

## Lecture 13

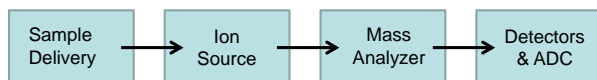
### Measurements I: Mass spectrometry

- Introduction
  - Uses, criteria, commonalities
- Ion Sources
  - Electron impact, TI, ICP, SI
- Analyzers
  - E & M sectors, QMS, TOF, Ion Traps
- Detectors
  - Faraday cups & electron multipliers
- Data Acquisition
  - Electrometers, ADC, statistics

### Uses for Mass Spectrometers

- Trace element determination
  - sensitivity & isolation
- Quantitative analysis
  - reproducibility & separation, isotope dilution
- Molecular structure determination
  - deconstruction of molecules
- Material identification and separation
- **Isotope abundance determination**
- Isotope mass determination

### Commonalities



- Sample delivery system
  - Pre-purification, optimal phase, preferably non-fractionating
- An ion source
  - Atoms must be charged to be manipulated
  - Preferably mono-energetic, but sometimes spread
  - Creates, accelerates, focuses, collimates a beam
- A “mass” analyzer (m/q)
  - Selects momentum, also maybe energy
  - Refocus & maybe collimates beam
- A detector(s) and Analog to Digital Conversion
  - Measure and discriminate beam(s), possibly isobars
  - Digitize for data acquisition.

### In addition

- Sample preparation and introduction system
  - Often integrated into machine
  - Depends on sample type and machine type
- Data gathering and processing system
  - Digitization of analog signals
  - Counting of ions
- Instrument control
  - Control of all parts
  - Automation and book-keeping
  - More reliable, cost effective, humane

## Decisions to make

- Sample character: gas, liquid or solid, purity?
- How abundant is the material?
- Isotopic abundance: similar or drastically different?
- Range: do isotope ratios vary a lot or a little?
- Are there isobaric (molecular or nuclear) interferences?
- Are there “matrix effects”?
- Do we need to know the “absolute” ratio?

## Figures of Merit

- Resolution: smallest difference in mass
- Sensitivity: how small a signal can be discerned?
- Blank: what do you get with no sample at all?
- Abundance sensitivity: the smallest isotopic ratio
- Stability and reproducibility
- Speed and throughput
- Discrimination and accuracy
- How much does a measurement depend on the presence or absence of other species?
- Memory effect? Does the system hold a grudge?

## Resolution

- Are there interfering isobars?
  - How small a mass difference can you distinguish?
- Caveat emptor: watch the definitions!
- You can separate entangled peaks if you know the peak shape very well
  - Not trivial to do...

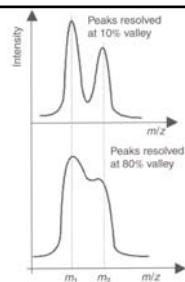
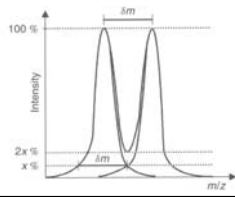


Figure 2.1  
Diagram showing the concepts of peak resolution and valley



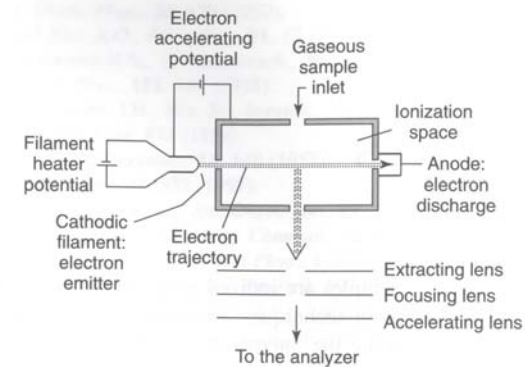
## Measurements I: Mass spectrometry

- Introduction
  - Uses, criteria, commonalities
- **Ion Sources**
  - **Electron impact, TI, ICP, SI**
- Analyzers
  - E & M sectors, QMS, TOF, Ion Traps
- Detectors
  - Faraday cups & electron multipliers
- Data Acquisition
  - Electrometers, ADC, statistics

## Types of Ion Sources

- Electron impact
  - Gases
- Microwave Plasma
  - Also gases
- Thermal ionization
  - solids
- Inductively coupled plasma source
  - liquids
- SIMS and Laser Ablation
  - solid surface analysis
- There are others
  - But not important for isotope work

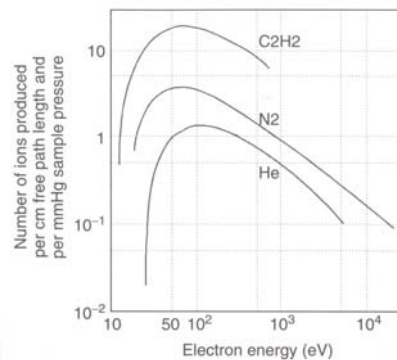
## Electron Impact Source



- Generally has more plates than this

## Electron impact ionization

- Depends on atomic structure
- Can produce multiply charged ions
- Sensitivity generally around  $n \times 10^{-4}$  amps/torr

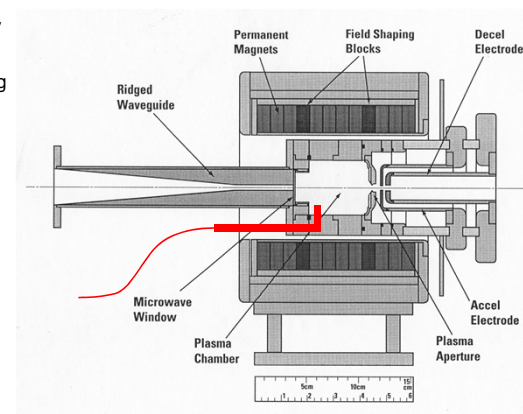


$$1 \text{ amp} = 6.24 \times 10^{18} \text{ e s}^{-1}$$

$$1 \text{ torr} = 1/760 \text{ atmospheres}$$

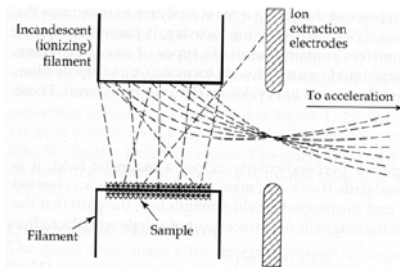
## Microwave Plasma Gas Ion Source

Plasma maintained by microwave discharge,  
Supported by inflowing gas through capillary  
Contained by axial magnetic fields  
Ions extracted and focused by electric fields



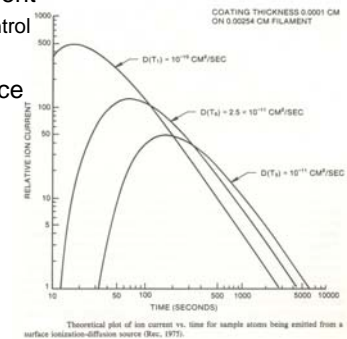
## Thermal Ionization Source

- Solid sample dissolved and concentrated into a few drops
- Deposited on a refractory filament
  - Often with 2ndary coatings
- Filament heated in an ion source
  - Diffusion of material to surface
  - Volatilization
  - Ionization
  - Usually multiple filaments



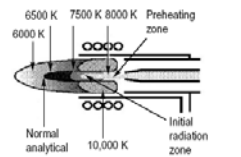
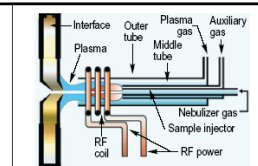
## Thermal Ionization Source

- Solid sample dissolved and concentrated into a few drops
- Deposited on a refractory filament
  - Often with 2ndary coatings to control release characteristics
- Filament heated in an ion source
  - Diffusion of material to surface
  - Volatilization
  - Ionization
  - Usually multiple filaments



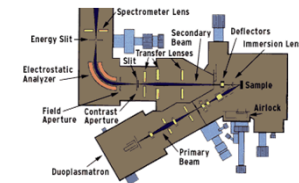
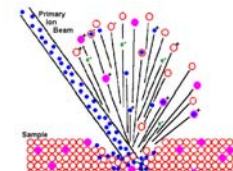
## Inductively Coupled Plasma Source

- For “direct” injection of solutions
  - Amenable to automation
  - Simpler chemistry (?)
- Ionization ~ 100%
- Process:
  - Ar from liquid Ar source
  - Torch made of quartz (high temp!)
  - electrostatic ignition of Ar
  - Maintenance of plasma by RF (1-2 KW)
  - Nebulization of solution into Ar flow
  - Coaxial flows of Ar (for cooling too)
  - Ion extraction by “skimmers” (cooled)
  - Skimmers interface vacuum to Ar pressure
- Issues: matrix effects
- Recent developments with laser ablation



## Secondary Ion Mass Spectrometry

- For surface analysis of solids
  - E.g., thin sections
- Spatial resolution ~  $10^{-6}$ m
  - Laterally and in depth
- Process
  - Create ion beam (Ar, Cs, etc)
  - Extract & ionize sputtered atoms typically < 10% ionized
  - Accelerate, collimate & focus
  - Big energy spread in ions
    - Need to energy filter
- Issues: surface & ionization biases
  - Standard surfaces/matrices?



## Measurements I: Mass spectrometry

- Introduction
  - Uses, criteria, commonalities
- Ion Sources
  - Electron impact, TI, ICP, SI
- **Analyzers**
  - **E & M sectors, QMS, TOF, Ion Traps**
- Detectors
  - Faraday cups & electron multipliers
- Data Acquisition
  - Electrometers, ADC, statistics

## Mass Analyzers

- Magnetic and Electric Sectors
  - Most common for isotope studies
- Time of Flight
  - Simple designs
- Quadrupolar Mass Analyzers
  - Robust rapid scanners
- Ion Traps
  - Compact and emerging tools

## Magnetic and Electric Sectors

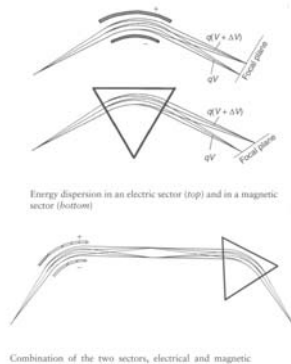
- Electric sectors:
  - Energy filters
  - Balance between:
    - Centripetal force  $mv^2/R$
    - Electrical force  $qV$

$$qV_{Analyzer} = \frac{mv^2}{R}$$

or

$$R = \frac{2E_K}{qV_{Analyzer}}$$

- Resolution related to R and defining slit dimensions



## Magnetic and Electric Sectors

- Magnetic sectors:
  - Momentum filters
  - Balance between:

$$F_c = \frac{mv^2}{R} = qvB$$

or

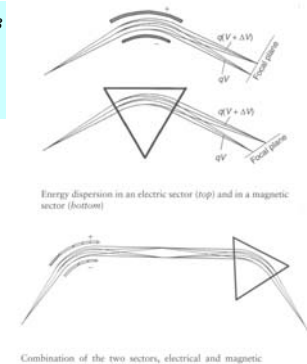
$$R = \frac{mv}{qB}$$

- Centripetal force  $mv^2/R$
- Magnetic force  $qvB$
- Since energy =  $qV$ 
  - We have  $E = \frac{1}{2}mv^2 = qV$

$$R^2 = \frac{2mV}{qB^2}$$

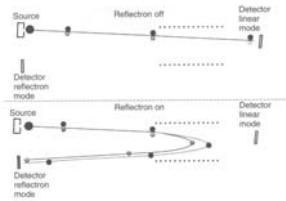
- Resolution related to R and defining slit dimensions

$$\frac{\Delta M}{M} \cong \frac{\Delta R_{Beam} + \Delta R_{Slit}}{R}$$



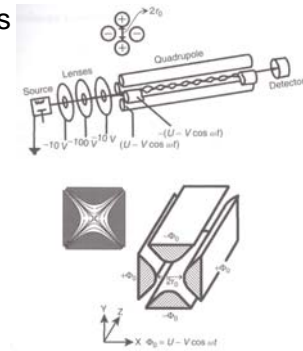
## Time of Flight Instruments

- Conceptually simple
- Not widely used for isotope studies
- Operated in pulse mode
  - “duty cycle” low
- Momentum filters:
  - $\Delta T = L / v$



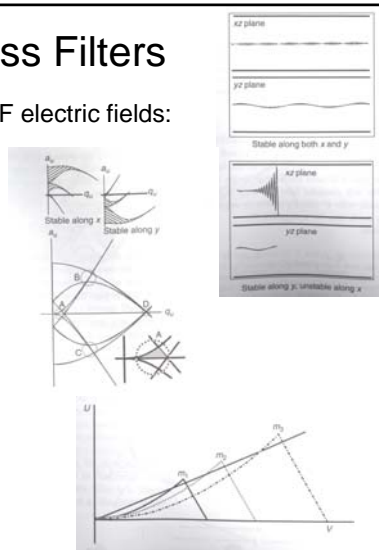
## Quadrupolar Mass Filters

- Reliant on orbital stability in RF electric fields:
  - Only stable orbits passed
- Complicated optical solutions
  - Integrated Mathieu equations
- Compact and robust
- Low voltage
- Fast scanners
- Wide range
- Constant resolution vs Mass



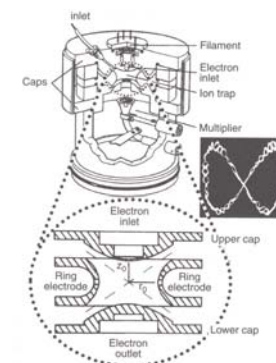
## Quadrupolar Mass Filters

- Reliant on orbital stability in RF electric fields:
  - Only stable orbits passed
- Complicated optical solutions
  - Integrated Mathieu equations
- Compact and robust
- Continuous mode operation
- Low voltage
- Fast scanners
- Wide range
- Constant resolution vs Mass



## Ion Traps

- Invented by the same guy (Paul)
- Opposite to QMS
  - All ions in stable orbits
  - Desired ions ejected
  - Operated in pulse mode
- Compact & easily made
- Emerging technology



## Measurements I: Mass spectrometry

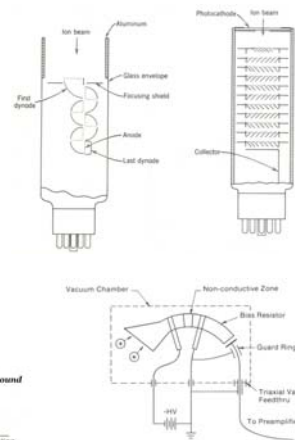
- Introduction
  - Uses, criteria, commonalities
- Ion Sources
  - Electron impact, TI, ICP, SI
- Analyzers
  - E & M sectors, QMS, TOF, Ion Traps
- **Detectors**
  - **Faraday cups & electron multipliers**
- Data Acquisition
  - Electrometers, ADC, statistics

## Faraday Cups

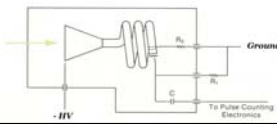
- For intense ion beams
  - From  $10^{-15}$  –  $10^{-9}$  ampere range
- Very simple: collect charge
  - Bleed through electrometer
  - Easily created arrays of cups
- Watch for secondary electrons
  - From beam in cup
  - From beam outside of cup
  - Secondary electron suppression plates

## Electron Multipliers

- For ion beams  $< 10^{-12}$  amps ( $< 10^7$  ions /s)
- Relies on secondary electron cascade
- Discrete or continuous
- Current or pulse mode
- Run at  $10^2 > \text{Gain} < 10^8$
- Ion feedback issues



Gain depends on high voltage applied to device

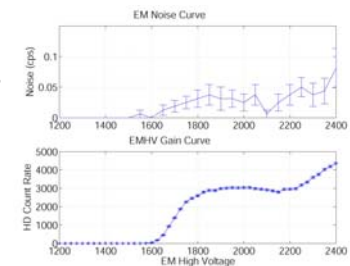


## Electron Multipliers

- For ion beams  $< 10^{-12}$  amps ( $< 10^7$  ions /s)
- Relies on secondary electron cascade
- Discrete or continuous
- Current or pulse mode
- Run at  $10^2 > \text{Gain} < 10^8$
- Ion feedback issues

Current vs. pulse counting modes:

- (a) gain dependency!
- (b) dark current discrimination



Gain and NOISE depends on high voltage applied to device

## Measurements I: Mass spectrometry

- Introduction
  - Uses, criteria, commonalities
- Ion Sources
  - Electron impact, TI, ICP, SI
- Analyzers
  - E & M sectors, QMS, TOF, Ion Traps
- Detectors
  - Faraday cups & electron multipliers
- **Data Acquisition**
  - **Electrometers, ADC, statistics**

## Measurement of ion currents

- From Faraday Cup
  - Electrometers (usually FET input)
    - Use feedback resistors
      - Inherent noise
      - Intrinsic time constant
- From S.E.M.
  - If “current mode”, use with electrometers
    - First order gain dependence on SEM history
  - If “pulse mode”, use preamplifier/discriminator and fast counting electronics

## Measurement of ion currents

- Pulse mode SEM already digital (ions/sec)
- Electrometer outputs need to be converted to digital
  - Either ADC (16-24 bit DACs now available)
  - Or VFC (longer term integrators, highly linear, good for ratiometric measurements)