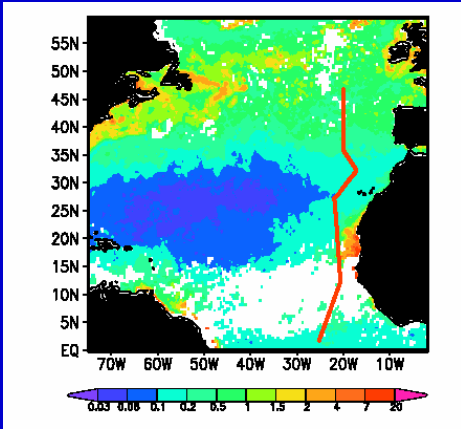
Annual report 2004



At DYFAMED site
(Med. Sea between Nice and corsica)



Natural iron fertilization from above and N₂ fixation



Mars 2000

Mahaffey et al. 2003

Septembre 2000

Sarthou et al. 2003

What are the characteristics of iron supply
to the ocean by natural processes?

How do they compare with those of purposeful additions?

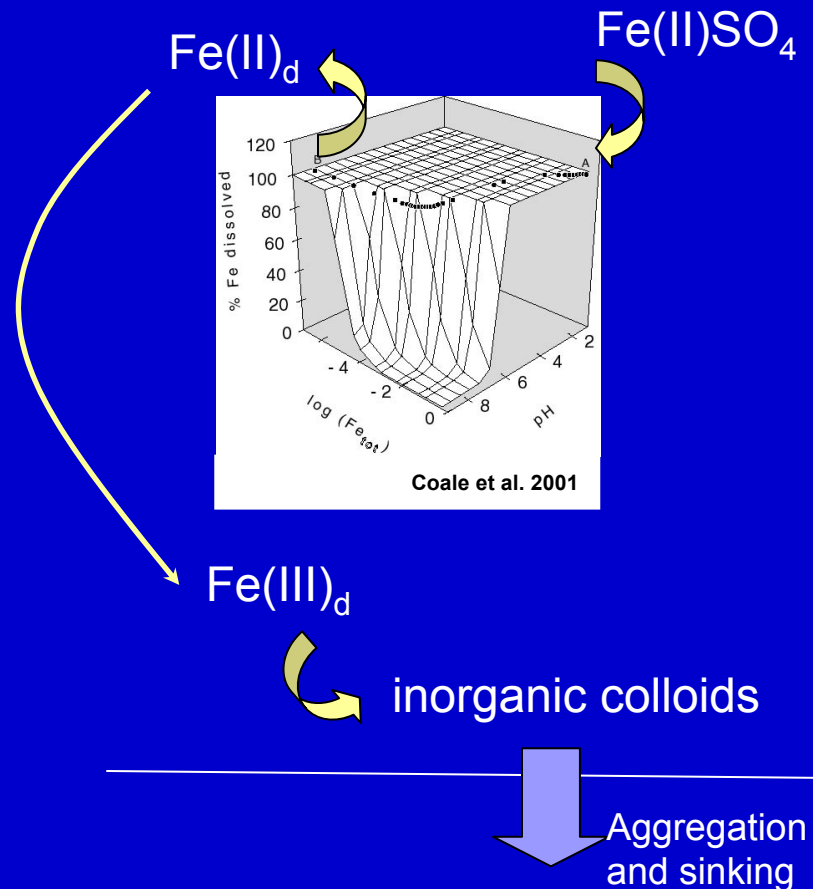
What are the natural chemical forms of iron coming from below ?

Natural speciation (KEOPS)

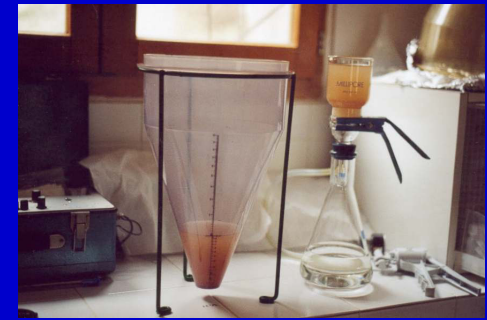
Dissolved phase dominated by organic complexation with an excess of ligands (Gueringa et al.)

Dissolution of lithogenic particulate Fe is a possible additional source.

Speciation after infusion



What are the natural chemical forms of iron coming from above ?



Mineral class	mineral	% in aerosol	% DFe	
Clay	illite	45	67.4	
	kaolinite	7	2.3	
	smectite	nontronite		
	beidellite	15	27	
feldspar	montmorillonite			
	oligoclase	6	2.8	
	orthoclase			
Iron (hydr-)oxide	hematite			
	goethite	4	0.5	
	magnetite			

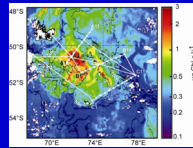


Mode of addition

Flux of DFe
 $\mu\text{mol m}^{-2} \text{d}^{-1}$

Flux of TFe
 $\mu\text{mol m}^{-2} \text{d}^{-1}$

more or less continuous

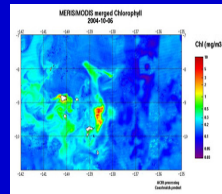


$< 10^{-1}$

1 ?

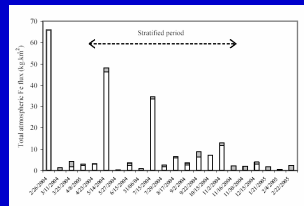
Natural
 From below

pulsed

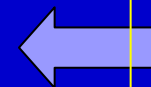


Natural
 From above

pulsed



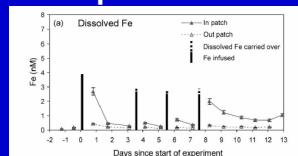
1- 10



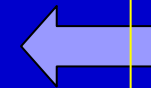
$\sim 10^3 - 10^2$

Acidic FeSO_4

pulsed

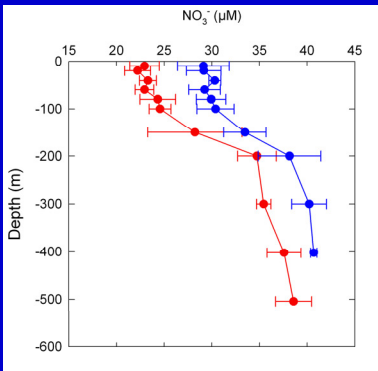
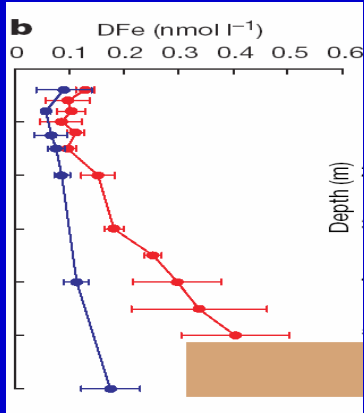


$10^2 - 10^3$

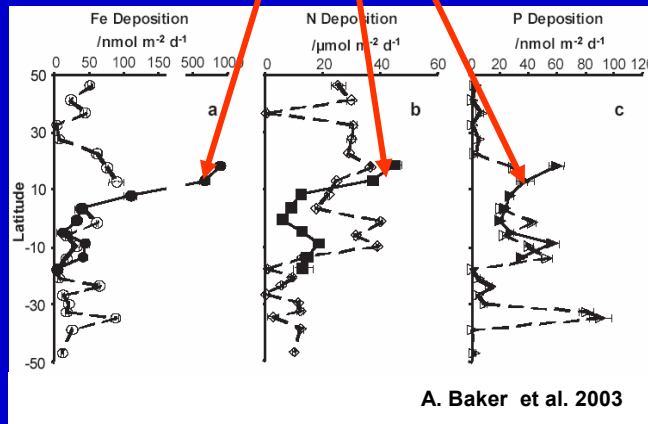


$\sim 10^4$

Iron is never supplied alone, macronutrients are also added.

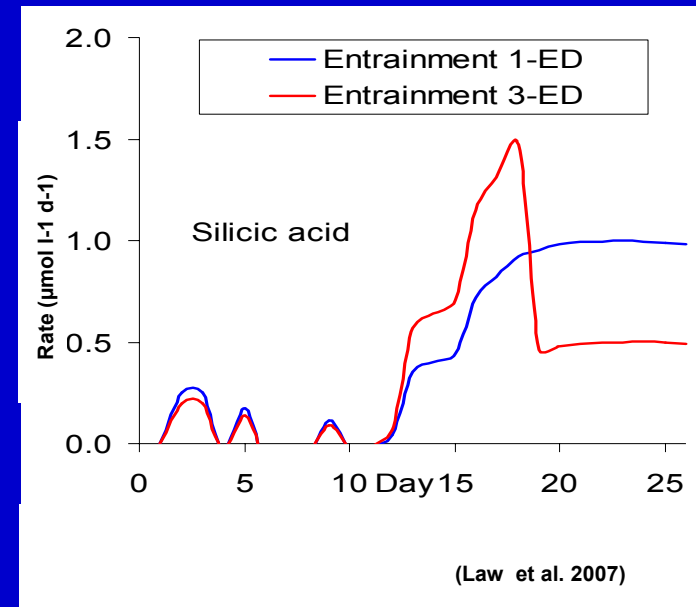


From below



From above

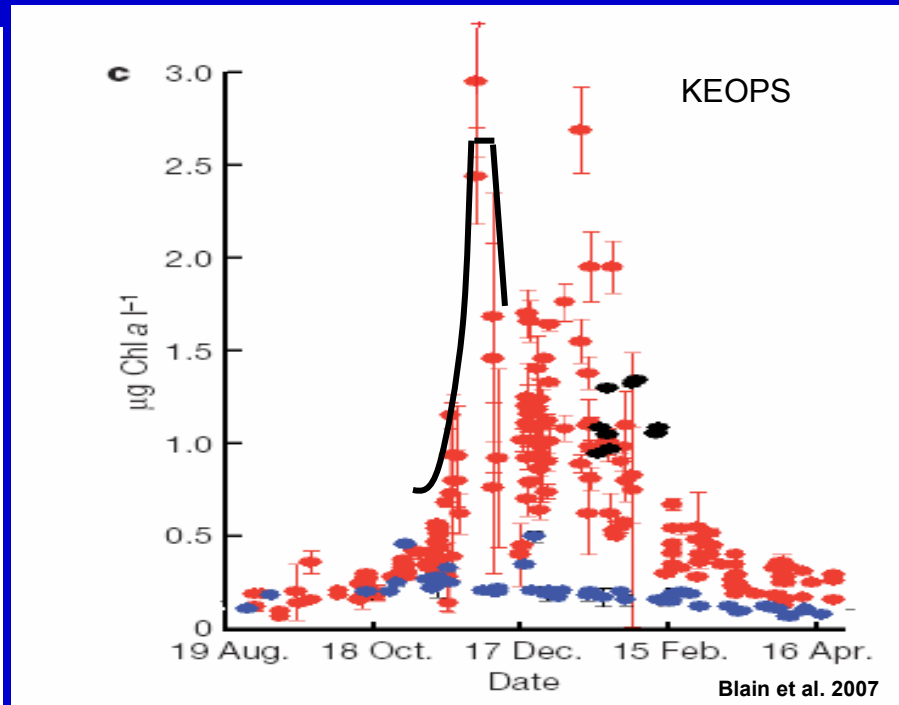
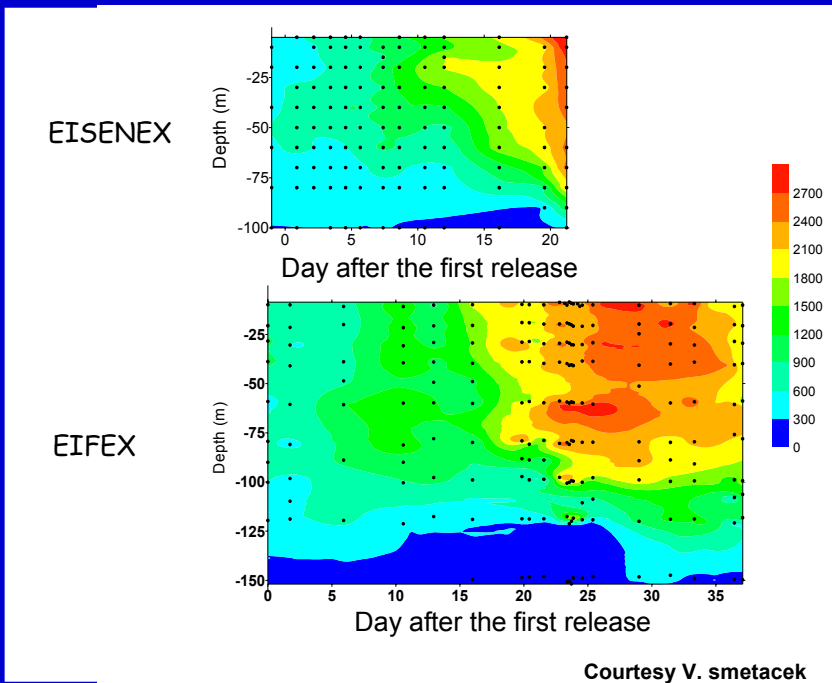
From deliberate release
Macronutrient entrainment
during SERIES



FeSO₄ addition is a poor imitation of the natural processes of fertilization,

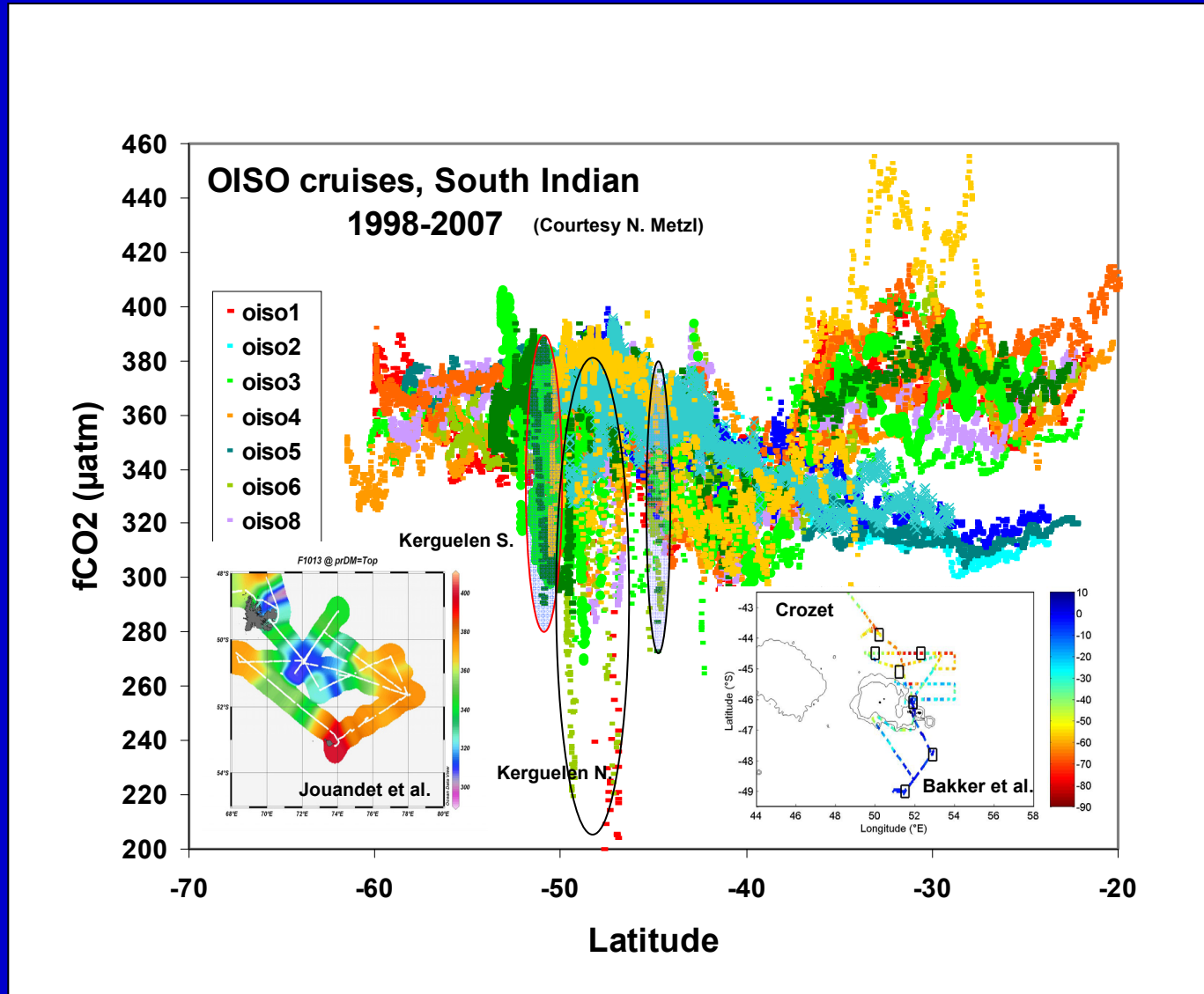
but is it an important issue ?

Yes it is



What can we learn from natural iron fertilization
on carbon sequestration in the ocean ?

CO₂ sink



Carbon export

Experiment	duration	excess of C export		references
		at 100m	at 200m	
(mmol m ⁻² d ⁻¹)				
SOIREE	13	0	0	Charette and Buesseler, 2000
EisenEx	22	0	0	Rutgers van der Loeff and Vöge 2001
SOFEX-S	27	7 ± 3	-	Buesseler et al 2005
KEOPS	~ 90	11 ± 5	14 ± 8	Savoye et al. (2007)
CROZEX	~ 90	9.4 ± 1.5		Morris et al. (2007)
EIFEX	36	13 ± 13		Savoye (unpublished)

$$\text{Efficiency} = \frac{\text{Excess of carbon export (mol)}}{\text{Excess of DFe supply (mol)}} \left\{ \begin{array}{l} 70,000 \pm 40,000 \\ 668,000 \end{array} \right. \quad (\text{blain et al. 2007})$$

Compared to 4,300 for SO exp. (de Baar et al. 2006)

Deep sequestration of C

C and Fe cycles coupling

KEOPS :Fraction of the int. PP exported at 100 m was twice **lower** in the bloom than in HNLC waters

(Savoie et al. 2007)

KEOPS :**No preferential remineralization of C and Fe** in the mixed layer and also in the seasonal thermocline of the Kerguelen bloom

(Obernosterer et al. 2007, Sarthou et al. 2007)

but

KEOPS :the fraction of the Cexp that was transferred below 450 m was **higher** below the bloom than in HNLC waters

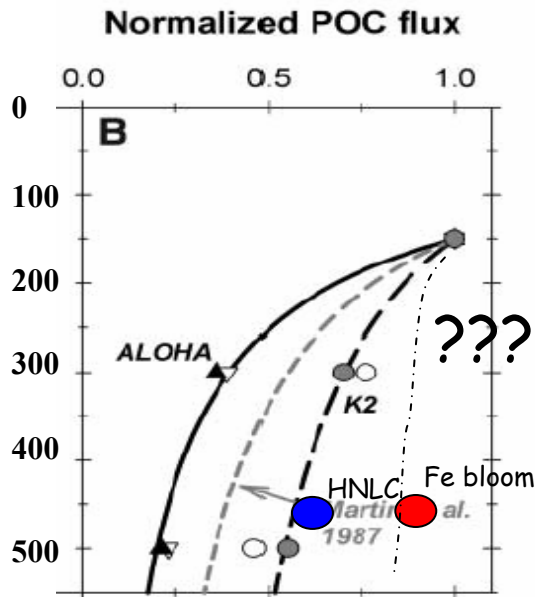
(Jacquet et al. 2007)

but

FECYCLE

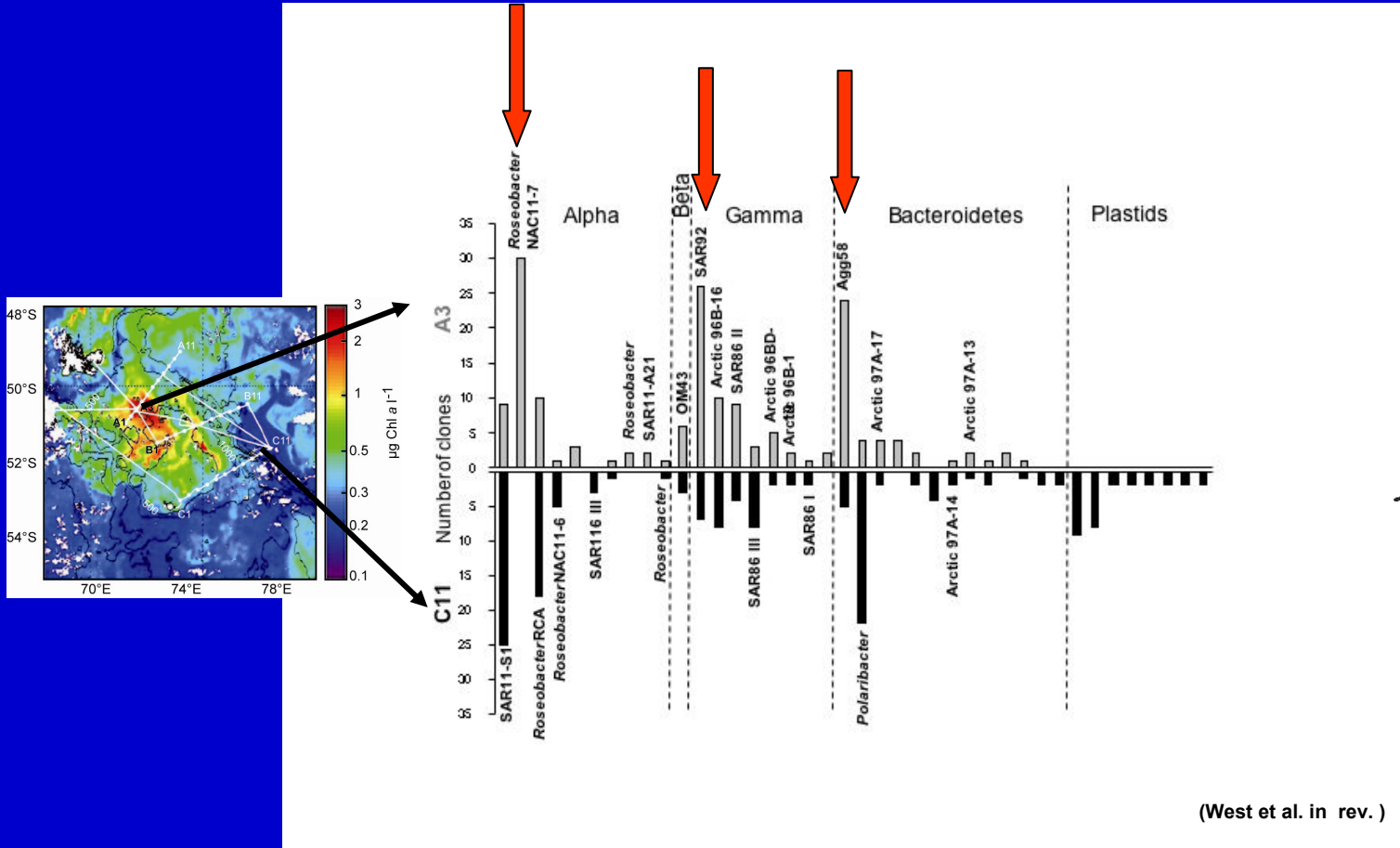
During the unperturbated experiment in the subantartctic, **Preferential remineralisation of C versus Fe** was observed below the MLD

(Boyd et al. 2005, Frew et al. 2005)



(Buesseler et al. 2007)

Composition of the heterotrophic bacterial community



Bacteria play a crucial role in coupling/decoupling C and Fe cycles in fertilized systems

➤ Natural iron fertilization does not only increase the rates of the biogeochemical processes but it drives a complete change of the ecosystem.



New parameterisations, new laws ?

➤ There is large variety of natural iron fertilized sites in the ocean with a large potential for new findings.

➤ The short term mesoscale iron fertilization experiment was a powerful tool for research in oceanography. The second generation of experiments should be more "subtle" to make their results more comparable with natural processes.

➤ The mode of iron addition, planned in large scale / commercial iron fertilization, does not imitate a natural process of fertilization.