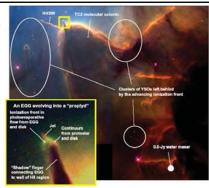
Earth Structure and Evolution



Bulk earth composition, radioactive heating, partitioning/fractionation, natural reactors

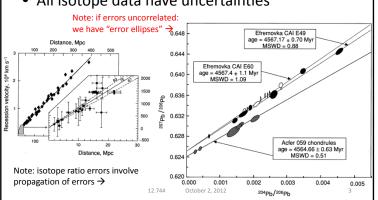
Earth Structure and Evolution

- Fitting Isotope/Isochron Data
- Bulk Earth Composition and Differentiation
- Radioactive Heating
- Natural Reactors

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A word about dealing with "real data" for isotope-isotope relations

All isotope data have uncertainties



A word about dealing with "real data" for isotope-isotope relations

- All isotope data have uncertainties
- Finding "best fit" (straight) lines must account for relative uncertainties/weights → and provide uncertainties in slope/intercept
- Normal "least squares" methods account only for uncertainties (weights) in the "y-coordinate"
- When there are uncertainties in both x- and ycoordinates, you need to use a "Type 2 Regression"; a non-trivial exercise

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Type 2 Regressions

- Involve "weighted distance normal to trend line"
 - Therefore are complicated & iterative in nature
 - Especially problematic when strongly correlated errors and large dynamic range
 - IsoPlot (a BSD openware app for Excel)
 - Problem: IsoPlot 3 only compatible with Excel 2003 & earlier (need IsoPlot 4 for 2007+ mutually incompatible & performance compromises)
 - See 12.747, Lecture 3 & MATLAB m-files, plus chapter 3 of Glover, D.
 M., W. J. Jenkins, et al. (2011). <u>Modeling methods for marine science.</u>
 <u>Cambridge, U.K., Cambridge University Press.</u>
- Some references (posted on web site):
 - York, D., N. M. Evensen, et al. (2004). "Unified equations for the slope, intercept and standard errors of the best straight line." <u>American Journal of Physics</u> 72(3): 367-375.
 - Reed, B. C. (1992). "Linear least-squares fits with errors in both coodinates. II: Comments on parameter variances." <u>American Journal of Physics</u> 60(1): 59-62.
 - Reed, B. C. (2010). "A spreadsheet for linear least-squares fitting with errors in both coordinates." Physics Education 45(1): 93-96.

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A note on terminology

• Traditionally isotope information presented in δ notation, referred to a generally accepted/available standard:

$$\delta_X = \left(\frac{R_{\text{Unknown}}}{R_{\text{Standard}}} - 1\right) \times 1,000 \%$$

• Another notation used for some systems:

$$\varepsilon_X = \left(\frac{R_{\text{Unknown}}}{R_{\text{Standard}}} - 1\right) \times 10,000$$

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Bulk Earth Composition

- We can only sample the top ~1% of the earth's material
 - Other inferences made from seismic, isotopic, mineralogical, and heat-flow information
- Speculate that earth may have formed from material similar to undifferentiated meteorites
 - Which meteorites?
 - C1 (carbonaceous chondrites) main candidate
 - Enstatite condrites

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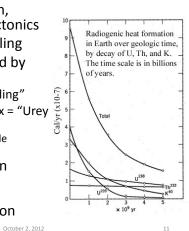
The terrestrial heat flux

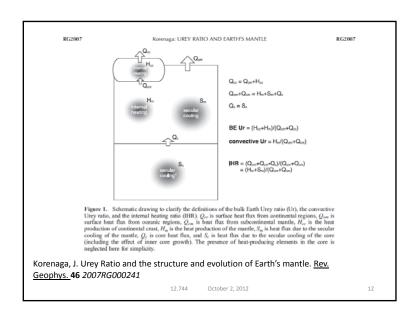
- Lord Kelvin (W. Thompson) in 1863 estimated age of earth to be 20-400 Ma based on observed conductive heat flux and secular conductive cooling
 - Before radioactivity was discovered
- Early in solar system/planetesimal history gravitational energy release,
 - ²⁶Al, ³⁶Cl, ⁶⁰Fe, and ²⁴⁴Pu were probably important
 - earth likely molten
- Later, it's just down to ⁴⁰K, ²³²Th, ²³⁵U, ²³⁸U
 - All decreasing with time at their own rates

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Radioactive heat generation

- Drives mantle convection, mantle plumes, plate tectonics
- Serves to slow earth cooling
- ~Half of 44 TW generated by radioactivity
 - Other half is "secular cooling"
 - Ratio radiogenic/total flux = "Urey Ratio"
 - Usage may/may not include continents
- Modeled heat production depends on bulk earth composition and differentiation/distribution

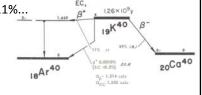




Lest you think neutrinos are just a curiosity...

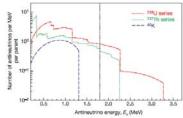
Geoneutrinos and the K/U ratio of the earth

- Present day geothermal heat (~44 TW) partly produced by U,Th series and ⁴⁰K decay
- What is terrestrial K/U ratio?
 - Based on chemical/physical formation models
 - Implications regarding chemical, physical, and thermal history of earth
 - different half-lives of decay
 - Chemical differentiation
- ⁴⁰K produces 89% ... ,11%...
- U,Th chains produce...
 - But at different energies



Geoneutrinos

- Stellar fusion produces...
- Alpha-chain and K-decay produce...
- Use "inverse neutron decay reaction"



 $n \xrightarrow{11 \text{min}} p + e^- + \overline{\nu}_e$

 $\overline{V}_e + p \rightarrow n + e^+$

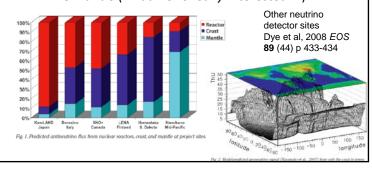
Figure 1 | The expected ²³⁸U, ²³²Th and ⁴⁰K decay chain electron antineutrino energy distributions. KamLAND can only detect electron antineutrinos to the right of the vertical dotted black line; hence it is insensitive to ⁴⁰K electron antineutrinos.

Araki et al, 2005 *Nature* **436** p 499-503

Where do geoneutrinos come from?

There are **other** sources of antineutrinos:

- Nuclear reactors (fission products)
- The crust (enriched in K, U, Th)
- The mantle (what we're really interested in)



Progress so far...

- Surface heat flow ~47 TW
- Radiogenic heat production ~24 TW
 - But ~5 TW escapes via antineutrinos
 - Thus a large portion is secular cooling

Dye, S.T. Geoneutrinos and the radioactive power of the Earth. <u>Rev. Geophys.</u> **50** 2012RG000400

Mareschal, J-C et al. Geoneutrinos and the energy budget of the Earth. <u>J. Geodyn.</u> **54** 43-54

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²³⁵U-fission

- ²³⁵U much more fissile than ²³⁸U
- Requires critical concentration of ²³⁵U
 - ~2Ga ago 235 U/ 238 U = ???
 - Ore formation via redox processes in groundwater
 - U⁺⁶ oxic is soluble
 - U⁺⁴ anoxic is insoluble
- Need water to slow neutrons down by inelastic collisions (probability of n-induced fission ~ 1/v)
- Further evidence?

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Natural Nuclear Reactors?

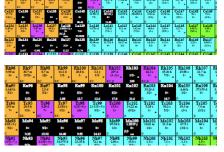
- Present global $^{235}U/^{238}U = 0.00720$
 - Extremely constant
 - small relative mass difference → less opportunity for geochemical/geological mass fractionation
 - French nuclear weapons lab noted small but significant ²³⁵U deficit in Oklo U-ore
 - $^{235}U/^{238}U = 0.00717 (\delta = -4\%)$
- Remember the following:
 - ²³⁵U is much more fissionable than ²³⁸U
 - Fission process is $n + {}^{235}U \rightarrow FP's + a$ few n
 - To have "critical mass" need ²³⁵U "density" high enough so that at least 1 of the "few n" yields another fission
 - ²³⁵U/²³⁸U is decreasing with time

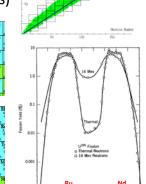
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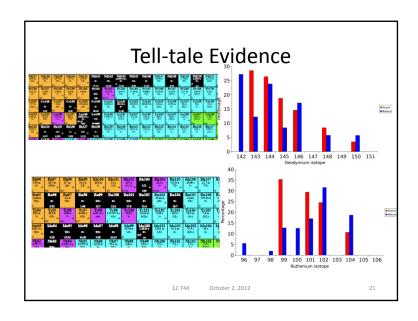
Tell-tale Evidence

Fission products! (2 examples)





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