# Terrestrial Organic Carbon Inputs to Marine Sediments

### **Reading List**

•Prahl F.G., Ertel J.R., Goni M.A., Sparrow M.A. and Eversmeyer B. (1994) Terrestrial organic contributions to sediments on the Washington margin.*Geochim. Cosmochim. Acta* **58**, 3035-3048.

• Goni M.A., Ruttenberg K.C. and Eglinton T.I. (1997) Sources and contribution of terrigenous organic carbon to surface sediments in the Gulf of Mexico. *Nature*, **389**, 275-278.

• Hedges J.I., Keil R.G. and Benner R. (1997) What happens to terrestrial organic matter in the ocean? *Org. Geochem.* **27**, 195-212.

• Hedges J.I. and Oades J.M. (1997) Comparative organic geochemistries of soils and marine sediments. *Org. Geochem.* **27**, 319-361.

• Leithold & Blair (2001) Watershed control on the carbon loading of marine sedimentary particles. GCA 65, 2231-2240.

• Goni et al. (2005) The supply and preservation of ancient and modern components of organic carbon in the Canadian Beaufort Shelf of the Arctic Ocean. Mar. Chem. 93, 53-73.

• Schefuss et al. 2003 Carbon isotope analyses of *n*-alkanes in dust from the lower atmosphere over the central eastern Atlantic. GCA 67, 1757-1767.











# Important Considerations

- Organic compounds synthesized by organisms are subject to biological and physicochemical processes that alter their chemical composition, and complicate their recognition and quantification in downstream organic carbon (OC) reservoirs such as soils and sediments.
- This "pre-conditioning" that modifies organic matter prior to burial may influence its reactivity in the sub-surface (e.g., by physical association or chemical reaction).
- The time-scales over which organic matter is processed prior to burial may also vary substantially, depending on its origin.
- As a result, contemporaneously deposited organic material of terrestrial and marine origin may exhibit a range of ages and labilities.
- In seeking to quantify the proportions of organic matter preserved in the sub-surface that stem from different sources it is important to find tracer properties that are largely independent of degradation.
- · Continental margins contain significant quantities of "pre-aged" organic carbon.

# Approaches to quantify OC inputs to marine sediments Bulk parameters C<sub>organic</sub>/N<sub>total</sub> δ<sup>13</sup>C<sub>TCC</sub> Molecular parameters Regression of terrestrial biomarker concentrations vs bulk properties (δ<sup>13</sup>C, C<sub>org</sub>/N) extrapolation to zero marker concentration yields a bulk marine end-member elemental or isotopic value that can be inserted into isotopic/elemental mass balance. Direct use of concentration measurements for biomarkers in "representative" end-member samples (e.g. plant wax biomarkers in riverine suspended sediments) to determine extent of dilution by marine OC. Directuse of condentration seasurements for biomarkers in "representative" end-member samples (e.g. plant wax biomarkers in riverine suspended sediments) to determine extent of dilution by marine OC. Directuse of condentration seasure considered (marine and vascular plant), and terrestrial end-member biased towards vascular plant inputs. Constancy in composition is assumed along transects.

## Bulk properties used to quantify terrestrial OC inputs

### C<sub>org</sub>./N ratios

- Principle: Vascular plant biomass is depleted in nitrogen (mainly comprised of cellulose and lignin), compared to [protein-rich] marine phytoplankton.
- Limitations:
- Diagenetic influences proteins are relatively labile, resulting in increased  $C_{\rm org}/N$  ratios with degradation.
- Inorganic N bound in clays can affect ratio, especially in low TOC sediments.

### $\delta^{13}$ C OC composition

- Principle: OC from marine primary production typically enriched in <sup>13</sup>C relative to C<sub>3</sub> vascular plant carbon.
- Limitations:
- Complications due to mixed inputs of C<sub>3</sub> and C<sub>4</sub> higher plant carbon.
- Past and present-day variations in  $\delta^{13}$ C value of marine end-member.
  - Potential diagenetic influences due to intermolecular isotopic variations (e.g. selective preservation of <sup>13</sup>C-depleted lipids over <sup>13</sup>C-enriched proteins)



Fig. 5. Atomic N/C ratio versus stable carbon isotopic composition ( $\delta^{13}$ C,  $\infty$ ) of organic matter in size and density fractions isolated from Washington coast sediments (Keil *et al.*, 1994). M is marine material, B is bulk sediment, C is clay-, L is silt- and S is sand-sized sediment. Subscript t is high density ( $\rho > 2.6$ ) fraction and subscript o is low density ( $\rho < 1.5$ ) fraction.

































# Evidence for minimal terrestrial OC contributions to marine sediments

- Variations in OC:SA and  $\delta^{13}C_{\text{OC}}$  in estuaries.
- Low C<sub>org</sub>/N values for marine sediments.
- Enriched  $\delta^{13}\text{C}$  values of marine sedimentary OC relative to terrestrial (C\_3 OC).
- Rapid decrease in lignin phenols with distance offshore.

# Evidence for significant terrestrial OC contributions to marine sediments

- Unknown contributions from <sup>13</sup>C-enriched (C<sub>4</sub>) terrestrial OC sources.
- Importance of hydrodynamic processes in exporting terrestrial OC.
- · Old core-top ages for continental margin sediments.
- Global influence of small, mountainous rivers.
- Arctic ocean undersampled.
- Widespread distribution of plant wax lipids in ocean sediments.
- Greater importance of terrestrial OC in glacial times (low sea-level stand)?



























Fractions	Concn. (gdw basis)	δ <sup>13</sup> C (‰)	∆ <sup>14</sup> C (‰)	<sup>14</sup> C age (yr BP)
Total Organic Carbon	1.02 %	-18.93	-149.6	1260 ± 40
Black Carbon	0.24 %	-15.13	-231.7	2070 ± 35
Plant wax alcohols	12 μg	-27.9	-80.8	649 ± 143
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