

# Why is Carbon Preserved in Marine Sediments?

**Enhanced productivity** – More organic matter is produced which leads to a greater flux through the water column and ultimately greater accumulation in sediments. Sediments underlying high productivity areas have higher % OC and higher C accumulation rates:

## **Mechanisms:**

Selective preservation – some OM is intrinsically more labile than others

Nonselective preservation – a) mineral protection either inside of or on minerals. Enhanced flux due to ballasting.  
b) reactive OM moves through the high activity sediment/water interface quickly

**Low oxygen**- Less oxygen changes the degradation rate (?) or pathway (?) of C degradation. Mechanisms are not clear, but sediments in low oxygen and anoxic basins have high %OC.

## **Mechanisms:**

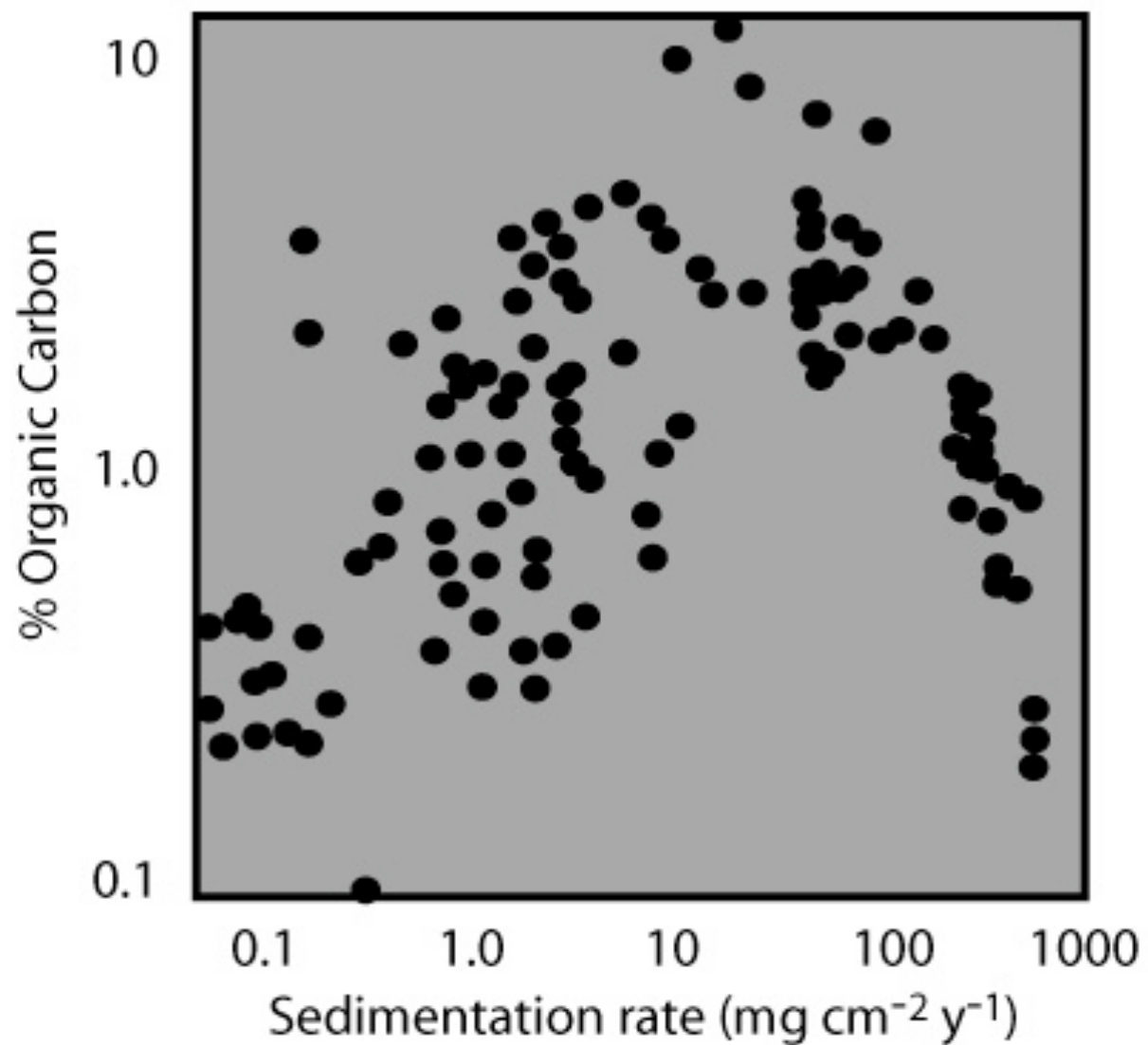
Anaerobic degradation inherently less efficient than aerobic degradation

Anaerobic degradation inherently slower than aerobic degradation

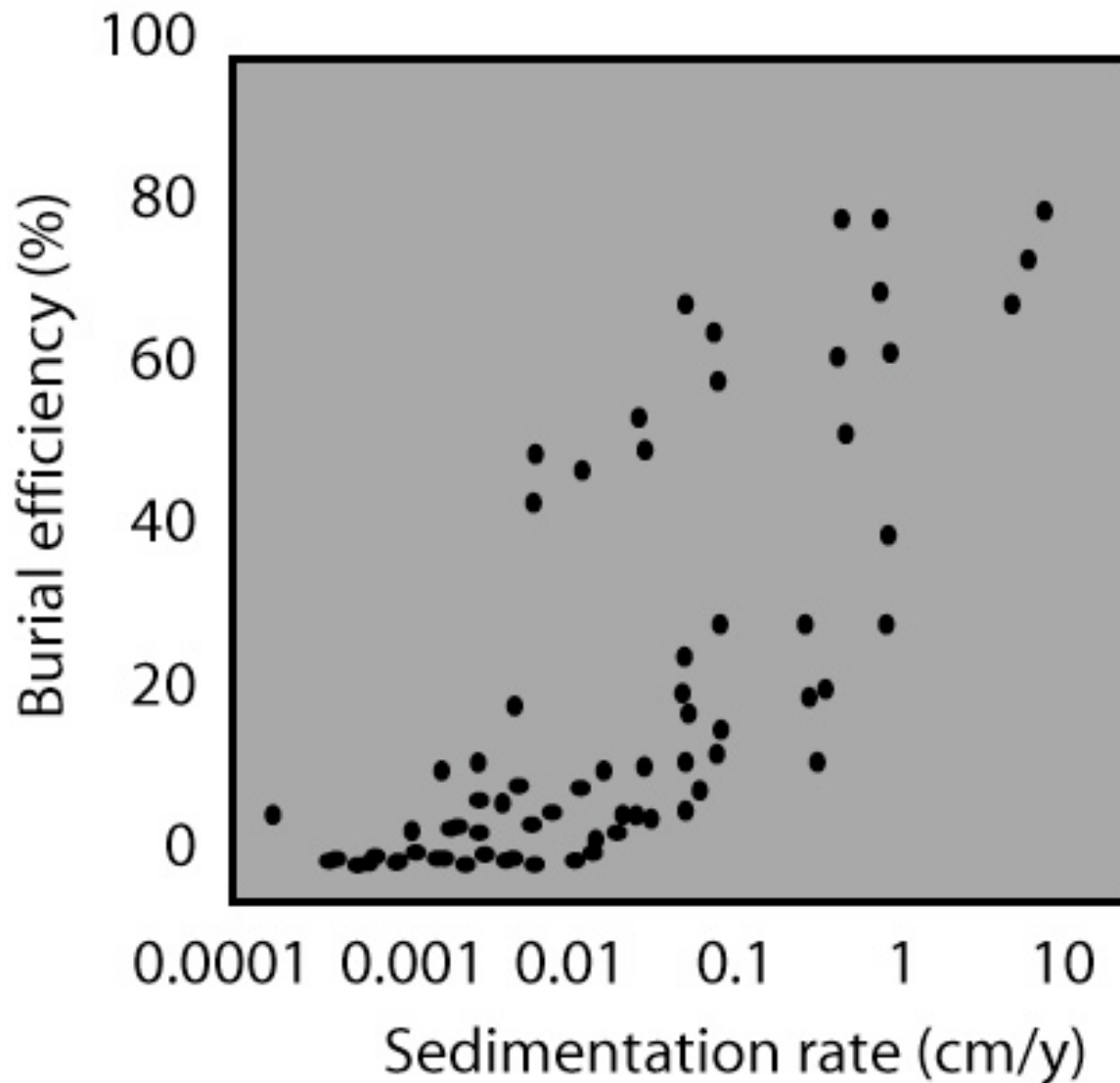
poisoning

a & b above

Relationship between % OC and sedimentation rate



## Relationship between burial efficiency and sedimentation rate

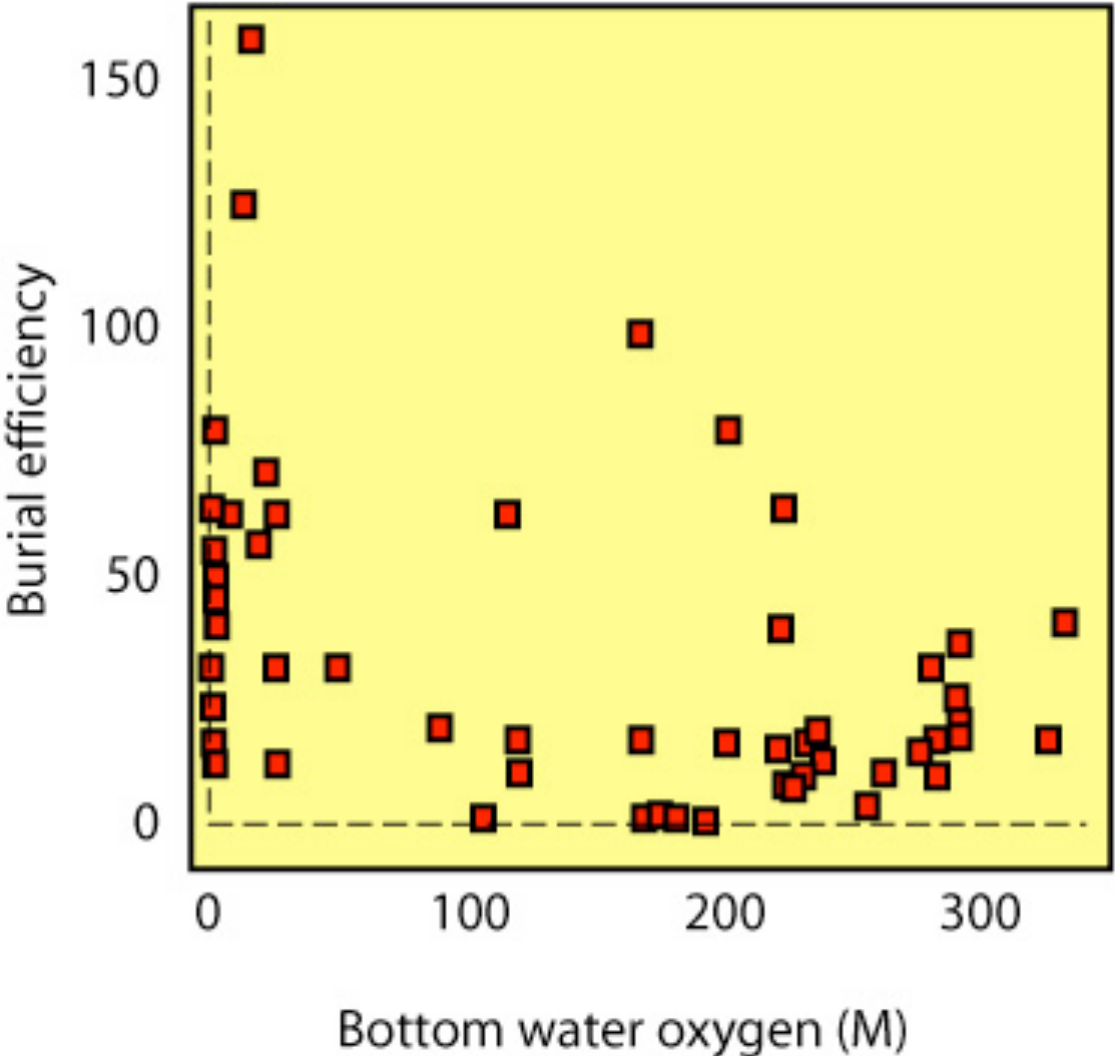


$$\frac{\text{Carbon buried}}{\text{Carbon flux}} = \text{BE}$$

Burial efficiency accounts for dilution by low carbon debris (carbonate, silica)

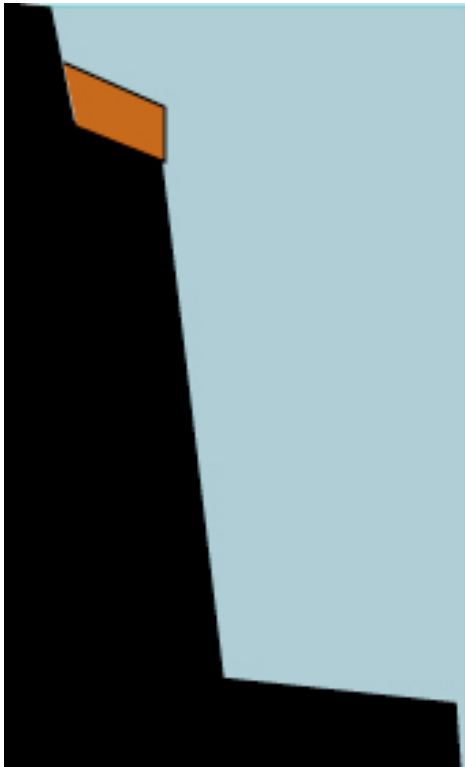
- 1) Rapid burial moves the C out of the zone of most intense remineralization.
- 2) Rapid burial “caps” the sediment column and inhibits exchange of  $\text{O}_2$ ,  $\text{NO}_3^-$ , etc.
- 3) Rapid burial often occurs at sites where there is a lot of recycled organic carbon.

# The effect of bottom water oxygen on burial efficiency

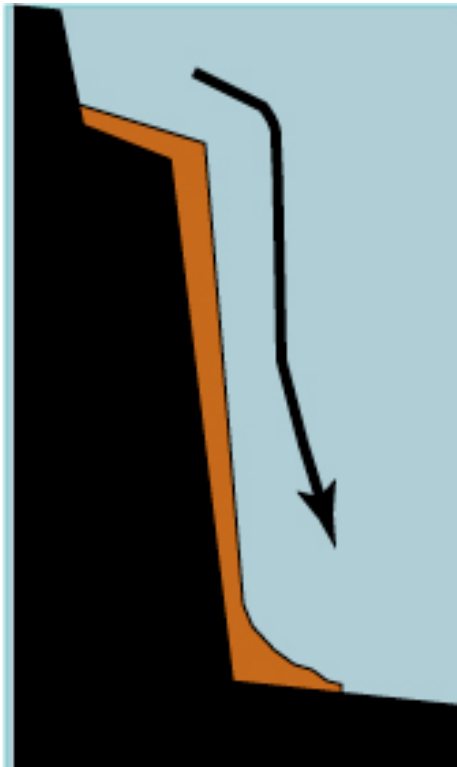


# The effect of oxygen on carbon preservation in Maderia Abyssal Plain Turbidites

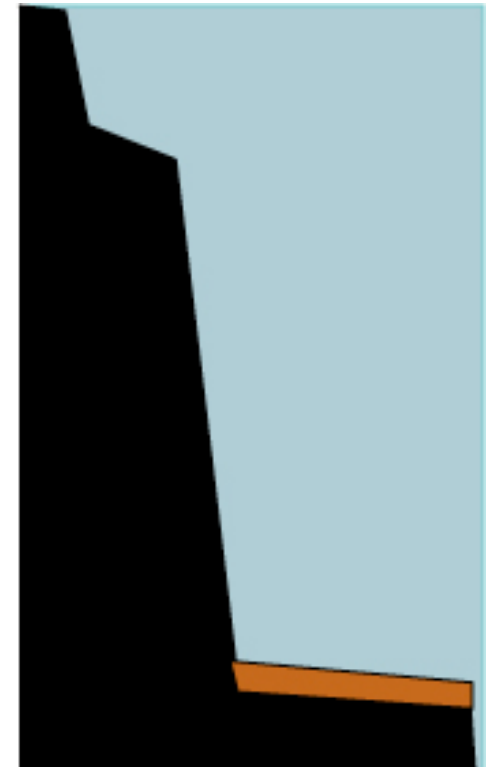
Before...



During...



and Voila!



# The effect of oxygen on carbon preservation in Madeira Abyssal Plain Turbidites

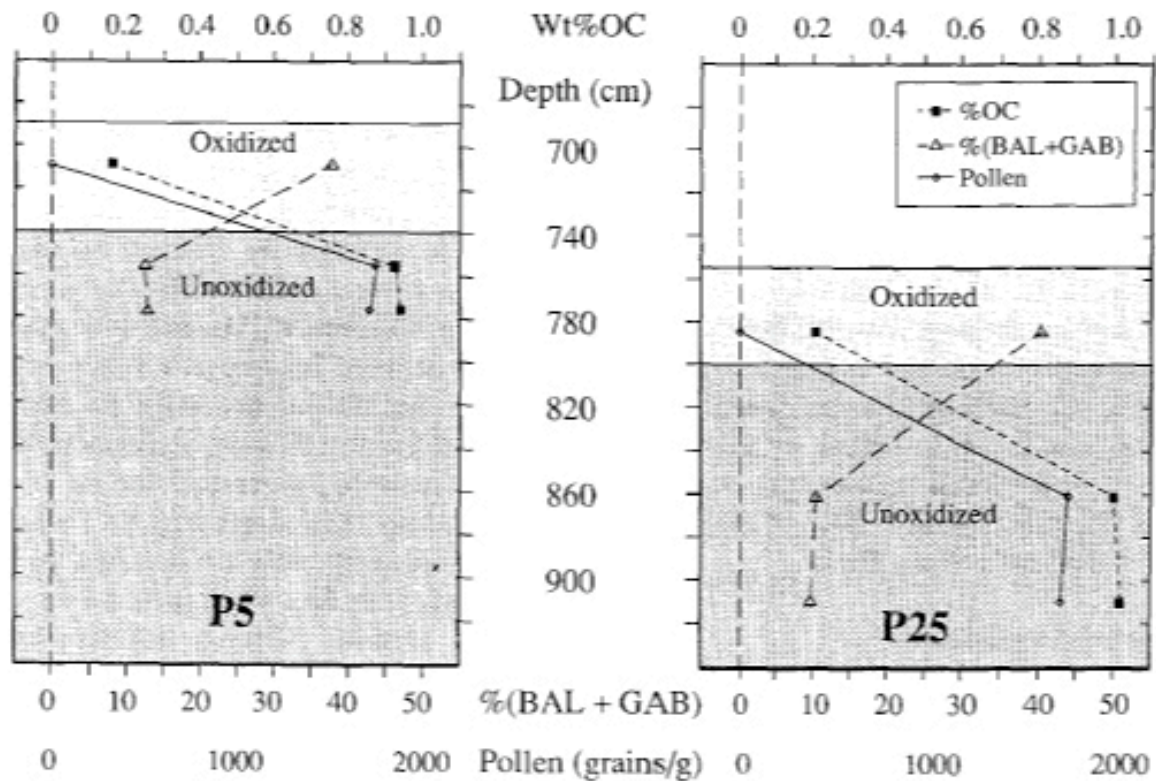
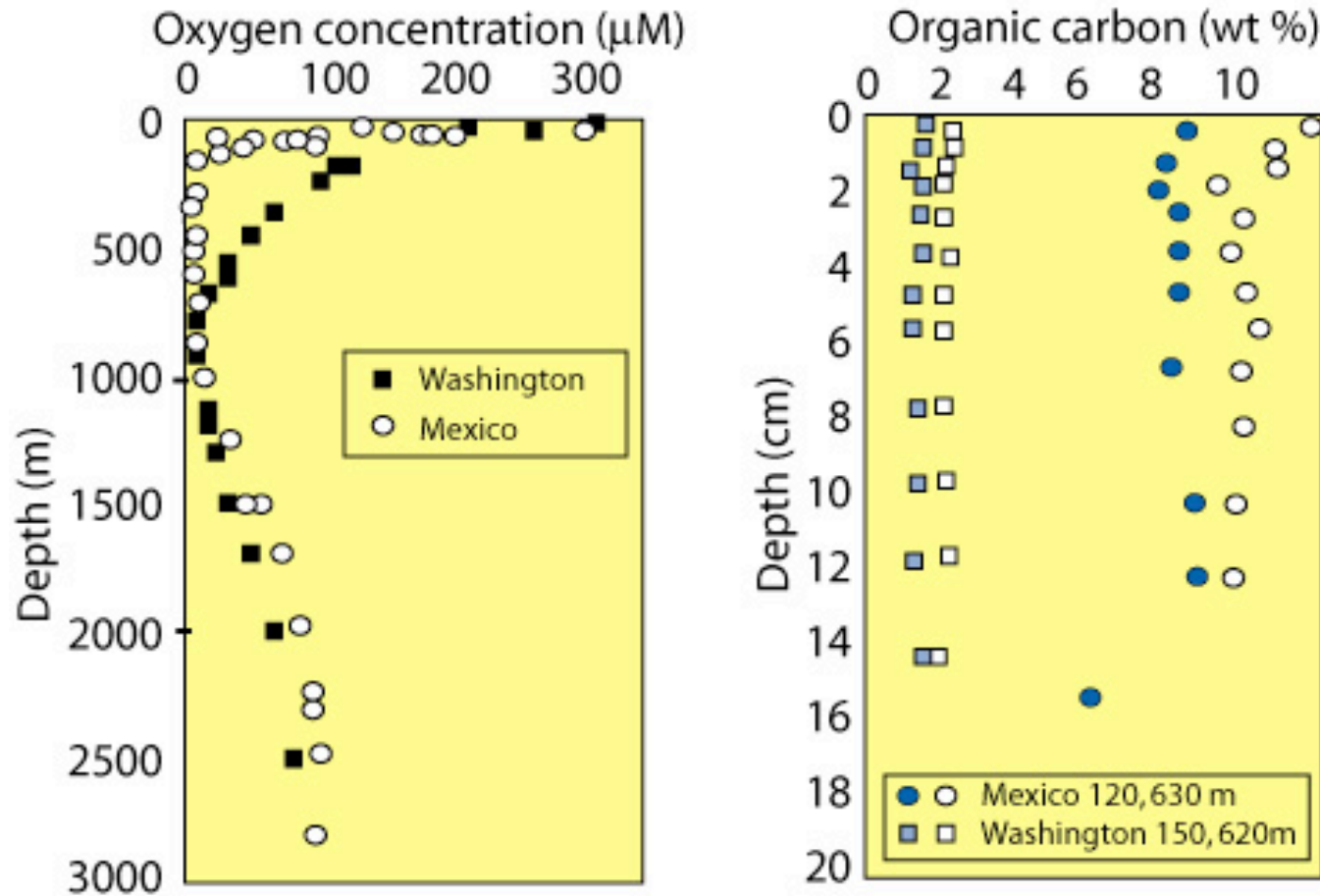


Fig. 10. Depth distributions of (a) weight percent organic carbon, (b) combined mole percent of two nonprotein amino acids ( $\beta$ -alanine plus  $\gamma$ -aminobutyric acid), and (c) total pollen abundances ( $\text{grains g}^{-1}$ ) in oxidized and unoxidized sediments from two cores of the f-turbidite collected at separate sites in the Madeira Abyssal Plain (data from Cowie et al., 1995; Keil et al., 1994b).

# The effect of oxygen on carbon preservation in continental margin sediments

pp on Mexican shelf < Washington shelf; sedimentation rates are similar; O<sub>2</sub> is very different



Hartnett et al. (1998) Nature v391, 572-574

## Comparison of Washington and Mexican margin sediments

**Table 1 Measured and calculated parameters for Washington and Mexican margin sediments**

	Bottom water O <sub>2</sub> ( $\mu\text{mol l}^{-1}$ )	Organic carbon (wt%)	Sediment accumulation rate* ( $\text{mg cm}^{-2}\text{yr}^{-1}$ )	Organic carbon burial ( $\mu\text{mol cm}^{-2}\text{yr}^{-1}$ )	Organic carbon oxidation ( $\mu\text{mol cm}^{-2}\text{yr}^{-1}$ )	Burial efficiency (%)	O <sub>2</sub> exposure time (yr)
<b>Washington</b>							
Shelf†							
Average ( <i>n</i> = 8)	92.8	1.31	102	117	675	15.1	3.92
Range	(77-106)	(0.55-1.8)	(61-130)	(28-169)	(506-790)	(3.75-17.8)	(1.1-10.5)
Upper slope							
Average ( <i>n</i> = 6)	71.5	1.59	66.9	90.1	316	25.7	6.42
Range	(38-104)	(0.5-2.8)	(37-100)	(27-210)	(91-607)	(4.69-42.7)	(1.4-14.4)
<b>Mexico</b>							
Shelf‡							
Average ( <i>n</i> = 4)	15.4	4.7	14.9	50.9	183	23.3	0.252
Range	(3-0)	(2.9-7.1)	(9.1-25.6)	(195-392)	(157-200)	(18.8-26.3)	(0.051-0.587)
Slope							
Average ( <i>n</i> = 4)	5.3	10.1	6.87	61.7	91.3	38.2	0.032
Range	(0.0-12)	(7.5-12.8)	(4.1-12.7)	(160-845)	(55-121)	(19.9-53.1)	(0.0-0.16)

\* Sediment accumulation rates and wt% OC determinations were made on the same cores; these cores were collected at the stations where benthic flux chamber measurements were made.

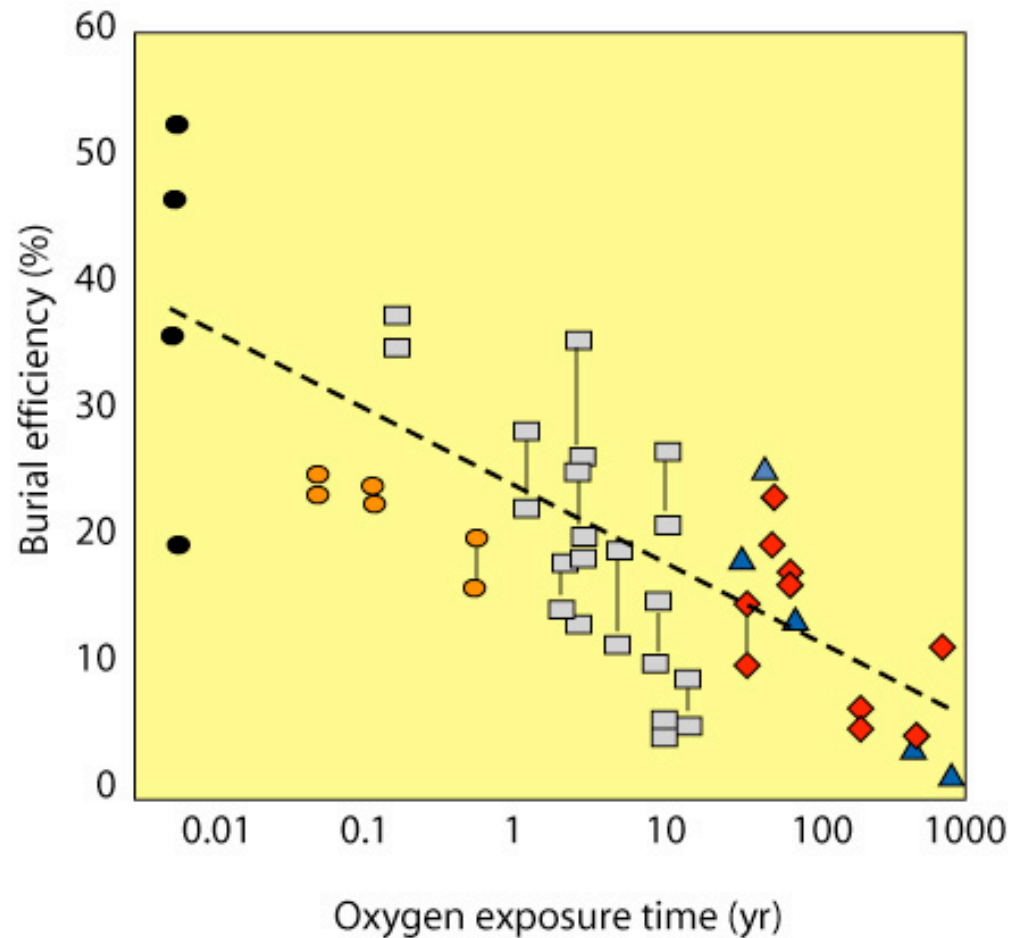
† Washington shelf and upper-slope stations had depths in the range 0-200 m and 200-600 m, respectively.

‡ The Mexican shelf extends to ~150 m and the Mexican slope stations range from 150 to 1,000 m.

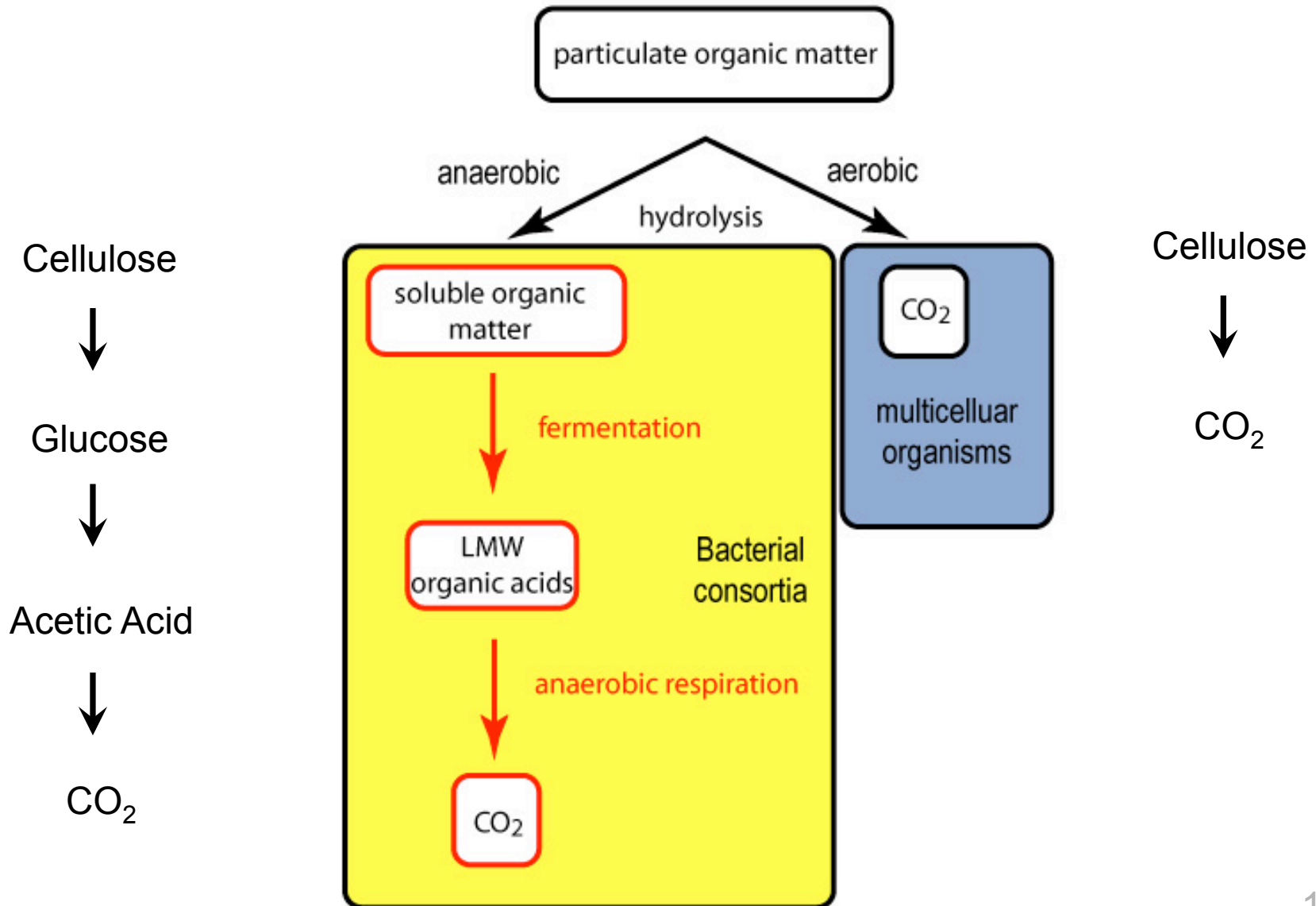


The effect of oxygen has been refined somewhat to adjust for differences in exposure time, which is related to sedimentation rate (depth of O<sub>2</sub> penetration/sedimentation rate) = OET

Effect of oxygen exposure time on burial efficiency

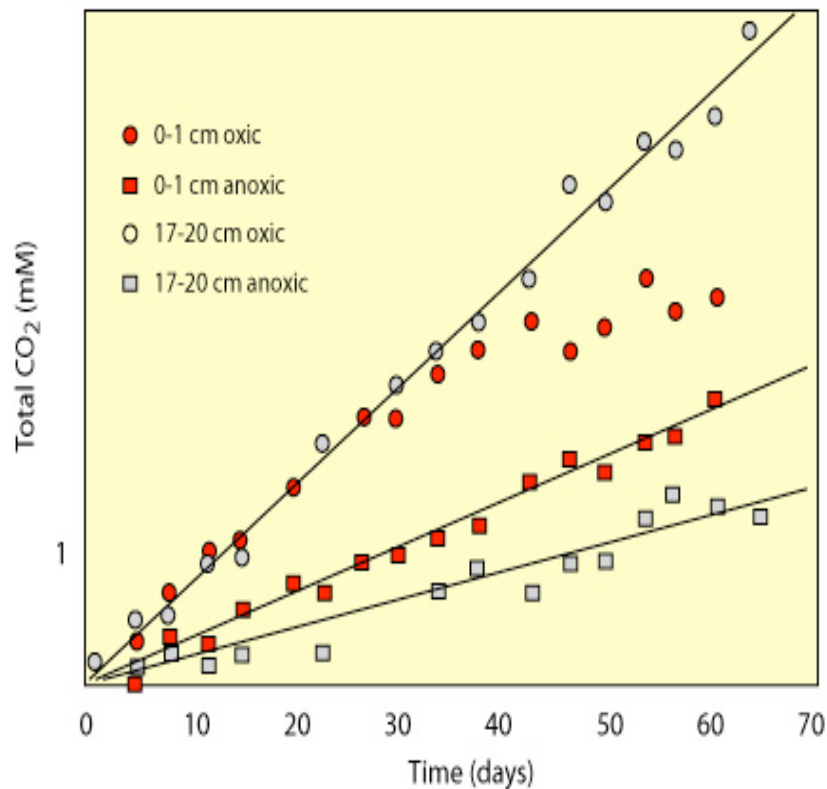


# How does oxygen act to decrease carbon preservation in sediments?

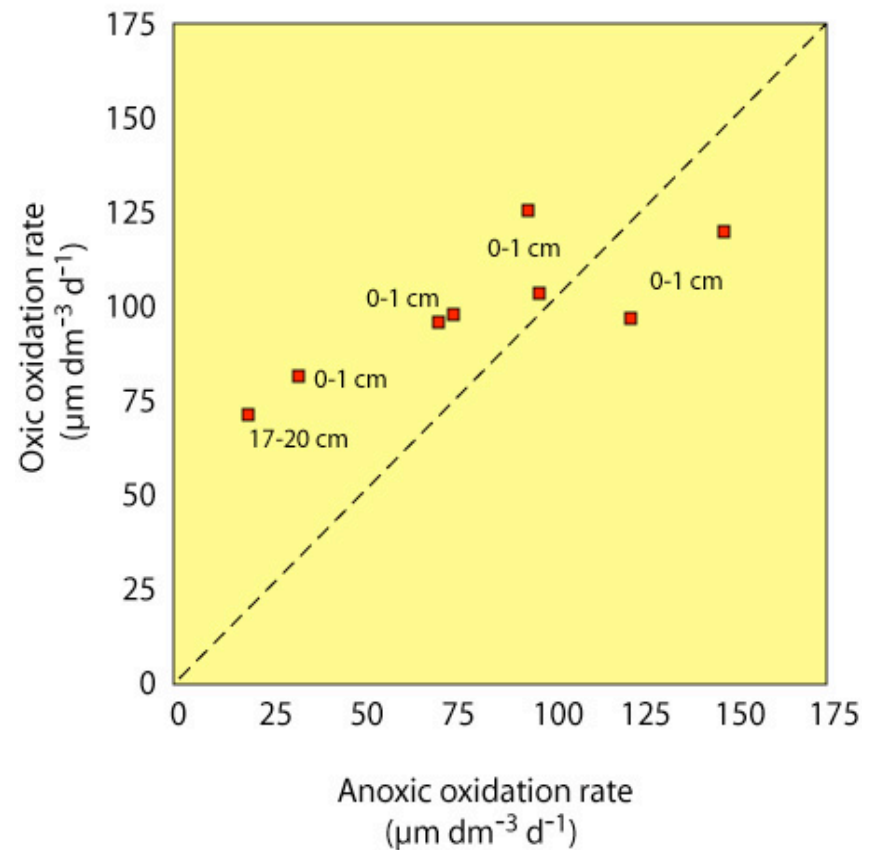


# Is carbon more efficiently respired under oxic or anoxic conditions ?

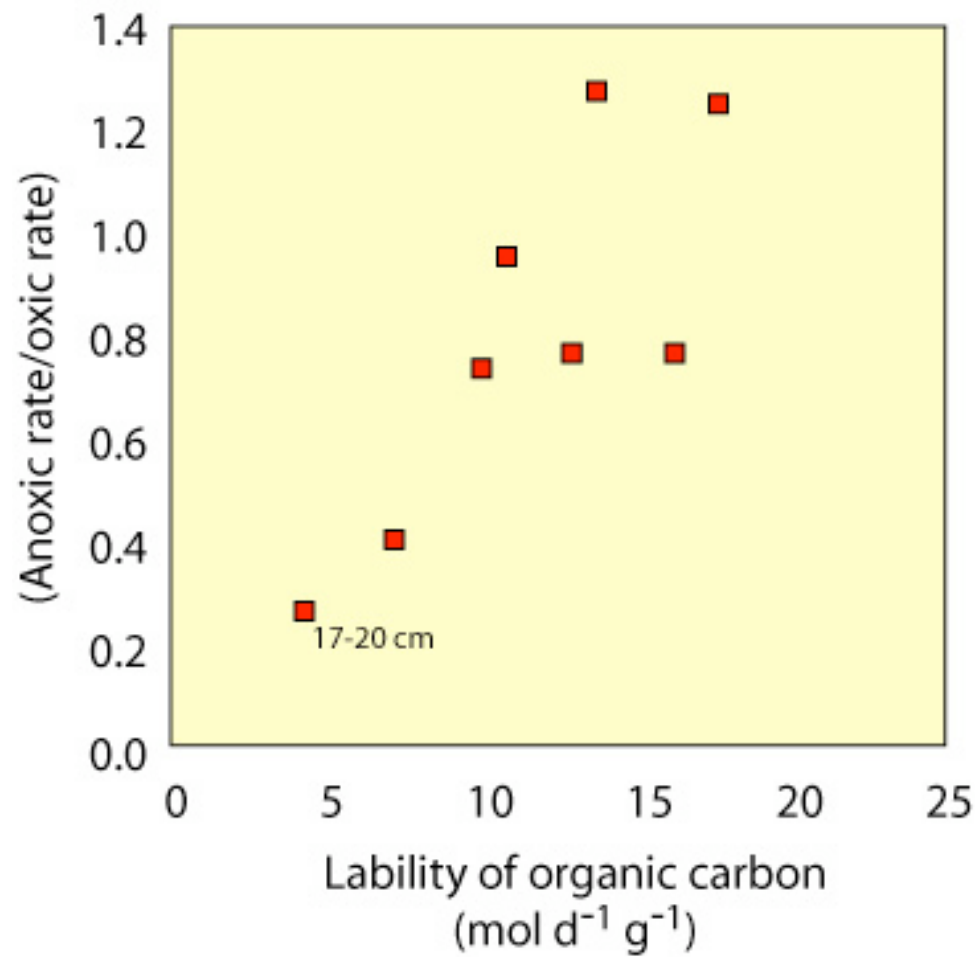
Respiration of carbon in 0-1 cm and 17-20 cm sediment under oxic and anoxic conditions



Comparison of oxic and anoxic degradation rates in surface and deep sediments



### Oxidation rate and the lability of organic carbon



If you want to understand why C is preserved in marine sediments, look at where it is buried....

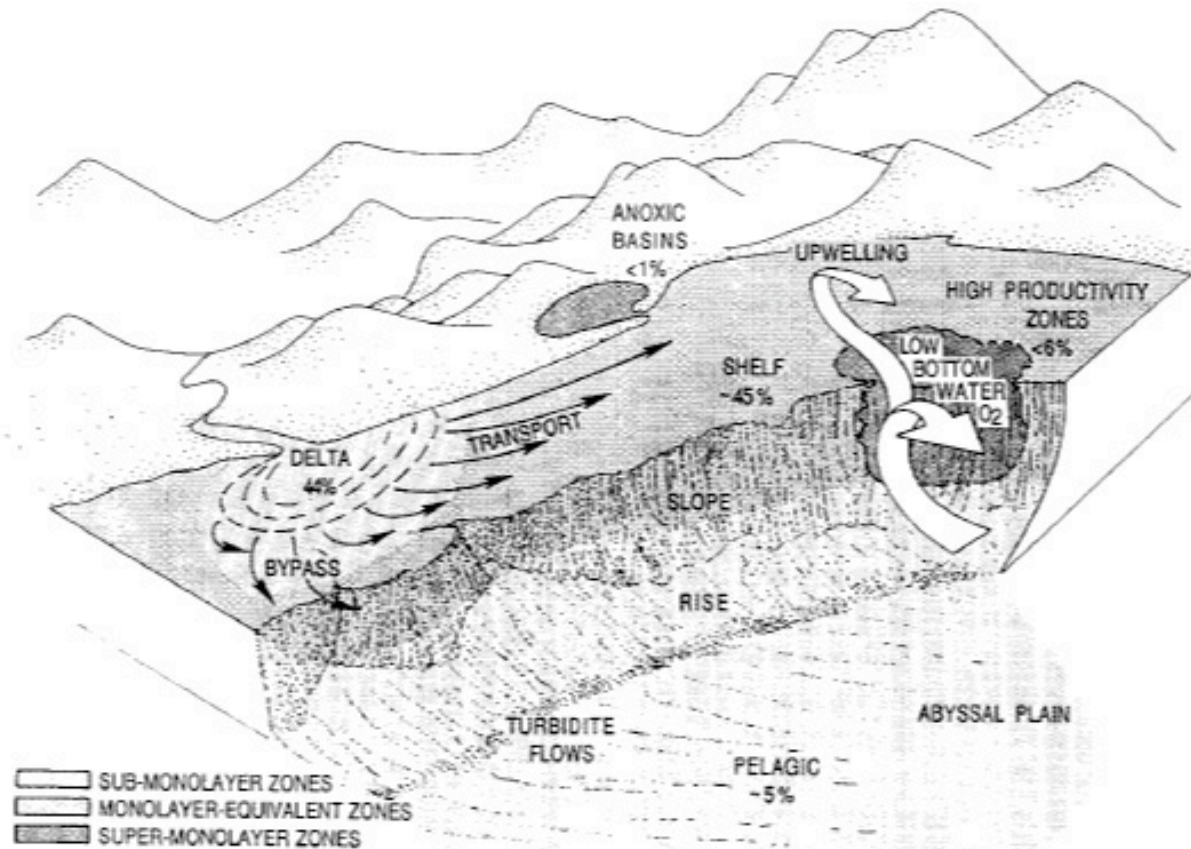
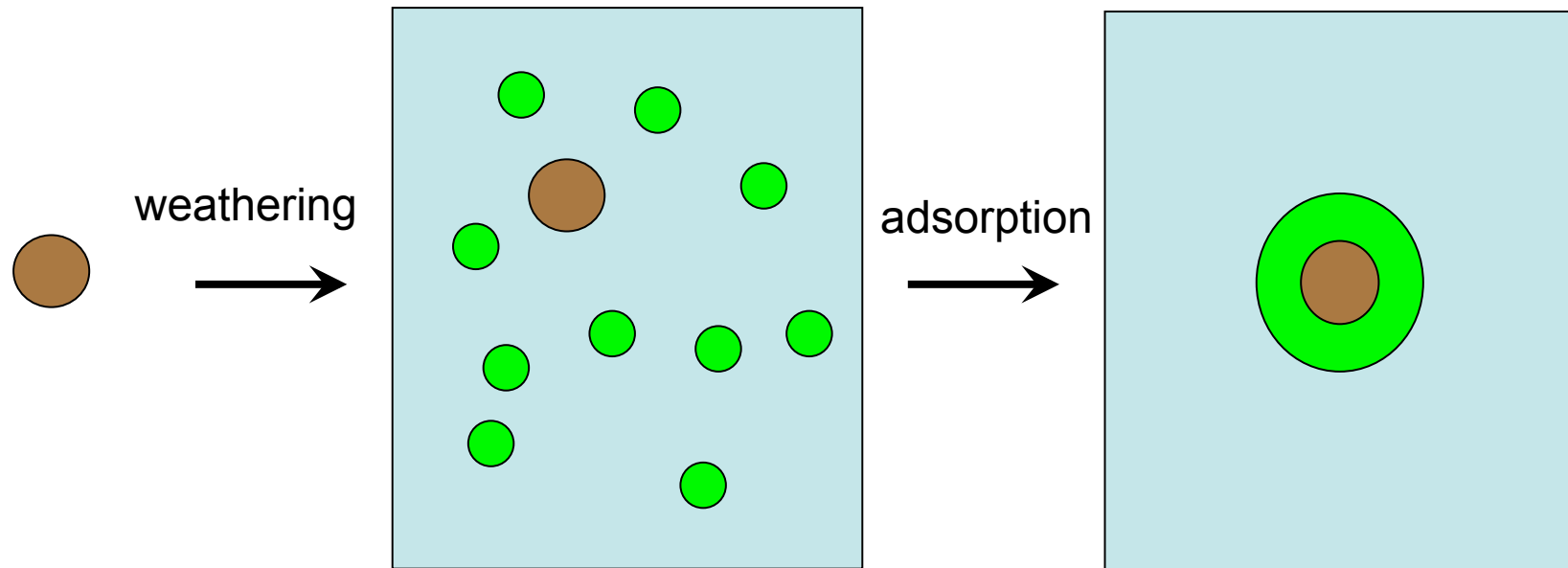


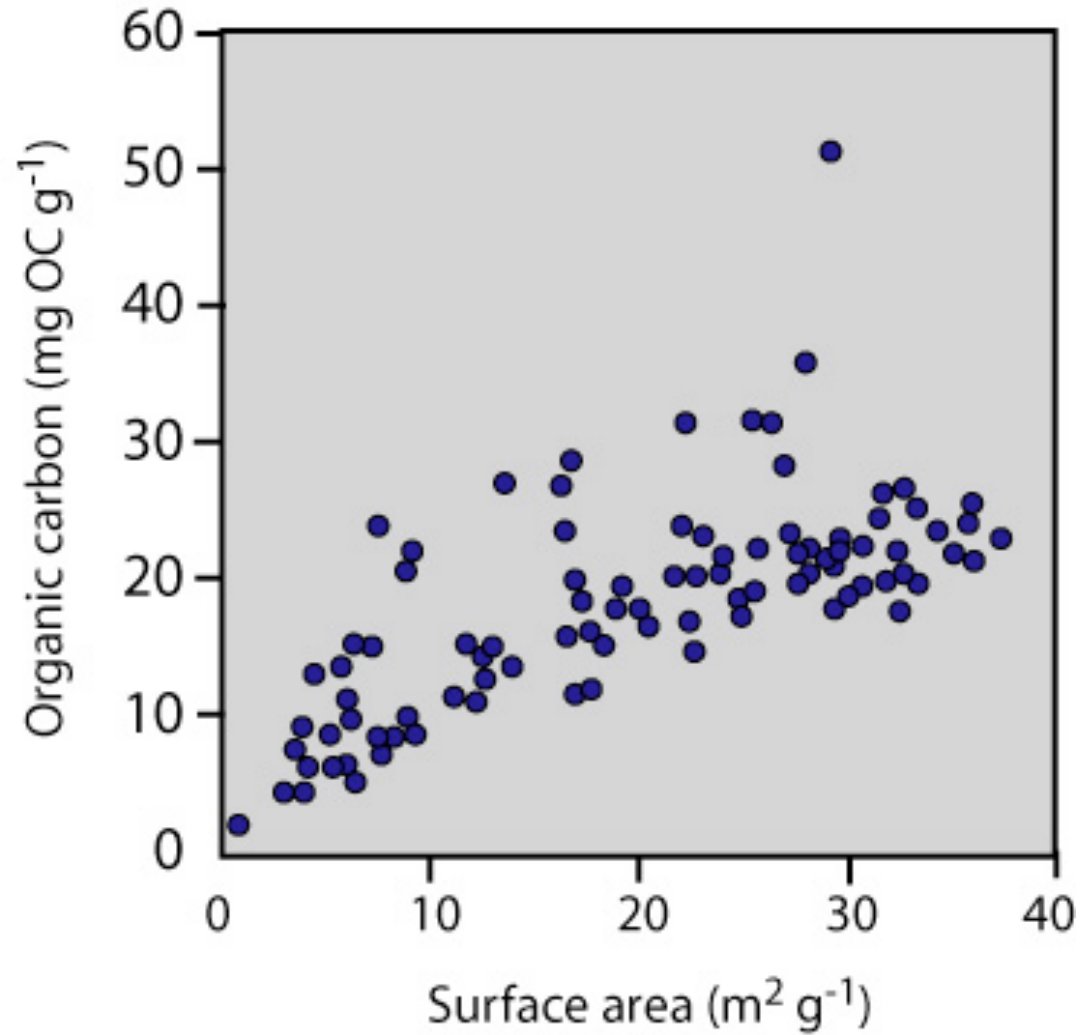
Fig. 1. Idealized diagram depicting current estimates of the percentage of total organic matter burial occurring within various marine sediment types (see Table 2). Light sections represent sediments which contain organic loadings lower than a monolayer equivalent. Stippled sediments contain monolayer-equivalent loadings, and dark sediments contain loadings that are more than monolayer-equivalent.

## Protection and preservation of C on mineral surfaces

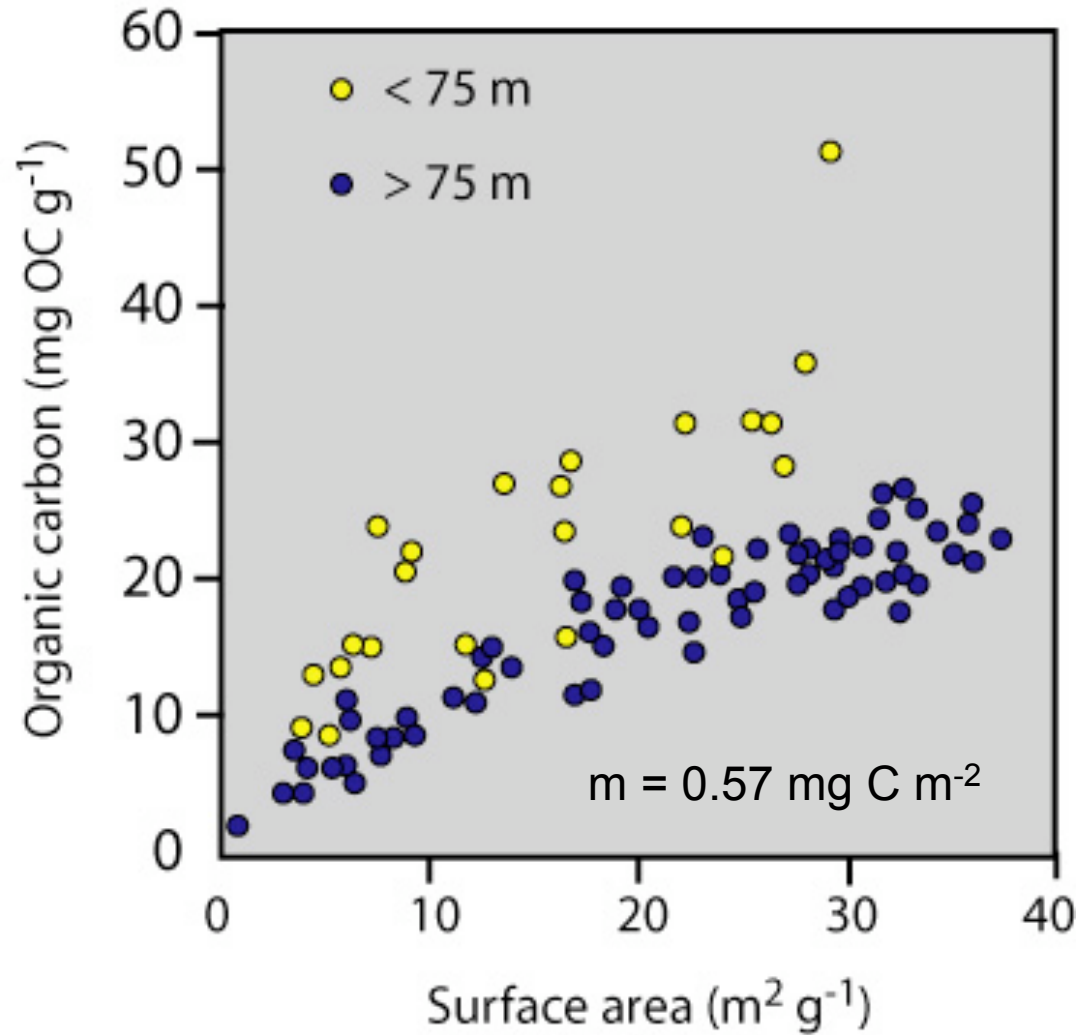
Larry Mayer and others reasoned that there is no such thing as a naked mineral surface in seawater. Further, the amount of C that can be loaded onto a sediment particle is proportional to its surface area.



# Organic carbon vs surface area for sediments from the Gulf of Maine

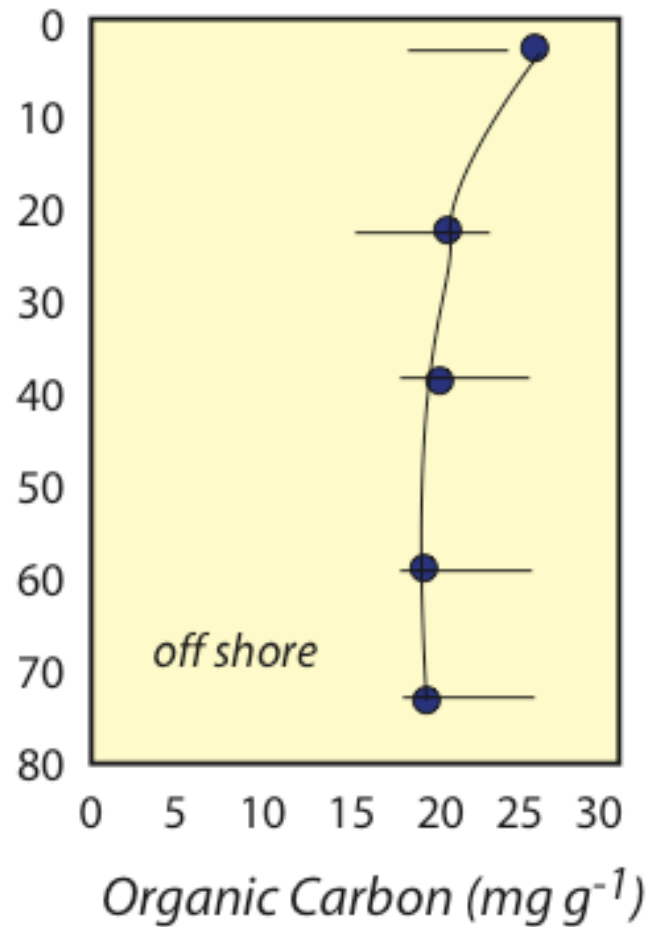
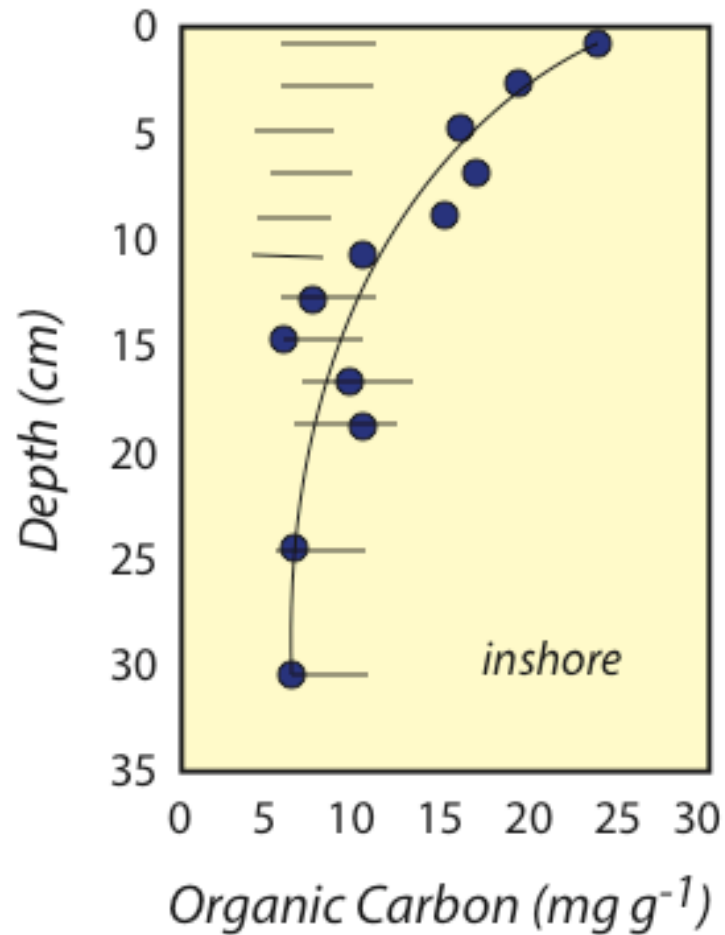


# Organic carbon vs surface area for sediments from the Gulf of Maine



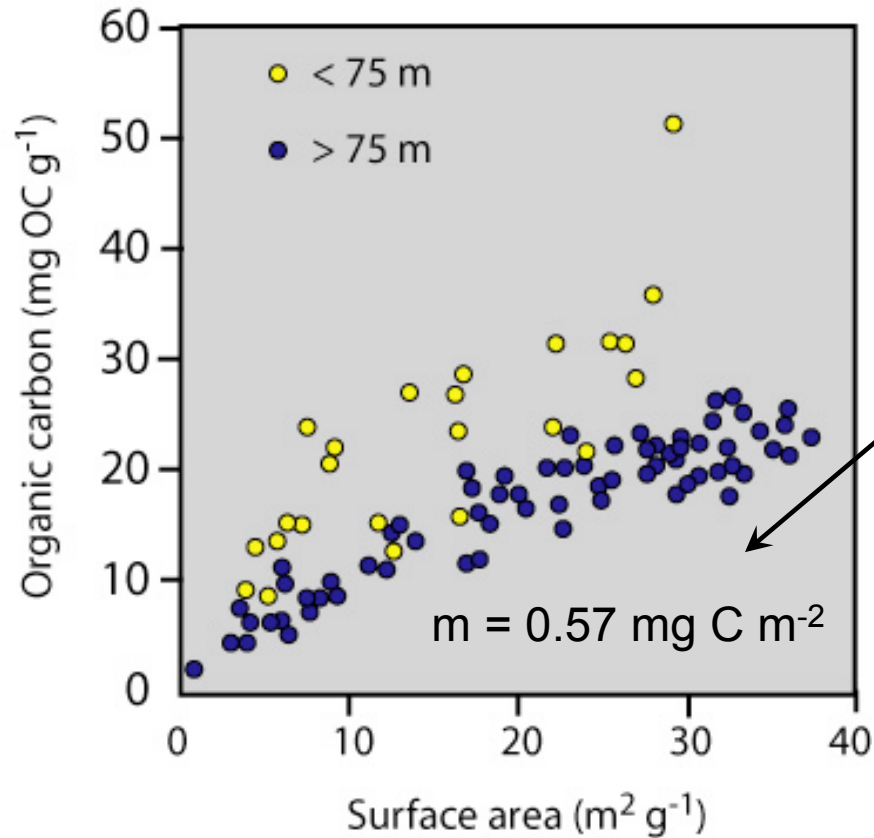


*Organic carbon, mineral surface area, and depth  
in Gulf of Maine Sediments*

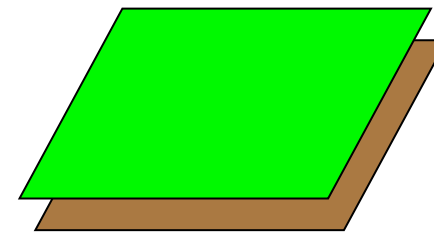


Sediments may be overloaded with C due to biogeochemical cycling, but eventually diagenesis will reduce the C load to a set surface area vs %OC value

# Organic carbon vs surface area for sediments from the Gulf of Maine



Monolayer equivalent (ME) loading



Surfaces are coated with organic matter to the equivalent of one molecule thick...

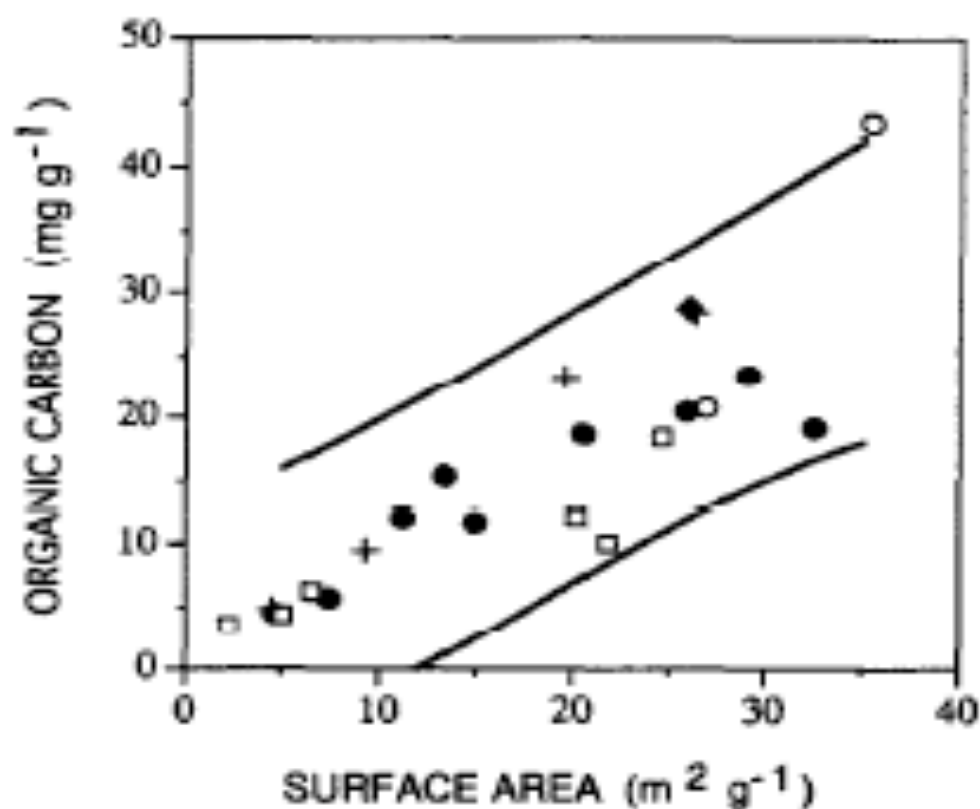
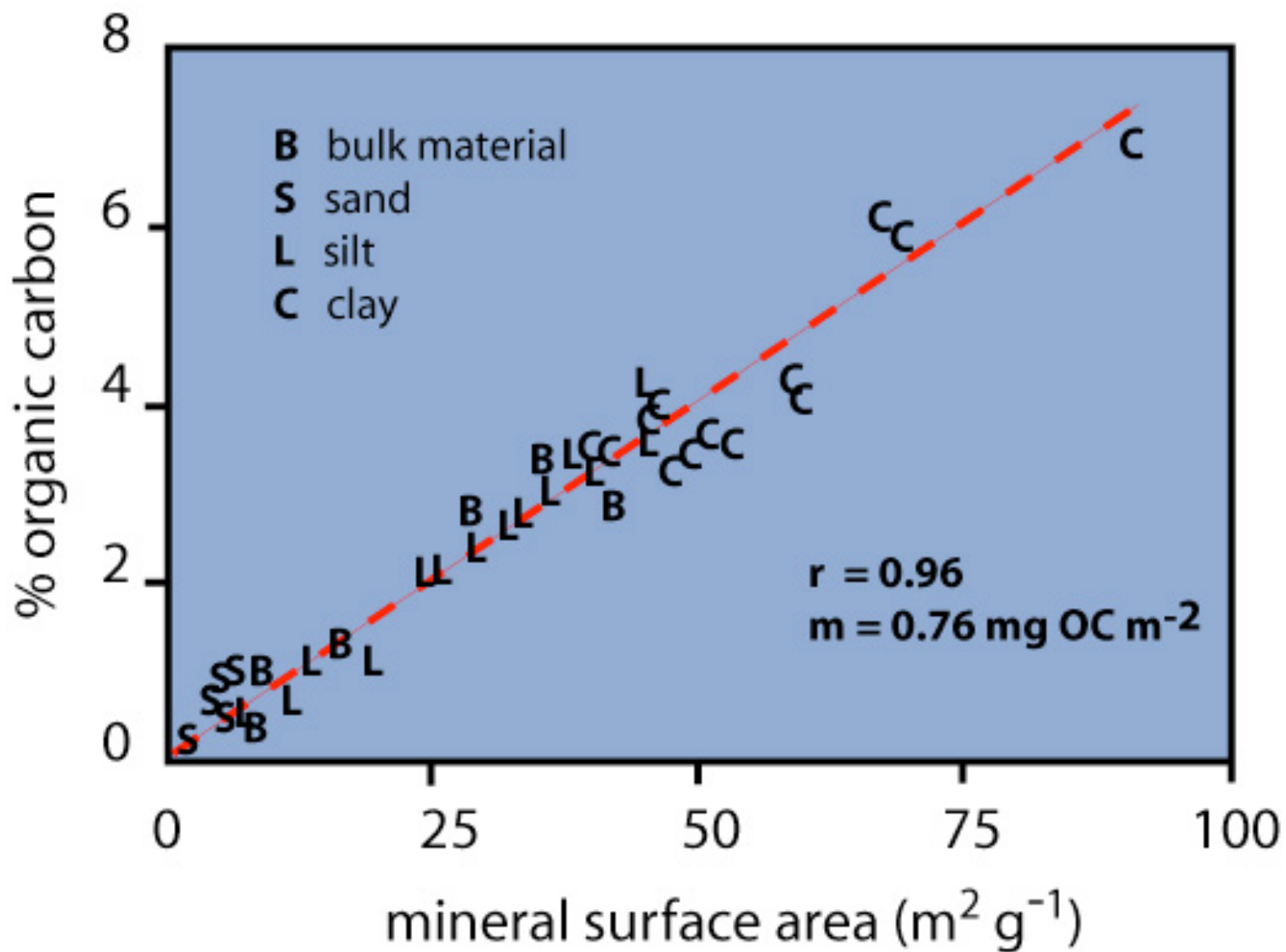
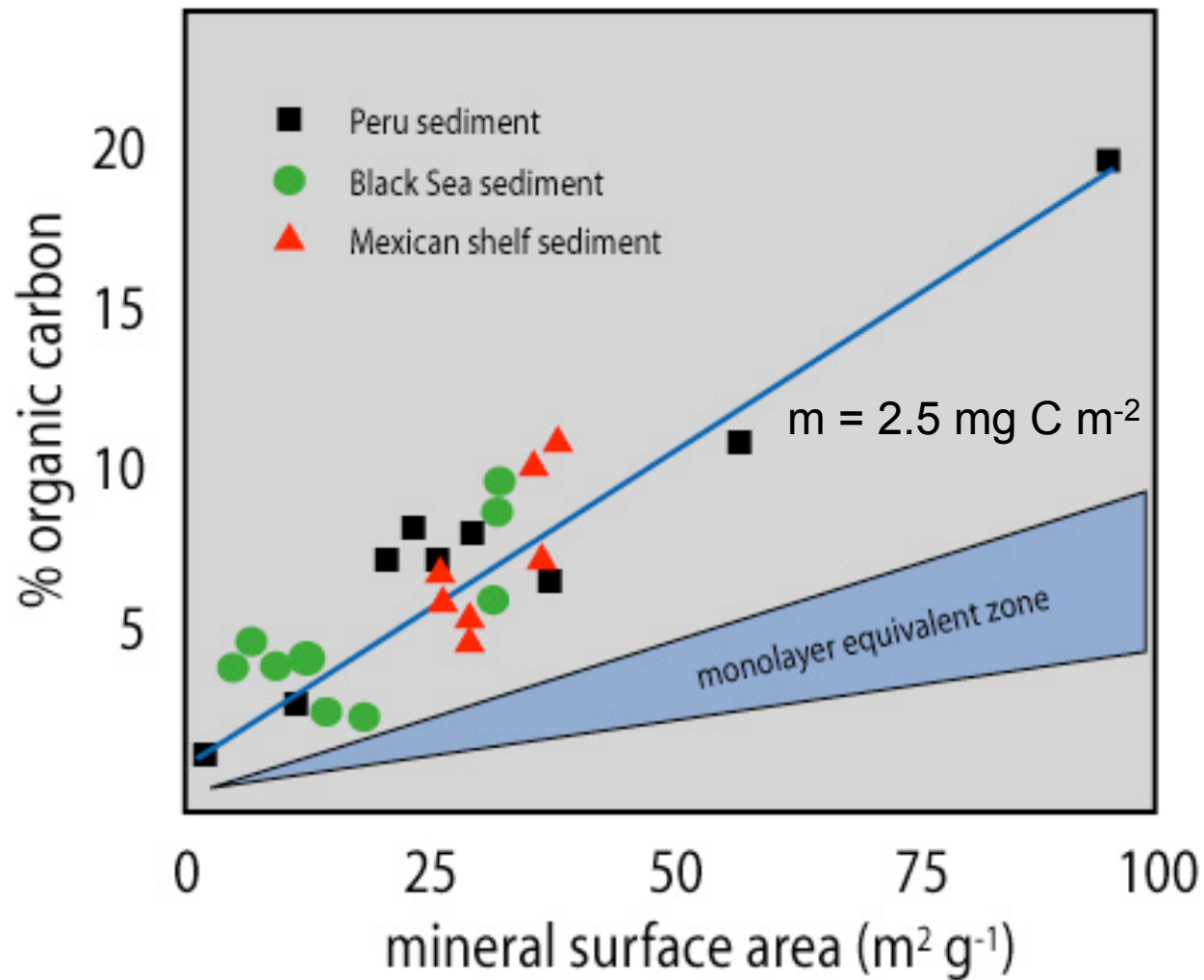


FIG. 4. Total organic carbon concentration vs. surface area for the refractory background of twenty-two cores from throughout North America. The OC concentration in the refractory background was calculated by fitting downcore data to a two-component decay model (see text). Diagonal bars are the 95% confidence intervals for this data set, and represent the ME zone as defined by the refractory background data. Symbols: □, CHUKCHI; ◆, SKAN BAY; ○, PUGET SOUND; +, CHESAPEAKE; ●, GULF OF MAINE.

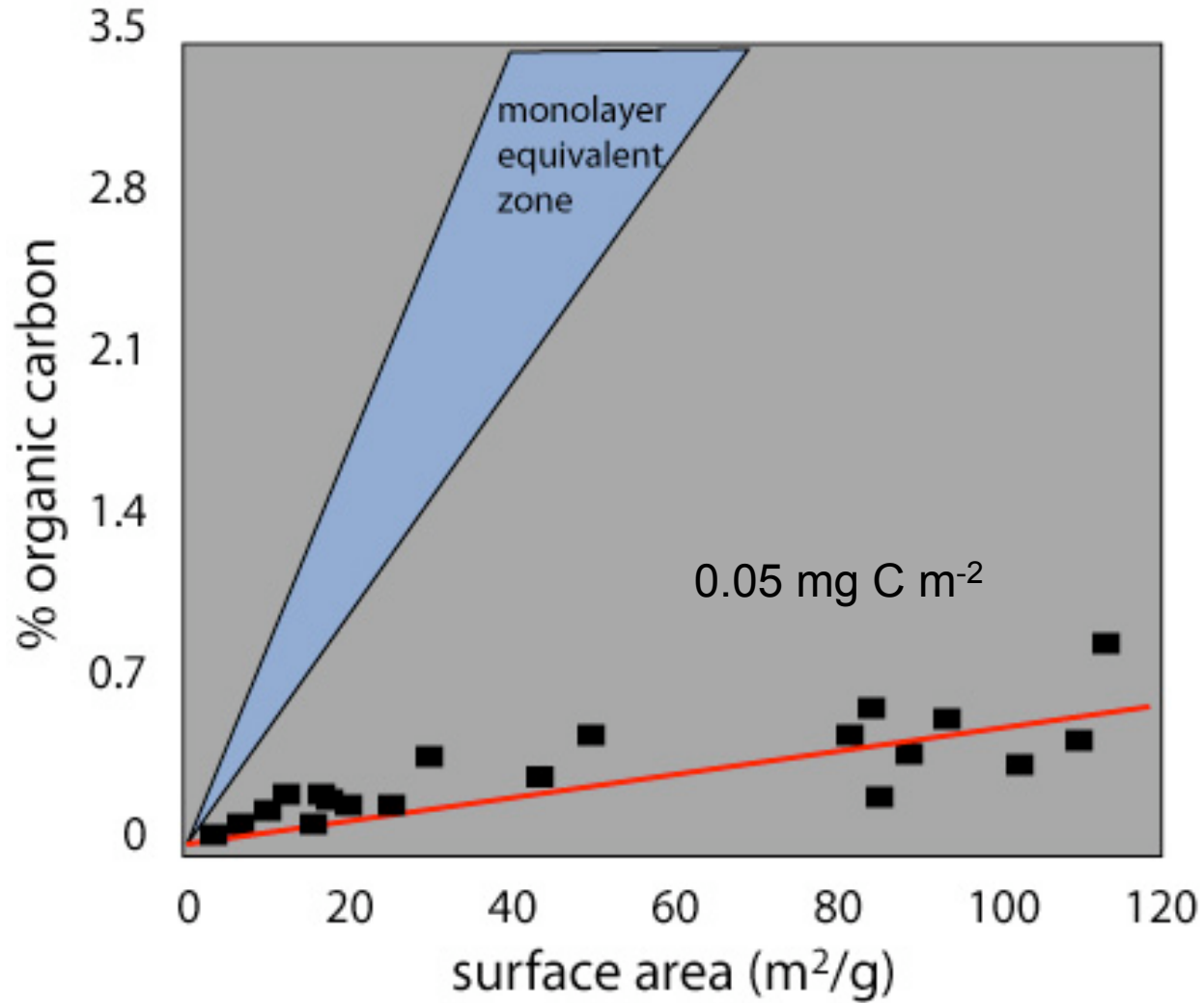
Mineral surface area vs %organic carbon for Columbia River Sediments  
(Hedges and Keil, Mar. Chem (1995) 49, 81-115.)



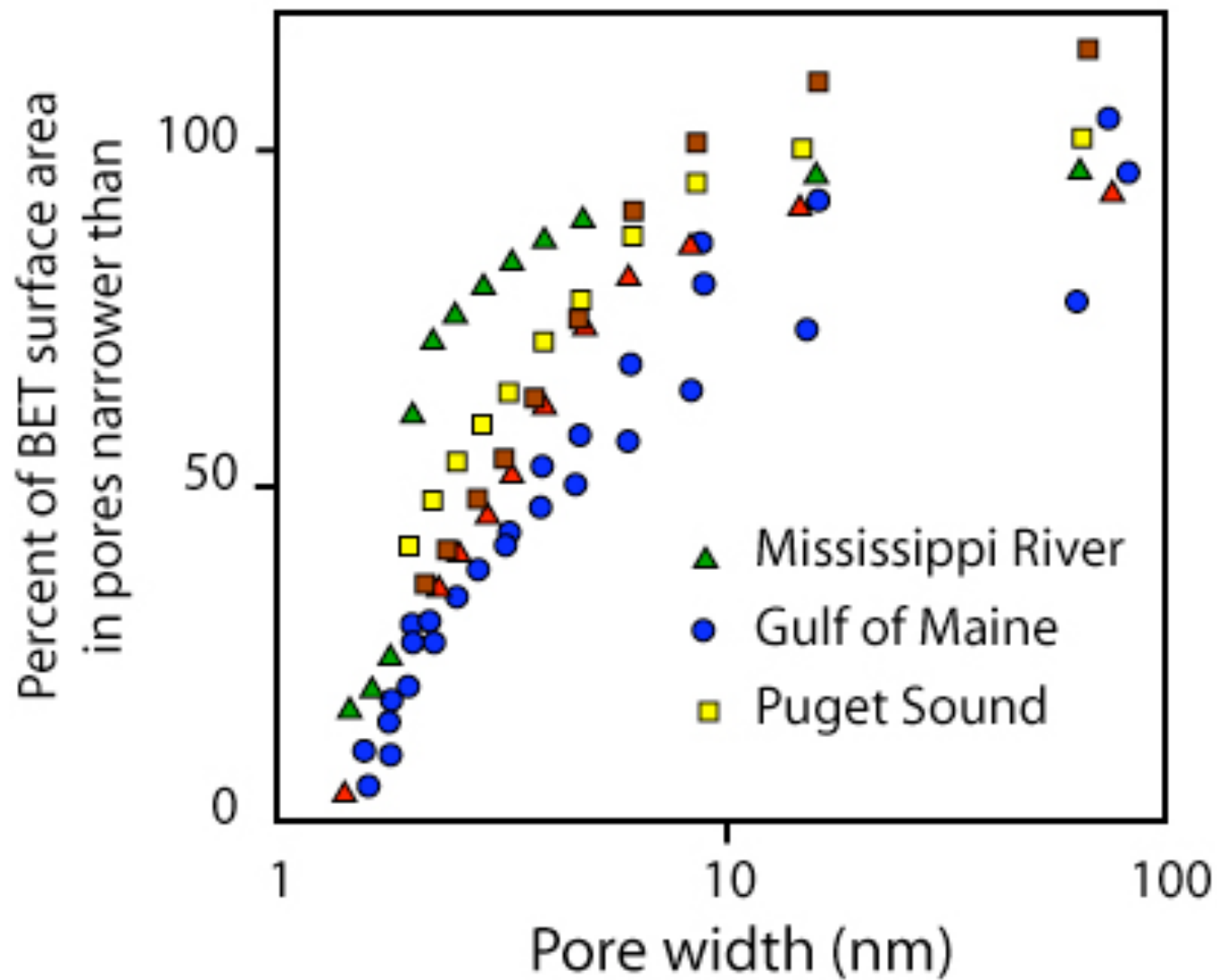
# Surface area vs %organic carbon for sediments from low oxygen depositional regimes



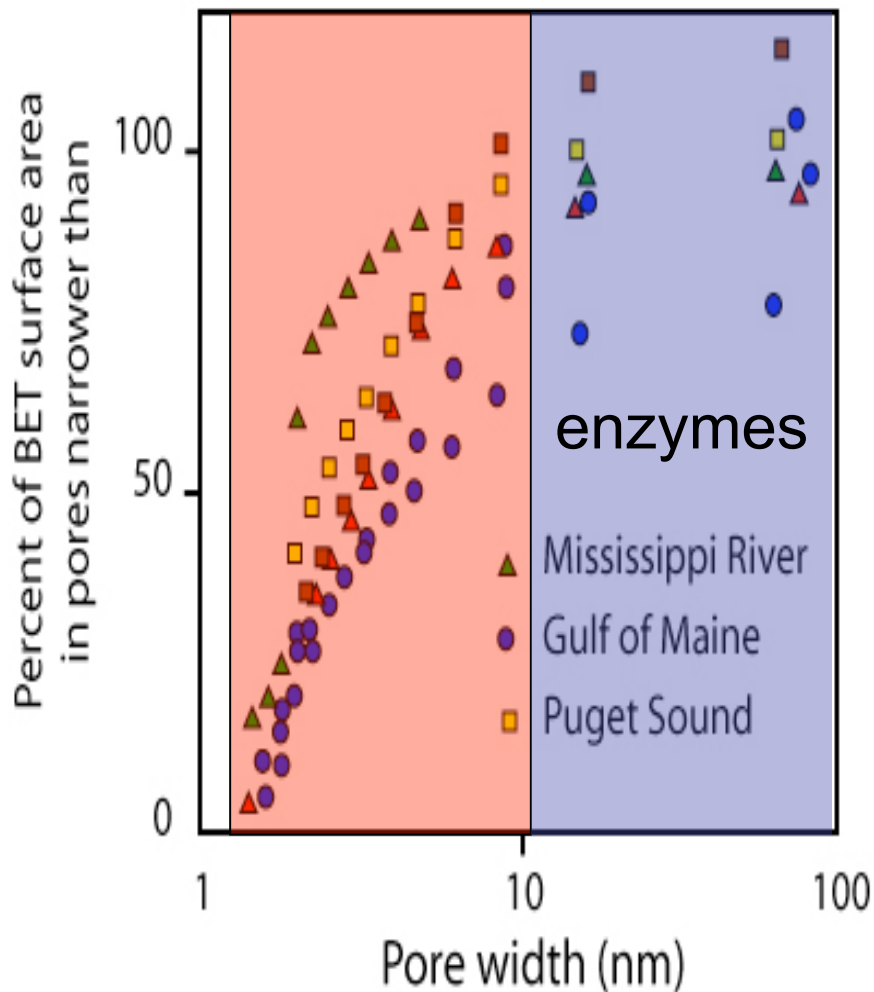
# Surface area vs % organic carbon for Equatorial Pacific sediments



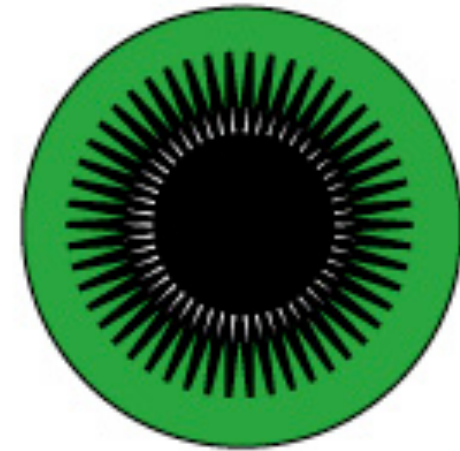
## Distribution of mineral pore sizes in marine sediments



Distribution of mineral pore sizes  
in marine sediments



organic coated particle



enzyme



ME coating of  
organic matter



## Surface area control on OC preservation in marine sediment..

Weathering introduces new mineral surfaces constantly to the environment.

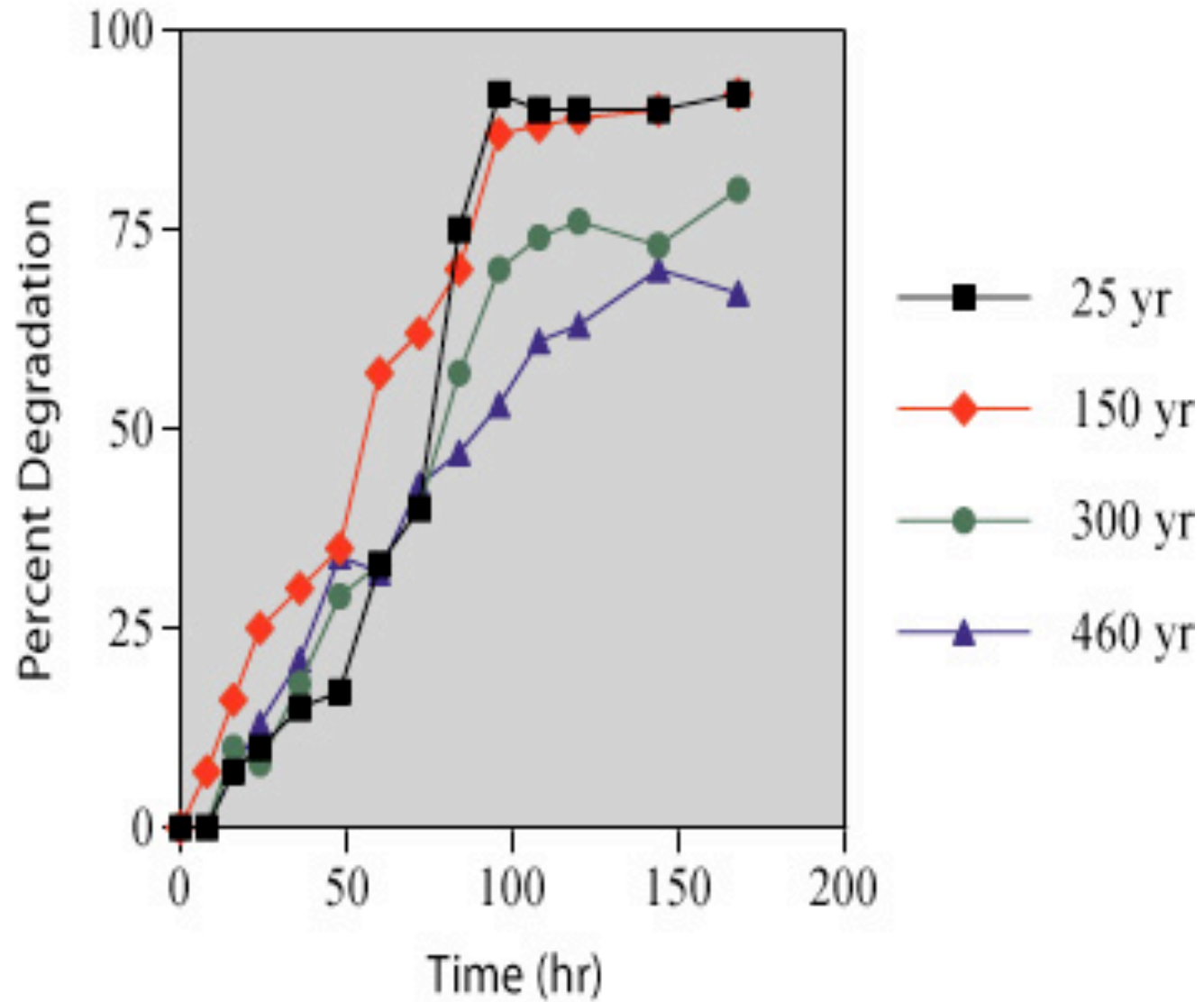
These surfaces ultimately become coated with organic matter, at approximately a monolayer equivalent loading.

Under conditions that are typical for sediment deposition on continental margins (where most C is buried) degradation proceeds to the ME loading and slows sufficiently there after to preserve this amount of carbon.

In open ocean setting, where oxygen exposure times are much longer, degradation proceeds to  $<$  ME loadings. In anoxic basins, where oxygen exposure times are much shorter, loadings are  $>$  ME.

**Mechanism** is preservation in small pores that are inaccessible to enzymes.  
e.g. *physical protection*.

Is this evidence for or against physical protection ?

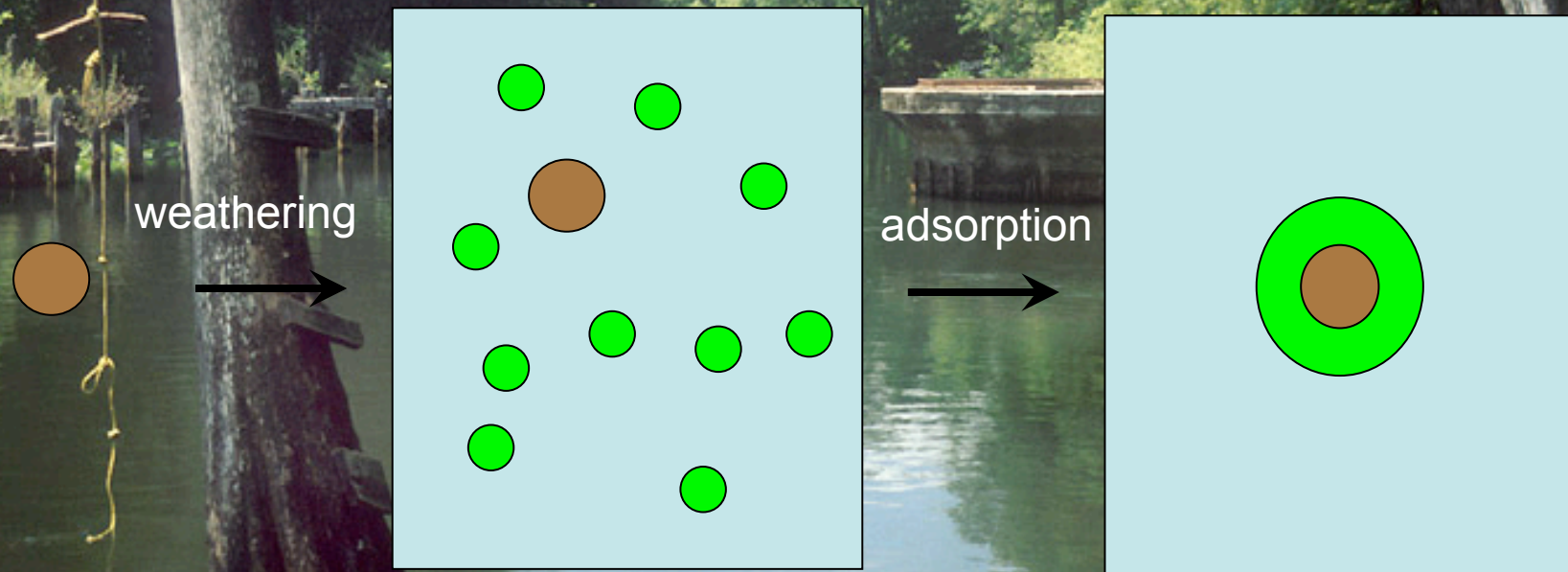




Oklawaha River  
Photo by A. Murray  
Copyright 2002 Univ. Florida



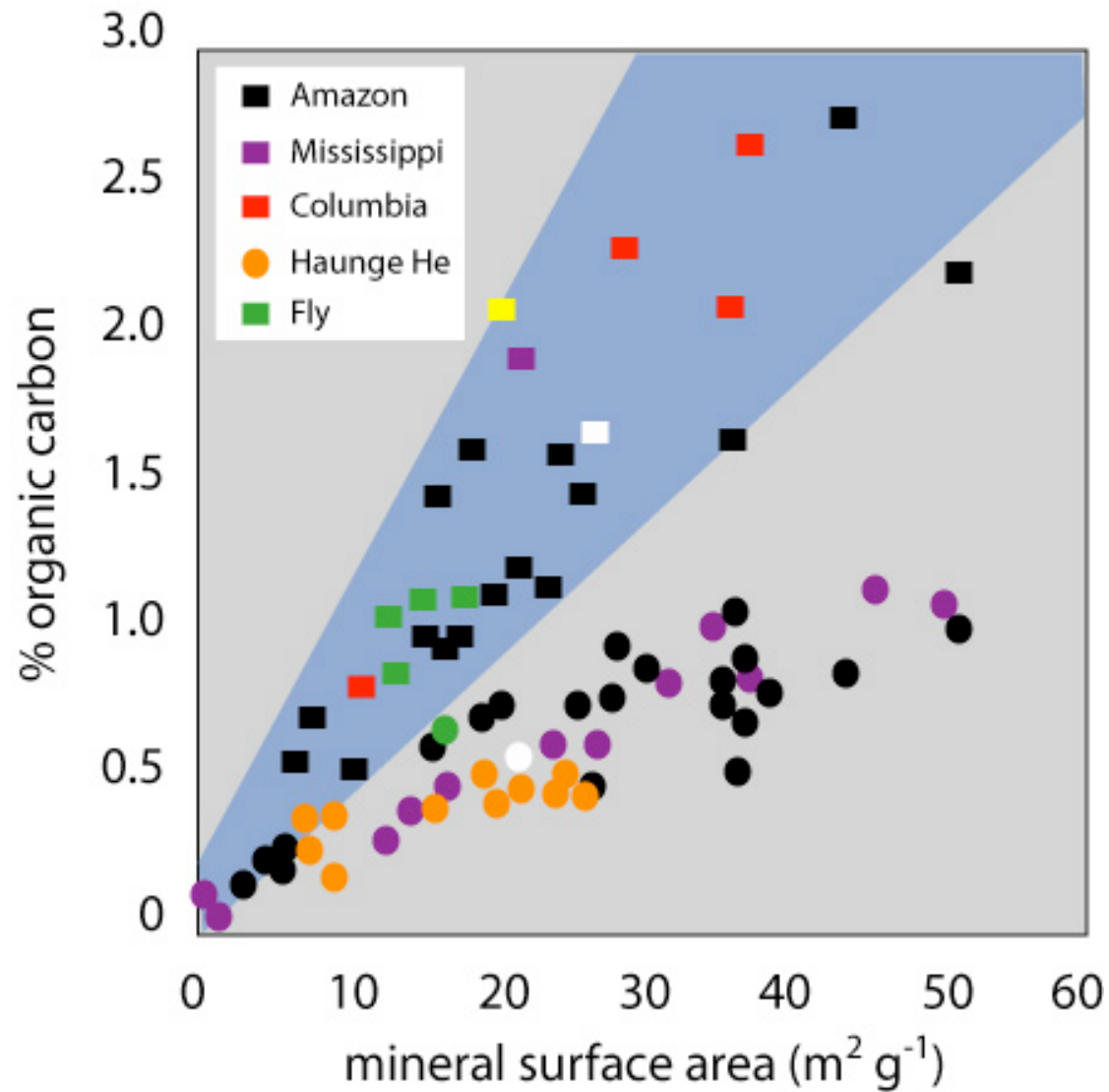
But there is the problem with this model...



.....think of the  $\delta^{13}\text{C}$  of marine sediments.

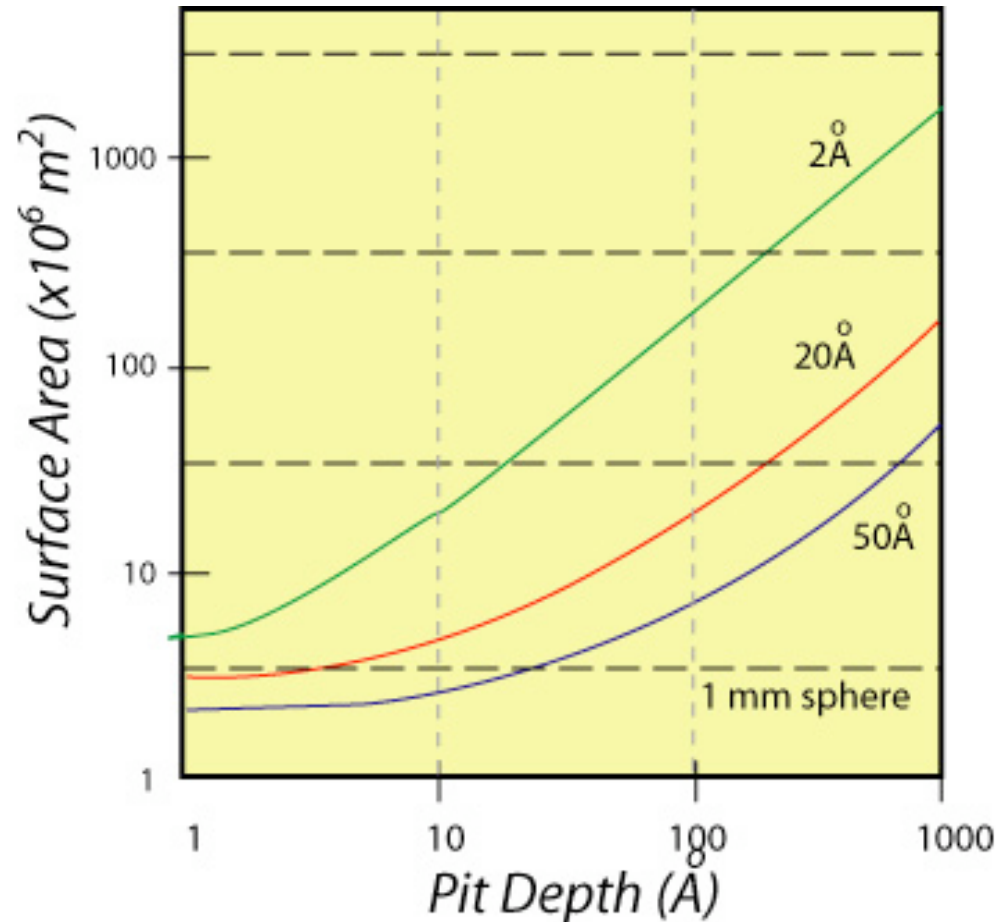
Okiawana River  
Photo by A. Murray  
Copyright 2002 Univ. Florida

## Surface area vs % organic carbon for deltaic and river sediments



*The rebuttal to surface area control on OC preservation.....*

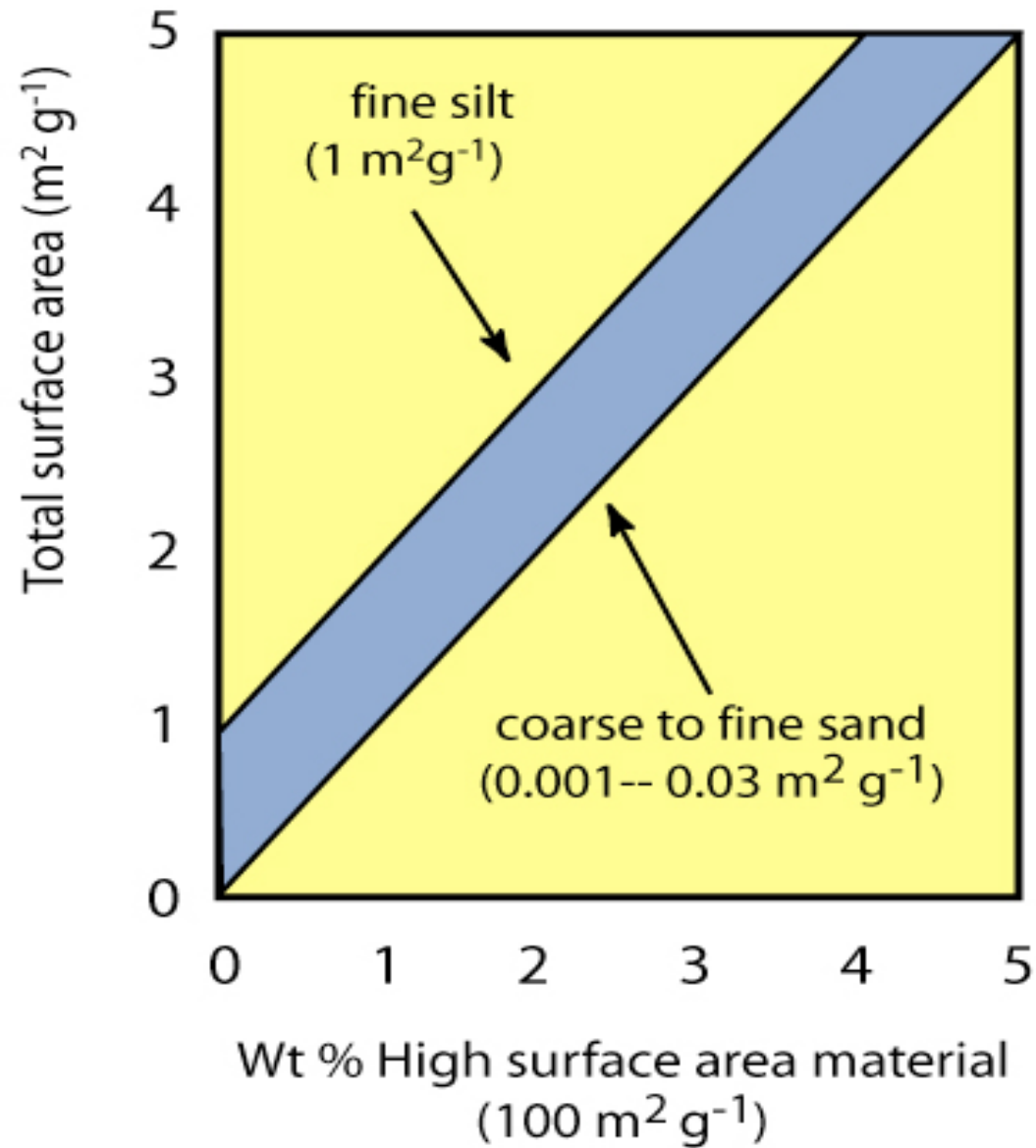
Theoretical surface area of a 1 mm pitted spherical particle



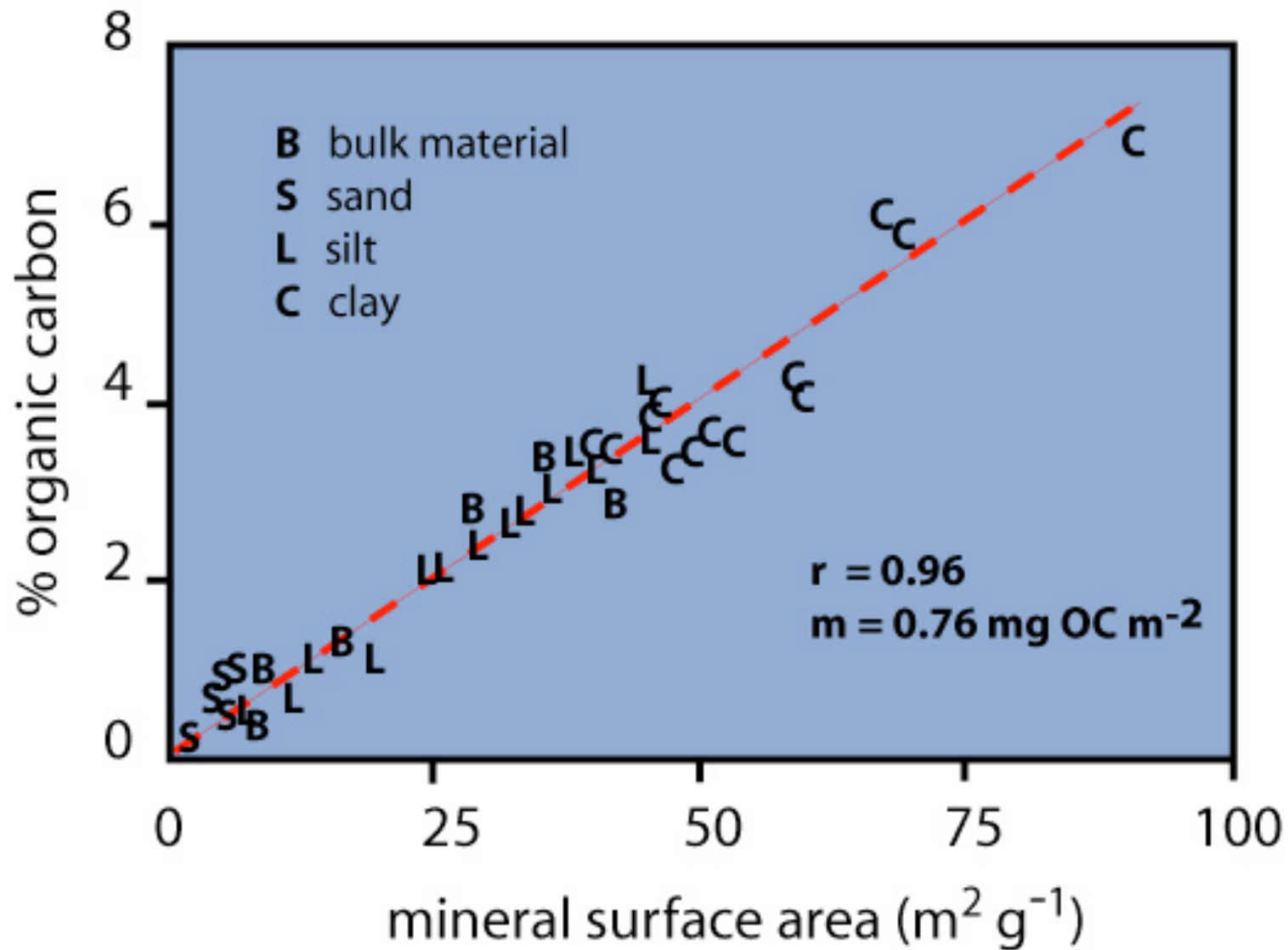
*It is impossible to physically protect that much organic matter in pits & cracks*

# Effect of high surface area material on total surface area

Ransom et al., GCA (1998) 62, 1329-1345

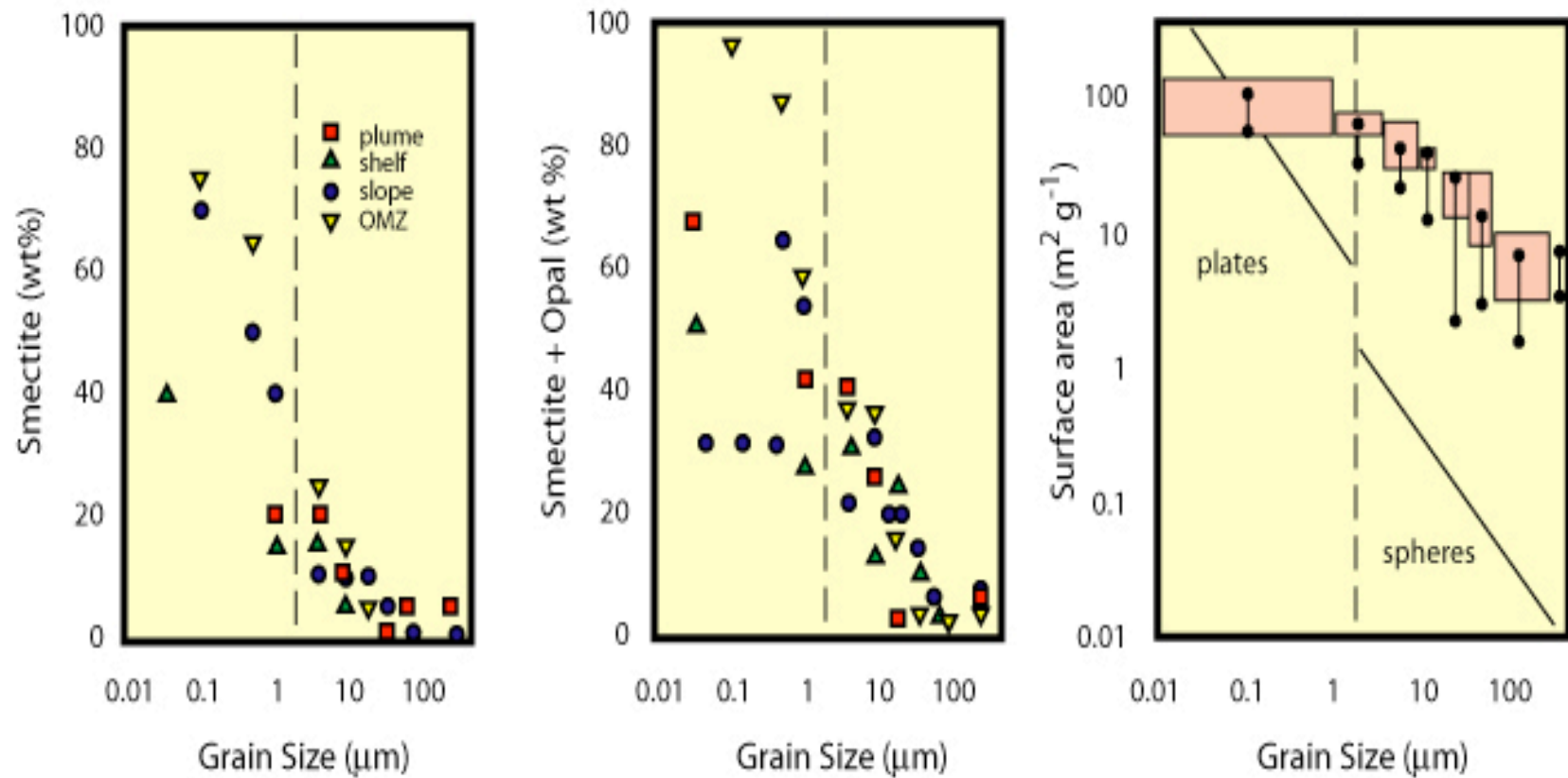


Mineral surface area vs %organic carbon for Columbia River Sediments  
(Hedges and Keil, Mar. Chem (1995) 49, 81-115.)

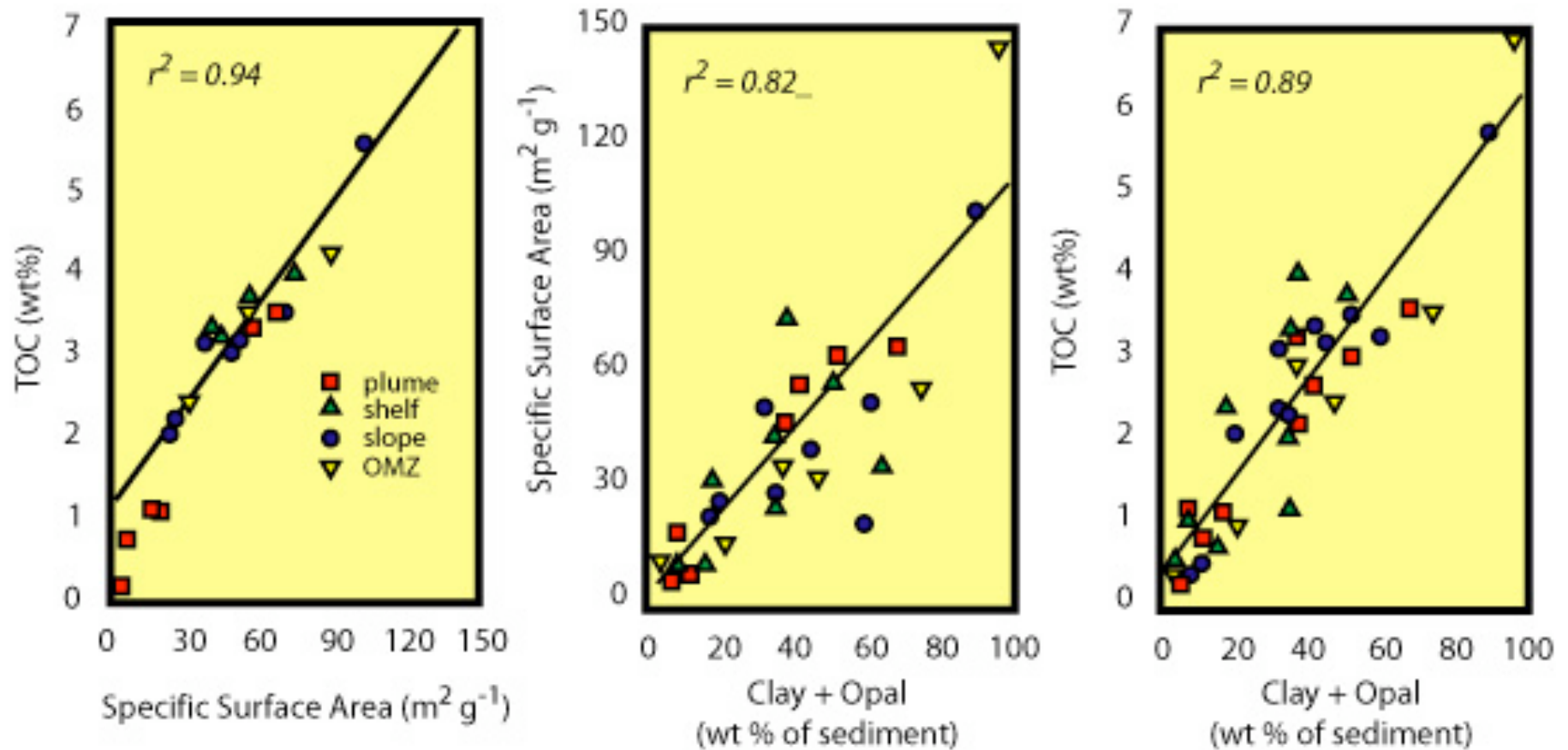




# Grain size, smectite, opal, and surface area in Washington margin sediments

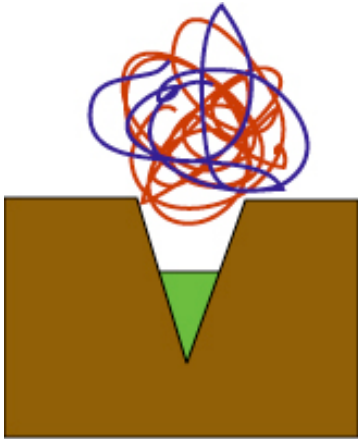


## Correlation of surface area, TOC, Clay minerals+opal in Washington margin sediments



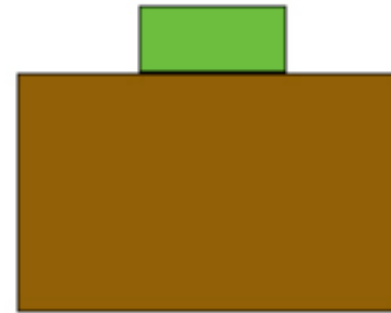
.....and finally the mechanism of preservation....

Mayer-Hedges-Keil hypothesis



Physical protection from enzymatic degradation in small pores/cracks

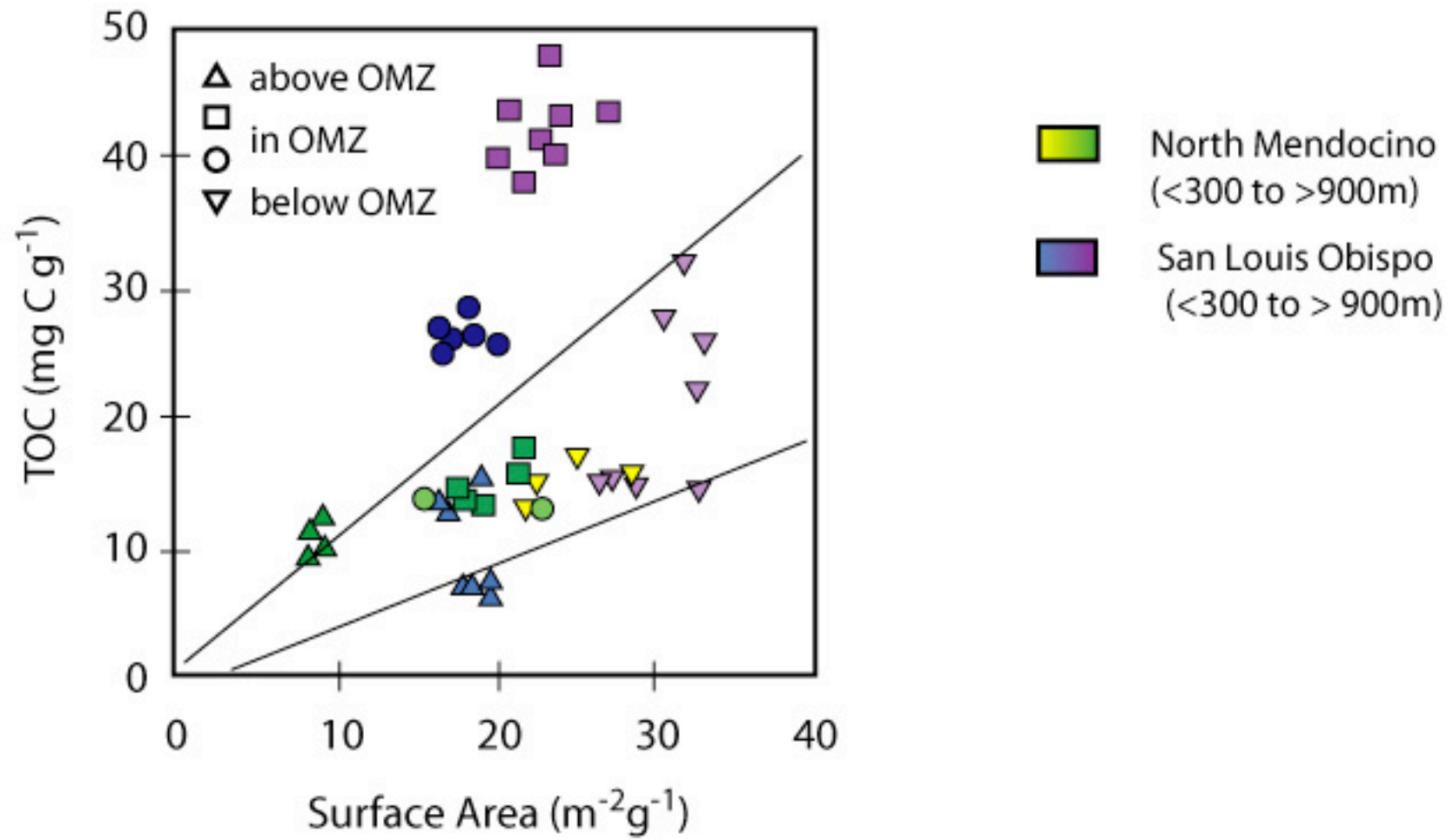
Ransom hypothesis



No physical protection  
OM is on surface and only a small fraction is in contact with mineral.

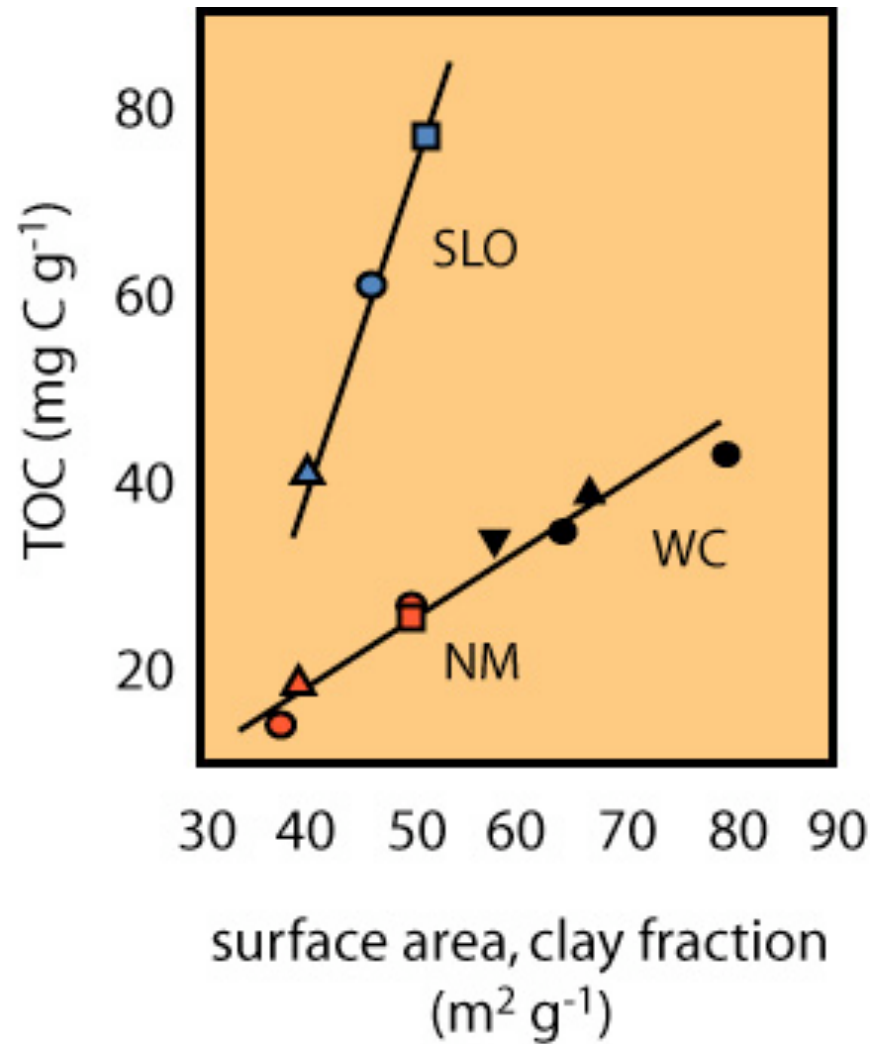
# TOC vs surface area for California margin sediments

Ransom et al., GCA (1998) 62, 1329-1345



# Correlation of clay minerals with TOC in coastal sediments

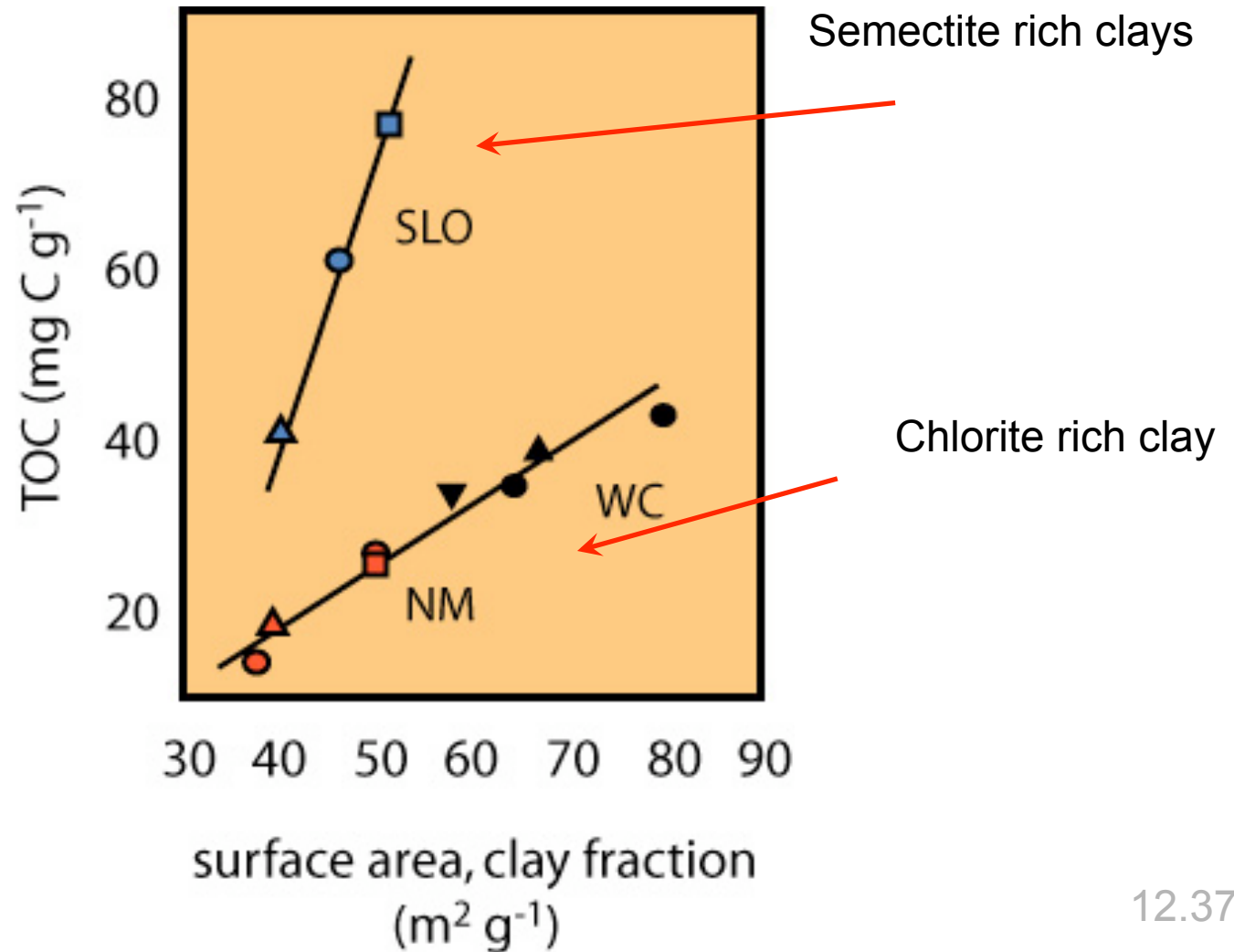
Ransom et al., GCA (1998) 62, 1329-1345



## Correlation of clay minerals with TOC in coastal sediments

**SLO clays**  
21-29% smectite  
0-3% chlorite

**NM clays**  
3-13% smectite  
13-24% chlorite



## Things to remember.....

Most OM is preserved in continental margin sediments

Carbon loading is proportional to surface area

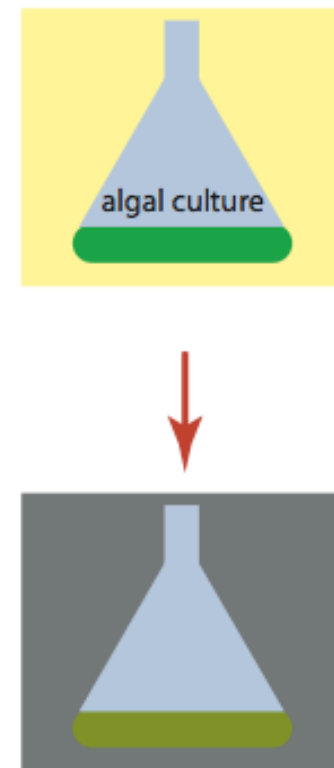
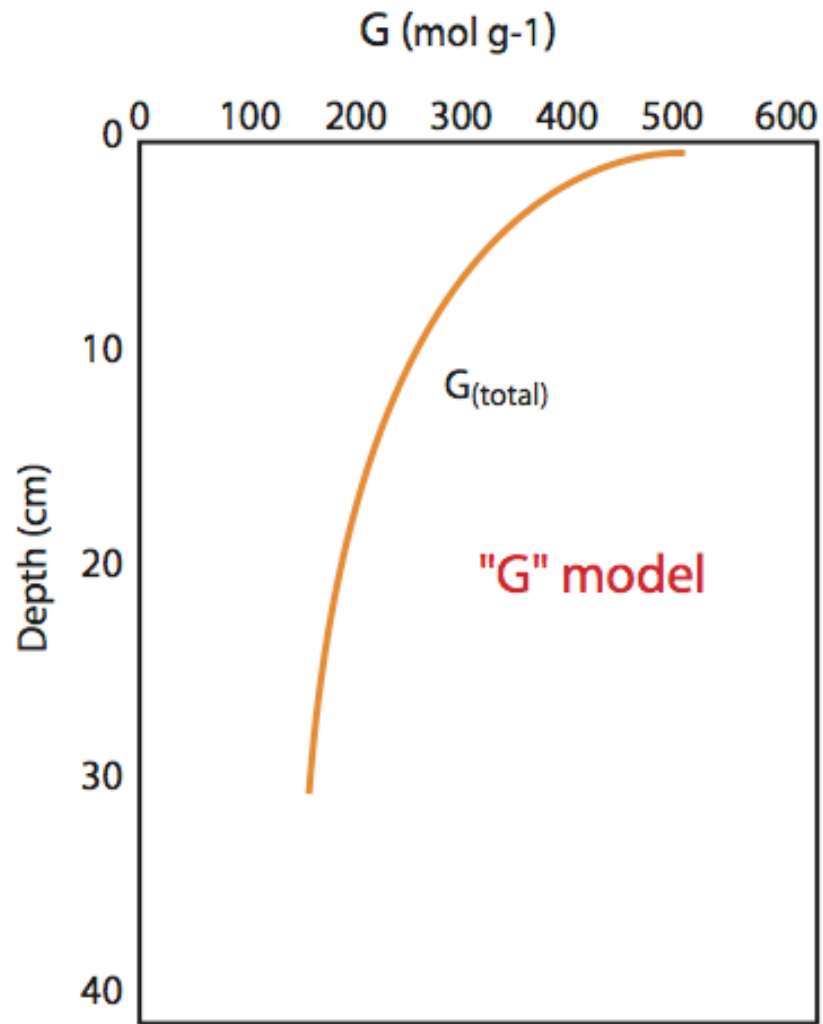
Sedimentation rate, or rate of burial may be a factor

Oxygen may be a factor

Minerology looks to be very important

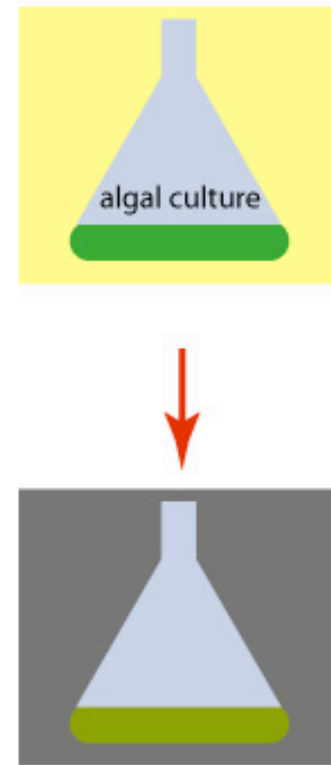
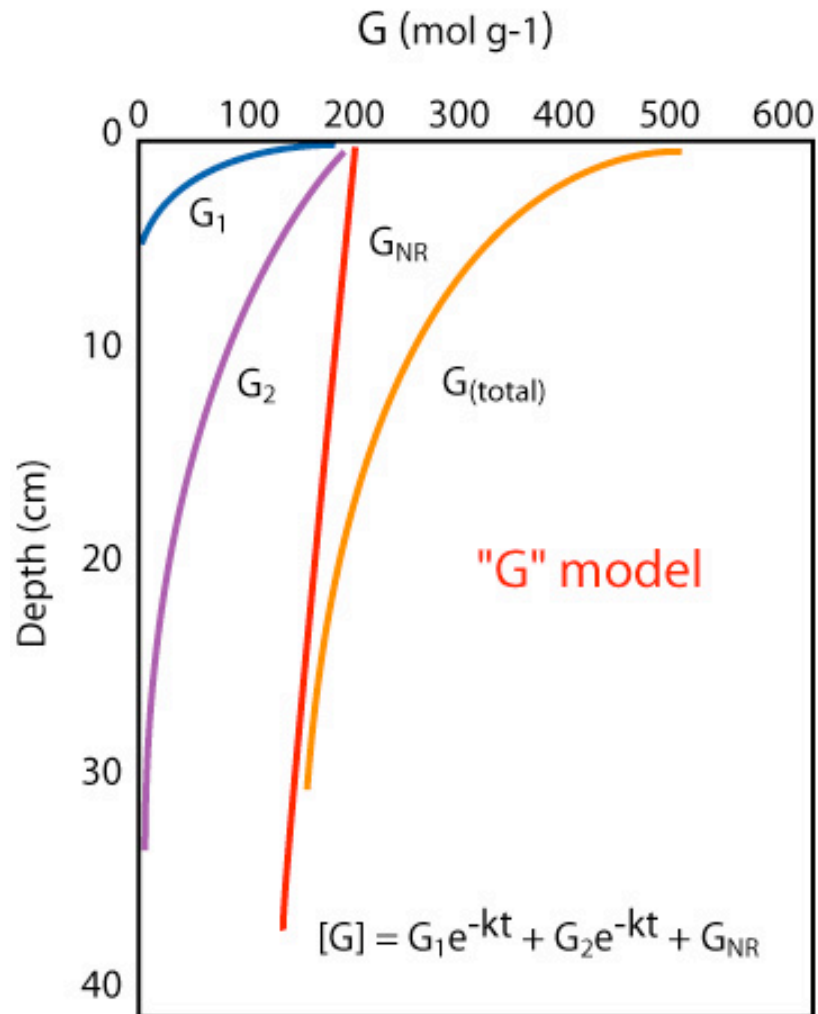
The “mechanisms” of carbon preservation are still not understood. Many relationships between %C, sedimentation rate, SA, oxygen, have been shown, but we do not have a mechanistic explanation for why these relationships are observed.

## Kinetics of organic matter degradation and the multi "G" model



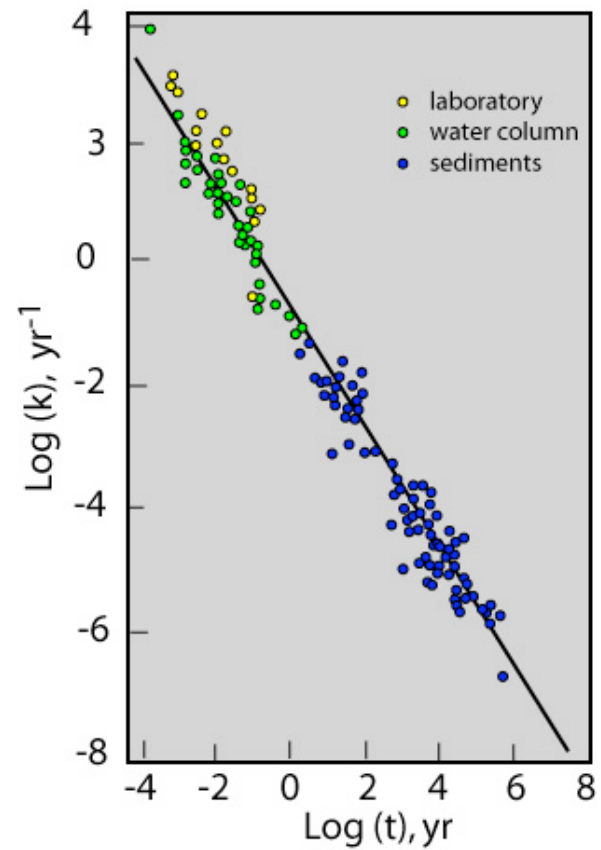
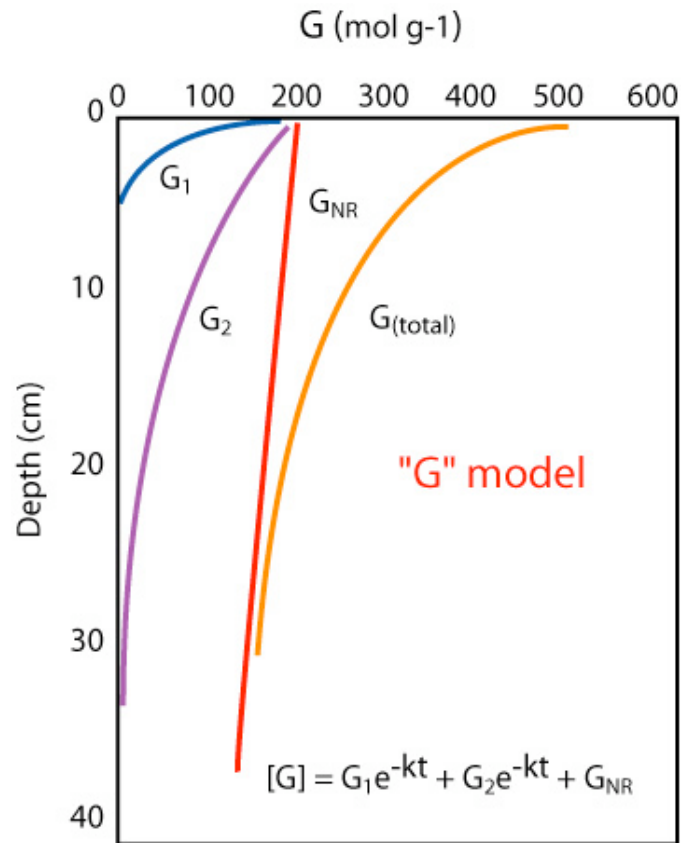


## Kinetics of organic matter degradation and the multi "G" model



Westrich and Berner, 1984  
Berner, 1982

# Is the "G" model just an observational artifact?





## Tabulation of C burial in marine sediments

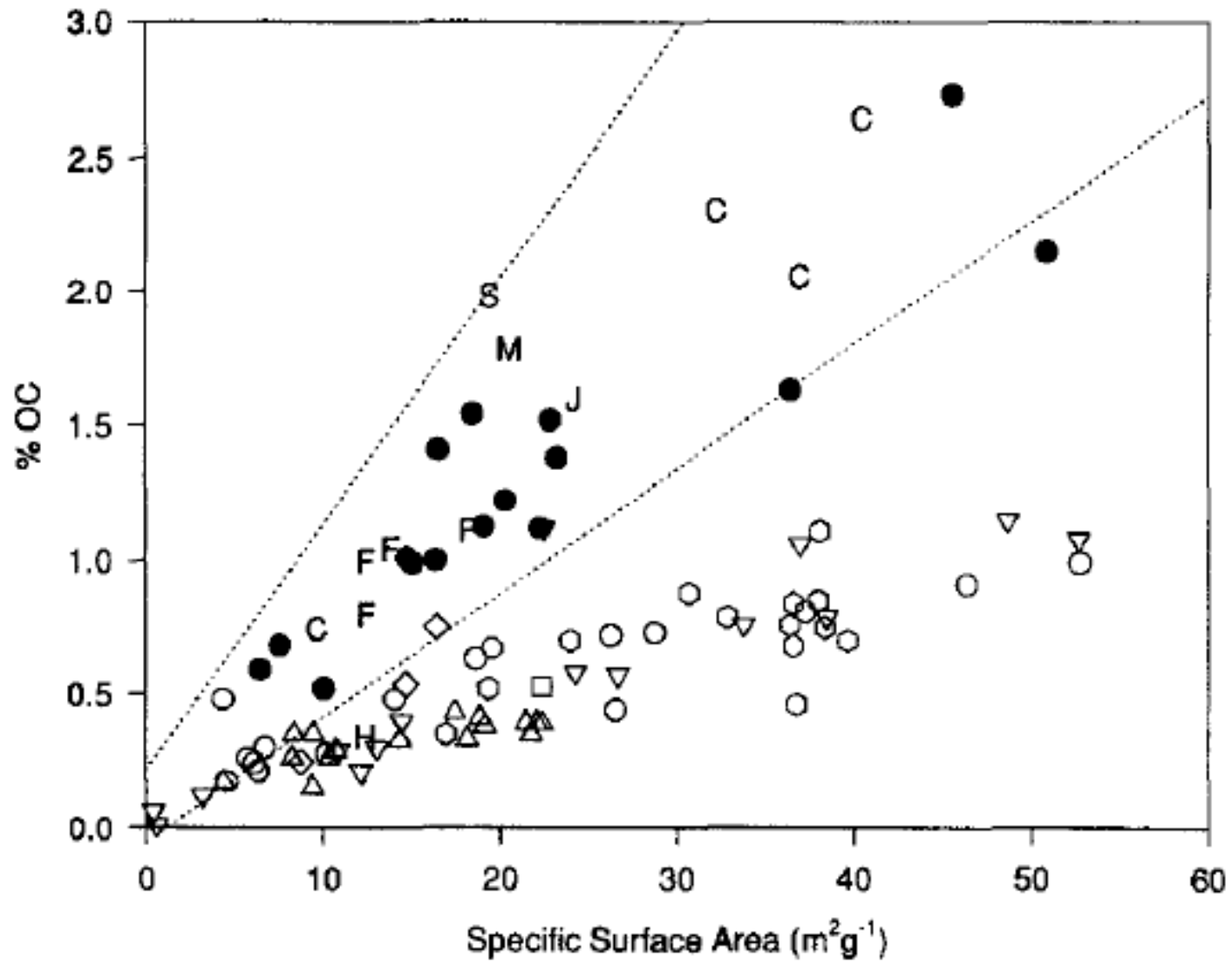
Table 2  
Organic carbon burial rates (and percentages) in different ocean regimes

Sediment type	Deltaic	Shelf	Slope	Pelagic	Total
<b>Data from Gershanovich et al. (1974)</b>					
All sediment types	0 (0)	23 (10)	195 (88)	5 (2)	223 $\Sigma = 223$
<b>Data from Berner (1989)</b>					
Terrigenous deltaic-shelf sediments	104 (82)	0	0	0	104
Biogenous sediments (high-productivity zones)	0	0	7 (6)	3 (2)	10
Shallow-water carbonates	0	6 (5)	0	0	6
Pelagic sediments (low-productivity zones)	0	0	0	5 (4)	5
Anoxic basins (e.g. Black Sea)	0	1 (1)	0	0	1 $\Sigma = 126$
<b>Recalculation of data from Berner (1989) <sup>a</sup></b>					
Deltaic sediments	70 (44)	0	0	0	70
Shelves and upper slopes	0	68 (42)	0	0	68
Biogenous sediments (high-productivity zones)	0	0	7 (4)	3 (2)	10
Shallow-water carbonates	0	6 (4)	0	0	6
Pelagic sediments (low-productivity zones)	0	0	0	5 (3)	5
Anoxic basins (e.g. Black Sea)	0	1 (0.5)	0	0	1 $\Sigma = 160$

Units are  $10^{12}$  g C yr<sup>-1</sup> (parenthetical units = % of total burial).

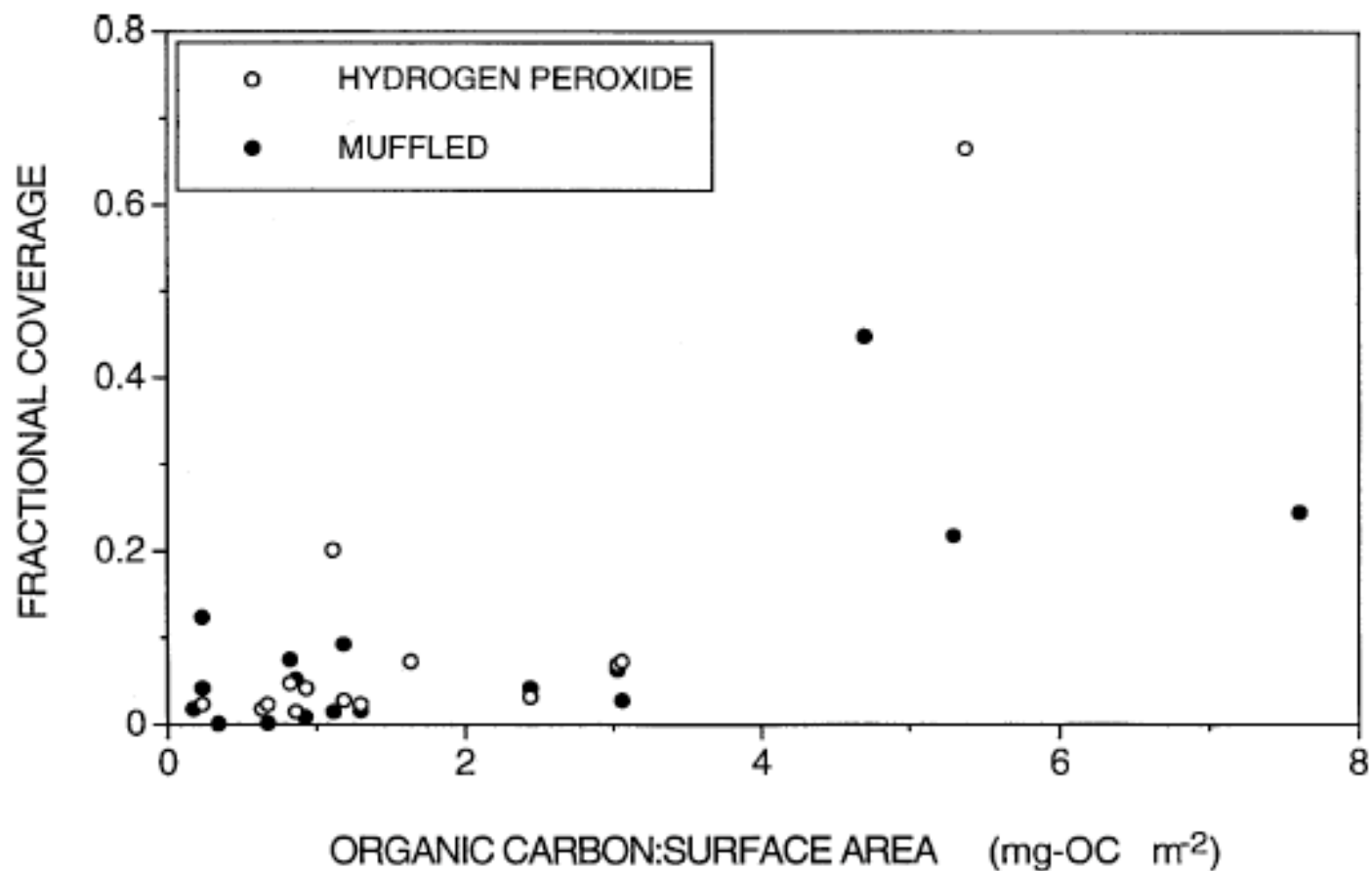
<sup>a</sup> Deltaic-shelf sediments were reapportioned assuming that 33% of the sediment discharge from rivers is deposited either along non-deltaic shelves or upper slopes, and assuming that those deposits have total loadings of 1.5% organic carbon rather than 0.7% as in deltaic regions. Estimates for all other regions remain the same.

# Mineral surface area and % OC in suspended particulate organic matter and deltaic sediments of the Amazon River



- Suspended particulate matter
- Deltaic particulate matter

Photographic and experimental evidence shows that organic matter coating onto particles was not even close to an even coverage, that OM is isolated at very specific sites in blobs or blebs



Larry Mayer

Geochimica et Cosmochimica Acta, Vol. 63, No. 2, pp. 207-215, 1999

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